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- (54) 2- Substituted saccharin derivative proteolytic enzyme inhibitors.
- (F) Novel 2 substituted saccharins which inhibit the enzymatic activity of proteolytic enzymes are useful in the treatment of degenerative diseases and have the formula

wherein:

L is -O-, -S-, -SO- or $-SO_2-$; m and n are each independently 0 or 1; R₁ is substituted phenyl, heterocyclyl or substitued heterocyclyl or, when L is -O- and n is 1, R₁ is cycloheptatrienon -2-yl or, when L is -S- and n is 1, R₁ is cyano or lower – alkoxythiocarbonyl or, when L is $-SO_2-$ and n is 1, R₁ is lower – alkyl or trifluoromethyl; R₂ is hydrogen, lower – alkoxycarbonyl, phenyl or phenylthio; and R₃ and R₄ are each hydrogen or various substituents, and processes for preparation and pharmaceutical compositions and method of use thereof are disclosed.

The invention relates to novel 2 – substituted saccharin derivatives, which inhibit the enzymatic activity of proteolytic enzymes, to pharmaceutical compositions containing the same, to the method of use thereof in the treatment of degenerative diseases and to processes for their preparation.

The inhibition of proteolytic enzymes by nontoxic reagents is useful in the treatment of degenerative disorders, such as emphysema, rheumatoid arthritis and pancreatitis, in which proteolysis is a substantive element. Protease inhibitors are widely utilized in biomedical research. Serine proteases are the most widely distributed class of proteolytic enzymes. Some serine proteases are characterized as chymotrypsin—like or elastase—like based upon their substrate specificity. Chymotrypsin and chymotrypsin—like enzymes normally cleave peptide bonds in proteins at a site at which the amino acid residue on the carbonyl side is typically Trp, Tyr, Phe, Met, Leu or another amino acid residue which contains aromatic or large alkyl side chains. Elastase and elastase—like enzymes normally cleave peptide bonds at a site at which the amino acid residue on the carbonyl side of the bond is typically Ala, Val, Ser, Leu or other similar, smaller amino acids. Both chymotrypsin—like and elastase—like enzymes are found in leukocytes, mast cells and pancreatic juice in higher organisms, and are secreted by many types of bacteria, yeast and parasites.

Several classes of compounds are known to be serine protease inhibitors. For example Powers U.S. Patent 4,659,855 discloses arylsulfonyl fluoride derivatives useful as elastase inhibitors. Doherty et al. in U.S. Patents 4,547,371 and 4,623,645 disclose cephalosporin sulfones and sulfoxides, respectively, which are stated to be potent elastase inhibitors useful in the treatment of inflammatory conditions, especially arthritis and emphysema.

Teshima et al., J. Biol. Chem.,257 (9), 5085-5091 (1982) report the results of studies on serine proteases (human leukocyte elastase, porcine pancreatic elastase, cathepsin G and bovine chymotrypsin A_{α}) with 4-nitrophenylesters and thioesters of N-trifluoroacetylanthranilates, 2-substituted-4H-3,1-benzoxazin-4-ones, 2-substituted-4-quinazolinones and 2-substituted-4-chloroquinazolines.

Cha, Biochem. Pharmacol., 24, 2177 – 2185 (1975) discusses kinetic approaches to the study of the binding of inhibitors to macromolecules, such as enzymes, and methods for determination of such parameters as the inhibition constants, reaction rates and bound and unbound enzyme concentrations.

Certain 2-substituted saccharin derivatives are known to have proteasetype enzyme inhibitory activity. For example Mulvey in U.S. Patent 4,195,023 discloses $R_1-2-R_2CO-1,2$ -benzisothiazol-3-ones, where R_1 is halogen, alkoxy, alkylamino, dialkylamino, alkoxycarbonyl, amino, nitro or hydrogen in the benzenoid ring thereof and R_2 is hydrogen, alkyl, alkenyl, alkynyl, cycloalkyl, halophenyl, heteroaryl or substituted heteroaryl, and $R_1-2-A-CO$ -saccharins, where R_1 has the same meanings as the benzenoid ring substituents in the 1,2-benzisothiazol-3-ones and A is alkyl, alkenyl, alkynyl, cycloalkyl, fluorophenyl, heteroaryl or substituted-heteroaryl. The compounds are said to have elastase inhibitory activity and to be useful in the treatment of emphysema.

Jones et al., U.S. Patent 4,276,298 discloses 2 - R - 1,2 - benzisothiazolinone - 1,1 - dioxides, where R is phenyl substituted by fluoro, dinitro, trifluoromethyl, cyano, alkoxycarbonyl, alkylcarbonyl, carboxyl, carbamoyl, alkylacylamino, alkylsulfonyl, N,N - dialkylsulfamoyl, trifluoromethoxy, trifluoromethylthio, trifluoromethylsulfonyl and trifluoro - methylsulfinyl, or R is pyridyl substituted as above for phenyl with the addition of substitution by mono - nitro. The compounds are said to have protease enzyme inhibitory activity, especially elastase inhibitory activity, and to be useful in the treatment of emphysema, rheumatoid arthritis and other inflammatory diseases.

Powers, Biochem., 24, 2048 – 2058 (1985) discloses studies of the inhibitions of four chymotrypsin – like enzymes, cathepsin G, \overline{rat} mast cell proteases I and II, human skin chymase and chymotrypsin A_a, by N – furoylsaccharin and N – (2,4 – dicyanophenyl)saccharin.

Svoboda et al., Coll. Czech. Chem. Commun., 51, 1133 – 1139 (1986) disclose the preparation of 4 – hydroxy – 2H – 1,2 – benzothiazine – 3 – carboxylates by intramolecular Dieckmann condensation of 2H – 1,2 – benzisothiazol – 3 – one – 2 – acetate – 1,1 – dioxide esters.

Chen, U.S. Patent 4,263,393, Reczek et al. U.S. Patents 4,350,752 and 4,363,865 and Vanmeter et al. U.S. Patent 4,410,618 relate to photographic reagents (Reczek 4,350,752 and Vanmeter et al.), photographic dyes (Reczek 4,363,865) and photographic elements and film units (Chen). They disclose various 2 – substituted – saccharins useful for such applications, for example 2 – aroylmethylsaccharins by Chen, photographic reagents bound through a heteroatom to an imidomethyl blocking group (Reczek 4,350,752), carrier – diffusible photographic dyes bound to the nitrogen atom of an imide through a 1,1 – alkylene group (Reczek 4,363,865) and N – acylmethylimides described as blocked photographic reagents, which have a residue of an organic photographic reagent containing a hetero atom through which it is bound to the blocking group (Vanmeter). Reczek 4,350,752 specifically discloses as Compound 28 the species 2 – (1 – phenyl – 1H – tetrazol – 5 – yl – thiomethyl)saccharin and Vanmeter specifically discloses a number of 2 –

(1 - R' - 1H - tetrazol - 5 - ylthio - methyl)saccharins substituted on the methylene function by an aroyl or t - butylcarbonyl group.

Freed U.S. Patent 3,314,960 discloses 2 – (1,1,3 – trioxo – 1,2 – benzisothiazol – 2 – yl)glutarimides which are stated to be useful as sedatives.

Japanese Patent Publication 72/00419 discloses a number of 2 – RZ – methylsaccharins, stated to have strong activity against rice blast, rice sheath blight, rice helminthosporium leaf spot and rice bacterial leaf blight disease, wherein RZ is lower – alkoxy, butoxyethoxy, ethylthioethoxy, di – lower – alkylaminoethoxy, ethylthio, 2 – chloroethoxy, 1 – (2 – propenyloxy), 1 – (2 – propynyloxy), 2 – saccharinylmethoxy, phenoxy (or phenoxy substituted by chlorine, methyl, nitro or methylthio), phenylthio, chlorophenylthio, benzylthio (or chlorobenzylthio), acetoxy, dichloroacetoxy, benzoyloxy (or benzoyloxy substituted by chlorine or nitro), acetylthio, dichloroacetyloxy, chlorobenzoylthio, methyl or ethylcarbamyloxy, dimethylcarbamyloxy, phenylcarbamyloxy, ethylcarbamylthio, phenylcarbamylthio, dimethylthioylcarbamothioyl, ethylthiothioylthio, ethoxycarbonylthio, ethoxythioylthio and ethylthiocarbonylthio.

2 – Chloromethylsaccharin is disclosed in French Patent 1,451,417 as an intermediate for the prepara – tion of N – methylsaccharin d,1 – transchrysanthemate, useful as an insecticide, and Lo U.S. Patent 3,002,884 discloses 2 – chloro, 2 – bromo and 2 – iodomethylsaccharins, useful as fungicidal agents.

Dunlap et al. PCT Application WO 90/13549 describes saccharin derivatives useful as protcolytic enzyme inhibitors having the structural formula:

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wherein:

L is
$$-O-$$
, $-S-$, $-SO-$ or $-SO_2-$; m and n are each independently 0 or 1;

R₁ is halogen, lower - alkanoyi, 1 - oxo - phenalenyl, phenyl (or phenyl substituted by halogen, lower alkyl, lower - alkoxy, nitro, amino, lower - alkylamino or di - lower - alkylamino) or heterocyclyl selected from 1H - (5 - tetrazolyl), 5 - oxo - 1 - tetrazolyl, 5 - thioxo - 1 - tetrazolyl (when R_2 as defined hereinbelow is other than phenylthio), pyrimidinyl, 2 - benzoxazolyl, 2 - benzothiazolyl, 2 - phthalimidyl, 2 - (1,3,4 thiadiazolyl), 5 - (1,2,4 - thiadiazolyl), 5 - thioxo - 3 - (1,2,4 - thiadiazolyl), 4 - (5 - oxo - 1,3,4 - thiadiazolyl)4-5 - thioxo - 1,3,4 - thiadiazolyl), 3-(1,2,4-triazolyl), 4-(1,2,4-triazolyl), 4-(1,2,3-triazolyl), 2-tm-1idazolyl or 3-(1,2,4-trizolo [4,3-a]-pyridinyl, or such heterocyclyl groups substituted on any available nitrogen atom by lower - alkyl, hydroxy - lower - alkyl, cycloalkyl, 2 - , 3 - or 4 - pyridinyl, carboxy - lower alkyl, lower - alkoxycarbonyl - lower - alkyl, aminocarbonyl - lower - alkyl, lower - alkylaminocarbonyl lower - alkyl, di - lower - alkylaminocarbonyl - lower - alkyl, amino - lower - alkyl, lower - alkylamino - lower alkyl, di - lower - alkylamino - lower - alkyl, 4 - morpholinyl - lower - alkyl, 1 - piperidinyl - lower - alkyl, 1 pyrrolidinyl - lower - alkyl or phenyl (or phenyl substituted by amino, lower - alkylamino, di - lower - al kylamino, lower - alkanamido, N - lower - alkyl - lower - alkanamido, carboxy - lower - alkanamido, carboxy, carbo - lower - alkoxy, lower - alkoxy or halogen), or such heterocyclyl groups substituted on any available carbon atom by nitro, lower-alkyl, amino, lower-alkylamino, di-lower-alkylamino, cycloalkylamino, mercapto, lower - alkylthio, amino - lower - alkyl - thio, lower - alkylamino - lower - alkylthio, di - lower alkylamino - lower - alkylthio, 4 - morpholinyl - lower - alkylthio, 1 - piperidinyl - lower - alkylthio, 1 pyrrolidinyl - lower - alkylthio, carbo - lower - alkoxy or phenyl (or phenyl substituted by amino, lower alkylamino, di - lower - alkylamino, lower - alkanamido, N - lower - alkyl - lower - alkanamido, lower - alkyl lower - alkoxy or halogen);

R₂ is hydrogen, carbo - lower - alkoxy, phenyl or phenylthio;

R₃ is hydrogen, halogen, primary or secondary lower-alkyl, lower-alkoxy, carbo-lower-alkoxy, phenyl, fluoro-lower-alkyl, lower-alkenyl or cyano;

R₄ is hydrogen or from one to two substituents selected from halogen, cyano, nitro, amino, lower – alkanamido, phenyl – lower – alkanamido, diphenyl – lower – alkanamido, lower – alkyl – sulfonylamino, polyfluoro – lower – alkylsulfonylamino, aminosulfonyl, lower – alkyl, polyhalo – lower – alkyl, cycloalkyl,

polyhalo – lower – alkoxy, hydroxy, lower – alkoxy, carboxy, hydroxymethyl, formyl, aminomethyl, lower – alkylsulfonyl, polyhalo – lower – alkylsulfonyl, lower – alkylsulfonylaminosulfonyl and lower – alkoxypoly – lower – alkyleneoxy; and wherein the $-CHR_2$ – group is always appended either to a hetero atom of the L moiety as defined above or it is appended to a hetero atom of the R_1 moiety, with the provisos that

- (i) when m and n are 0 and R_2 , R_3 and R_4 are all hydrogen, R_1 cannot be halogen;
- (ii) when m is 0, n is 1, L is -S- and R_2 , R_3 and R_4 are each hydrogen, R_1 cannot be 1- phenyl -1H- (5-tetrazolyl);
- iii) when m is 0, n is 1, L is -O- or -S- and R_2 , R_3 and R_4 are all hydrogen, R_1 cannot be lower-alkanoyl;
- (iv) when m is 0, n is 1, L is -O-, -S- or -SO-, and R_2 , R_3 and R_4 are all hydrogen, or when m is 0, n is 1 L is -S-, R_2 and R_4 are hydrogen and R_3 is halogen, or when m is 0, n is 1, L is -SO- or $-SO_2-$, R_2 is carbo-lower-alkoxy and R_3 and R_4 are both hydrogen, R_1 cannot be phenyl or substituted phenyl.

It has now been found that the following novel 2-substituted compounds of Formula I also have protease enzyme inhibitory activity and are useful in the treatment of degenerative diseases.

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In the compounds of Formula I,

L is $-O_{-}$, $-S_{-}$, $-SO_{-}$ or $-SO_{2}_{-}$;

m and n are each independently 0 or 1;

 R_1 is phenyl substituted by 1-(4-lower-alkylpiperazin-1-yl)carbonyl, 4-morpholinylsulfonyl, for myl, lower-alkoxycarbonyl, 4-thiamorpholinylsulfonyl or the S-dioxide thereof, hydroxy-lower-alkyl, halo-lower-alkyl, 4-morpholinyl-lower-alkylaminocarbonyl, 4-morpholinyl-lower-alkoxycarbonyl, 1-(4-lower-alkyl-piperazin-1-yl)sulfonyl, 4-morpholinyl-lower-alkoxy, di-lower-alkylamino-lower-alkylaminosulfonyl or an N-lower-alkyl derivative thereof, lower-alkylsulfonyl, 4,5-dihydrooxazol-2-yl, lower-alkyltetrazol-5-yl, 4-morpholinylcarbonyl, nitrophenylazo, carboxyl or di-lower-alkylphosphonyl,

or any of the above in combination with substitution of phenyl by halogen, lower-alkyl, lower-alkoxy, nitro, amino, lower-alkylamino or di-lower-alkylamino as defined in published PCT application WO/13549,

heterocyclyl selected from pyridazin -3-yl, 4-pyron-3-yl, quinolin -8-yl, 1,3,4-oxadiazol-2-yl, coumarin -7-yl, saccharin -6-yl, imidazol -1-yl, 1,3,4-triazol-2-yl, thiazol -2-yl, 2-thioxo-2,3-dihydro-1,3,4-oxadiazol-3-yl, 1,2,5-thiadiazol-3-yl, 2-thioxo-2,3-dihydro-1,3,4-thiadiazol-3-yl, 2-thioxo-2,3-dihydro-1,3,4-thiadiazol-5-yl, 1,2,3-triazol-2-yl, 1,2,4-triazin-5-yl, 5-oxo-6-hydroxy-4,5-dihydro-1,2,4-triazin-3-yl, isoxazol-5-yl, isoxazol-3-yl, 4,5-dihydro-5-oxo-1,2,4-oxadiazol-4-yl, pyridyl, 1,1,3-trioxo-tetrahydro-1,2,5-thiadiazol-2-yl, 6,7-dihydro-1H-1,2,4-triazolo[3,4-b][1,3]thiazin-3-yl, 4,5-dihydro-5-oxo-1,2,4-oxadiazol-4-yl, 2,5-dioxopyrrolidin-1-yl, 3-indolyl, oxazol-2-yl, thiazol-4-yl, 2,3-dihydro-2-oxo-5-phenyl-1,3,4-thiadiazol-3-yl, 2,3-dihydro-2-oxo-5-phenyl-1,3,4-oxadiazol-3-yl, 6-oxo-1,2-dihydro-1,2,4-triazin-1-yl, 1,2,3-triazin-1-yl and 1-indolyl,

or said heterocyclyl substituted on any available nitrogen atom by lower – alkyl, hydroxy – lower – alkyl, cycloalkyl, 2 – ,3 – or 4 – pyridinyl, lower – alkoxycarbonyl, lower – alkoxycarbonyl – lower – alkyl, aminocarbonyl – lower – alkyl, lower – alkylaminocarbonyl – lower – alkyl amino – lower – alkyl, lower – alkylamino – lower – alkyl di – lower – alkylamino – lower – alkyl, 4 – morpholinyl – lower – alkyl, 1 – piperidinyl – lower – alkyl, 1 – pyrrolidinyl – lower – alkyl or phenyl substituted by amino, lower – alkylamino, di – lower – alkylamino, lower – alkanamido, N – lower – alkyl – lower – alkanamido carboxy – lower – alkanamido, carboxy, lower – alkoxycarbony], lower – alkoxy or halo)

or said heterocyclyl substituted on any available carbon atom by nitro, lower - alkyl, amino, lower - alkylamino, di - lower - alkylamino, cycloalkyl - amino, mercapto, lower - alkylthio, amino - lower - alkylthio,

lower – alkylamino – lower – alkylthio, di – lower – alkylamino – lower – alkylthio, 4 – morpholinyl – lower – alkylthio, 1 – piperidiny – lower – alkylthio, 1 – piperidiny – lower – alkylthio, lower – alkylthio, lower – alkylylamino – lower – alkyl 4 – morpholinyl – lower – alkyl – amino, cyano, 1 – piperidinyl – lower – alkyl hydroxy – lower – alkyl, phenylsulfonyl, toluenesulfonyl, halo, tri – lower – alkylsilyl, carboxy or alkali metal salt thereof, furyl, trifluoromethyl, 2 – benzothiazolyl, lower – alkylsulfonyl, aminocarbonyl, benzyl, 4 – mor – pholinyl, pyridinyl, lower – alkoxy, pyrazinyl, lower – alkoxycarbonyl – lower – alkyl, di – lower – alkylaminosulfonyl, 1 – piperidinylcarbonyl, benzyloxy, hydroxy, 4 – morpholinyl – lower – alkyl benzoyl or be – nzoyl substituted by lower – alkoxy or halo, or phenyl (or phenyl substituted by amino, lower – alkylamino, di – lower – alkylamino, lower – alkoxy – poly – lower – alkoxy, methylenedioxy or lower alkoxycarbonyl),

or heterocyclyl selected from 1H – (5 – tetrazolyl), 5 – oxo – 1 – tetrazolinyl, 5 – thioxo – 1 – tetrazolinyl (when R₂ as defined hereinbelow is other than phenylthio), pyrimidinyl, 2 – benzoxazolyl, 2 – benzothiazolyl, 2 – phthalimidyl, 2 – (1,3,4 – thiadiazolyl), 5 – (1,2,4 – thiadiazolyl), 5 – thioxo – 3 – (1,2,4 – thiadiazolinyl), 4 – (5 – oxo – 1,3,4 – thiadiazolinyl), 4 – (5 – thioxo – 1,3,4 – thiadiazolinyl), 3 – (1,2,4 – triazolyl), 4 – (1,2,4 – triazolyl), 1,2,3 – triazol – 1 – yl, 2 – imidazolyl and 3 – (1,2,4 – triazolo[4,3 – a]pyridinyl) when substituted on any available carbon atom by di – lower – alkylamino – lower – alkyl, 4 – morpholinyl – lower – alkylamino, cyano, 1 – piperidinyl – lower – alkyl, hydroxy – lower – alkyl, phenylsulfonyl, toluenesulfonyl, halo, tri – lower – alkylsilyl, carboxy or alkali metal salt thereof, furyl, trifluoromethyl, 2 – benzothiazolyl, lower – alkylsulfonyl, aminocarbonyl, benzyl, 4 – morpholinyl, pyridinyl, lower – alkoxy, pyrazinyl, lower – alkoxycarbonyl – lower – alkyl benzoyl or benzoyl substituted by lower – alkoxy or halo, or phenyl substituted by trifluoromethyl, lower – alkoxy – poly – lower – alkoxy, methylenedioxy or lower alkoxycarbonyl,

or, when R₄ is carboxy – lower – alkoxy lower – alkoxycarbonyl – lower – alkoxy or di – lower – alkylamin – carbonyloxy, heterocyclyl is 1 – phenyltetrazol – 5 – yl,

or, when L is -O- and n is 1, R¹ is cycloheptatrienon -2-yl or, when L is -S- and n is 1, R¹ is cyano or lower – alkoxythiocarbonyl or, when L is $-SO_2$ – and n is 1, R¹ is lower – alkyl or trifluoromethyl;

R₂ is hydrogen, lower - alkoxycarbonyl phenyl or phenylthio;

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R₃ is hydrogen, halo, primary or secondary lower – alkyl, lower – alkoxy, lower – alkoxycarbonyl, phenyl, fluoro – lower – alkyl, lower – alkenyl, cyano or di – lower – alkylamino; and

R₄ is hydrogen or from one to three substituents selected from halo, cyano, nitro, amino, lower – alkanamido, phenyl – lower – alkanamido diphenyl – lower – alkanamido, lower – alkylsulfonylamino, polyfluoro – lower – alkylsulfonylamino, aminosulfonyl, lower – alkyl, polyhalo – lower – alkylsulfonyl, polyhalo – lower – alkoxy, hydroxy, lower – alkoxy, carboxy, hydroxymethyl, formyl, aminomethyl, lower – alkylsulfonyl, polyhalo – lower – alkylsulfonyl, lower – alkylsulfonyl, lower – alkoxy – lower – alkoxy or di – lower – alkylaminocarbonyloxy; provided that

(1) when n is 0, R₁ can only be heterocyclyl and CHR₂ can only be bonded to a ring nitrogen atom of R₁ and

(2) when m is 0, n is 1, L is -O-, -S- or -SO-, and R_2 , R_3 and R_4 are all hydrogen or, when m is 0, n is 1, L is -S-, R_2 and R_4 are hydrogen and R_3 is halo or, when m is 0, n is 1, L is -SO- or $-SO_2-$, R_2 is lower-alkoxycarbonyl and R_3 and R_4 are both hydrogen, R_1 cannot be substituted phenyl.

According to the present invention therefore there are provided novel 2 – substituted saccharin deriva – tives of Formula I which have protease enzyme inhibitory activity and which are useful in the treatment of degenerative diseases. In another aspect the invention relates to compositions for the treatment of degenerative diseases which comprise a pharmaceutical carrier and an effective proteolytic enzyme inhibiting amount of a novel 2 – substituted saccharin derivative of Formula I.

In a further aspect there is provided a method of use of the said 2-substituted saccharins of Formula I for preparation of a medicament for treatment of a degenerative disease in a patient in need of such treatment, the medicament containing an effective proteolytic enzyme inhibiting amount of a said 2-substituted saccharin.

Preferred compounds of Formula I described above are compounds of formula I wherein m is 0, R_2 is hydrogen, R_3 is hydrogen, halo, primary or secondary lower-alkyl or lower-alkoxy, R_4 is hydrogen, hydroxy, lower-alkoxy, lower-alkoxy-lower-alkoxy-poly-lower-alkyleneoxy, carboxy-lower-alkoxy or lower-alkoxycarbonyl-lower-alkoxy and

n is 1, L is -O- and R₁ is phenyl substituted by 1-(4-lower-alkyl-piperazin-1-yl)carbonyl, 4-morpholinylsulfonyl, 4-thiamorpholinylsulfonyl or the S-dioxide thereof, 4-morpholinyl-lower-al-kylaminocarbonyl, 4-morpholinyl-lower-alkoxycarbonyl, 1-(4-lower-alkylpiperazin-1-yl)sulfonyl,

di – lower – alkylamino – lower – alkylaminosulfonyl and/or 4 – morpholinylcarbonyl or 1,2,5 – thiadiazol – 3 – yl,or 1,2,5 – thiadiazol – 3 – yl substituted by 4 – morpholinyl, or isoxazolyl substituted by lower – alkoxycar – bonyl: or

n is 1, L is -S- and R_1 is 1,3,4-oxadiazol-2-yl substituted on any available carbon atom by furyl, benzyl, pyridinyl, pyrazinyl or phenyl; or

n is 0 and R_1 is 1,2,3 - trizol - 1 - yl or 1,2,3 - triazol - 2 - yl substituted on any available carbon atom by cyano or lower - alkylsulfonyl.

Particularly preferred compounds are those wherein L is -0- or -S-, m is 0, n is 0 or 1 and R_2 is hydrogen, especially wherein R_3 is hydrogen, chloro, bromo, methyl, ethyl, propyl, isopropyl, sec. – butyl, methoxy, ethoxy, isopropoxy or phenyl and wherein R_4 is hydrogen, 7-chloro, 5-nitro, 6-nitro, 5-amino, 5-acetylamino, 5-(3,3-diphenyl-propionamido), 5-(1,1,3,3-tetramethylbutyl), 6-hydroxy, 7-hydroxy, 5-methoxy, 6-methoxy, 7-methoxy, 7-dimethoxy, 7-dimethox, 7-dim

It should be understood that the compounds having the general structural Formula I are usually named in the chemical literature as 1,2-benzisothiazol-(2H)-3-one-1,1-dioxides. However for the sake of brevity, such compounds are frequently named as saceharin derivatives, and that nomenclature is used herein in describing the compounds of the invention and their biological properties.

As used herein the terms lower-alkyl, lower-alkoxy and lower-alkane mean monovalent aliphatic radicals, including branched chain radicals, of from one to ten carbon atoms. Thus the lower-alkyl (or lower-alkane) moiety of such groups include, for example, methyl, ethyl, propyl, iso-propyl, n-butyl, sec-butyl, t-butyl, n-pentyl, 2-methyl-3-butyl 1-methylbutyl, 2-methylbutyl, neopentyl, n-hexyl, 1-methylpentyl, 3-methylpentyl, 1-ethylbutyl, 2-ethylbutyl, 2-hexyl, 3-hexyl, 1,1,3,3-tetramethyl-pentyl, 1,1-dimethyloctyl and the like. Lower-alkanoyl has from two to ten carbon atoms and is branched or unbranched.

As used herein the term halogen (or halo) means fluorine, chlorine, bromine or iodine.

As used herein the term cycloalkyl means carbocyclic rings having from three to six ring carbon atoms, including cyclopropyl, cyclobutyl, cyclopentyl and cyclohexyl which may be substituted on any ring carbon atom thereof by one or more lower – alkyl groups.

As used herein the term lower-alkenyl means monovalent, unsaturated radicals, including branched chain radicals, of from three to ten carbon atoms and thus includes 1 - (2 - propenyl), 1 - (2 - butenyl), 1 - (1 - methyl - 2 - propenyl), 1 - (4 - methyl - 2 - pentenyl), 1 - (4 - methyl

The compounds of the present invention inhibit the activity of serine proteases, specifically human leukocyte elastase and the chymotrypsin-like enzymes, and are thus useful in the treatment of degen-erative disease conditions such as emphysema, rheumatoid arthritis and pancreatitis. In the process of binding to, and inhibiting, the activity of a proteolytic enzyme, it is believed that the compounds of the invention are cleaved at the bond between the methylene (CHR₂) and the L_nR_1 functions, and that the L_nR_1 group is split off as an anion which can thus be described as a leaving group. This cleavage is believed to be facilitated by the presence of an electron withdrawing group, such as cyano, halogen, nitro, carboxy, lower-alkoxycarbonyl, acyl or phenylthio, in the R_1 functionality to thereby increase its electronegativity which can be expressed in terms of the pKa values of the acid form of the leaving group which ideally should be less than about 7. A particularly preferred group of such compounds are those of formula I where R_3 is other than hydrogen.

In another aspect the invention relates to a process for the preparation of said 2-saccharin derivatives which comprises reacting a 2-halomethylsaccharin either with an alkali metal salt of a L_nR_1 moiety or with a L_nR_1 moiety in the presence of an acid-acceptor.

In a further aspect the invention relates to a process for the preparation of said 2 – saccharin derivatives which comprises reacting an alkali metal or thallous salt of a 2 – unsubstituted saccharin either with a halo – CHR_2 – L_nR_1 moiety to obtain the desired product or with a 3 – chlor – 3 – (phenylthio)propyl – L_nR_1 species followed by oxidation of the product with a peracid to obtain a 2 – [1 – (phenylsulfinyl)propy – L_nR_1] saccharin and heating the latter to obtain a 2 – [1 – (2 – propenyl) – L_nR_1] saccharin.

In yet another aspect of the invention there is provided a process for the preparation of 4-primary-or secondary – lower – alkyl – R_4 – 2- unsubstituted saccharins, useful as intermediates for the preparation of the corresponding 2- saccharin derivatives, which comprises reacting a 2- primary – lower – alkyl – N,N- di – lower – alkyl bennamide with a lower – alkyl lithium in an inert organic solvent; reacting the resulting lithium salt with a lower – alkyl halide; reacting the resulting 2- primary or secondary – lower – alkyl – R_4- N,N- di – lower – alkyl benzamide with a lower – alkyl lithium; reacting the resulting lithium salt with sulfur

dioxide followed by hydroxylaminesulfonic acid in the presence of base; and heating the product in an acid medium.

Thus the compounds of Formula I where m is 0, R_2 is hydrogen and L is -0- or -S- are prepared by reaction of a 2-halomethylsaccharin derivative of formula I where R_1 is halogen, R_2 is hydrogen, m and n are 0 and R_3 and R_4 have the meanings given above, with an appropriate L_nR_1 moiety. The reaction can either be carried out in the presence of an acid-acceptor, such as an alkali metal carbonate, a tri-lower-alkylamine, an alkali metal or thallous-lower-alkoxide or an alkali metal hydride, or alternatively an alkali metal salt of the L_nR_1 moiety can be used. The reaction is carried out in an organic solvent inert under the conditions of the reaction, e.g., acetone, methylethylketone (MEK), tetrahydrofuran (THF), diethylether, dimethylformamide (DMF), methylene dichloride (MDC) or lower-alkanols, at a temperature in the range from ambient up to the boiling point of the solvent used. The corresponding compounds where L is -SO- or $-SO_2-$ are prepared by oxidation of the corresponding compounds of Formula I where L is -S- with one or two molar equivalents, as appropriate, of a peracid such as 3-chloroperbenzoic acid.

Alternatively, the compounds of Formula I where m is 0 can be prepared by reaction of an alkali metal or thallous saccharin salt (prepared by reaction of an appropriate $4 - R_3 - R_4 - 2$ – unsubstituted saccharin with an alkali metal alkoxide or a thallous lower – alkoxide) with a halo – $CHR_2 - L_nR_1$ moiety, where R_1 , R_2 , R_3 , R_4 , L and n have the meanings given above with respect to Formula I. A thallium salt can be used when R_2 has all the meanings given above, but an alkali metal salt can be used only when R_2 is hydrogen. The reaction is carried out in an inert organic solvent, for example a lower – alkanol or DMF, at temperatures in the range from 20°C to the boiling point of the solvent used.

The compounds of Formula I where m is 1 and R_2 is hydrogen are prepared by reaction of a 3-(phenylthio)propyl- L_nR_1 compound, prepared by reaction of a L_nR_1 -propyl halide with sodium thiophenoxide in methylethylketone (MEK), with N-chlorosuccinimide to give a 3-chloro-3-(phenylthio)propyl- L_nR_1 species. Reaction of the latter with a thallium salt of an appropriate $4-R_3-R_4$ -saccharin, using the same conditions described above for the preparation of the compounds of Formula I from a saccharin salt and a halo- $CHR_2-L_nR_1$ moiety, affords a $2-[1-(phenylthio)propyl-L_nR_1]$ saccharin. Oxdation of the latter to the corresponding $2[1-(phenylsulfinyl)propyl-L_nR_1]$ saccharin followed by heating the product in an alkylene glycol ether, for example ethylene glycol dimethyl ether, affords the compounds of Formula I where m is 1 and R_2 is hydrogen.

The compounds of Formula I, where R_1 is lower-alkanoyl, R_2 is hydrogen, L is -0-, m is 0, n is 1, and R_3 and R_4 have the meanings given above, are prepared by treating the corresponding 2-hydroxymethylsaccharin with an appropriate acid anhydride in the presence of a catalytic amount of a mineral acid or a strong organic acid, for example sulfuric acid or p-toluenesulfonic acid.

A compound of Formula I wherein n is 0 and R_1 is 1,2,3-triazol-1-yI is prepared by condensation of the corresponding compound of formula I wherein R_1 is halo with an alkali metal azide and then cycloaddition of the resulting azide with the corresponding substituted or unsubstituted acetylene. The preferred alkali metal azide is sodium azide. The condensation is carried out with or without heating or cooling, preferably at room temperature, in an inert solvent, for example benzene, toluene or dimethylfor-mamide, optionally using a crown ether, for example 18-crown-6 ether. The cycloaddition is preferably carried out in the same inert solvent with heating.

The 2-halomethylsaccharins of Formula I where R_1 is halogen, R_2 is hydrogen, m and n are 0, and R_3 and R_4 have the meanings given above with respect to formula I, and the corresponding $4-R_3-R_4-2-1$ unsubstituted saccharins required for the preparation of the compounds of Formula I where R_1 , L, m and n have the other meanings given above, are prepared by the methods described by D'Alelio et al., J. Macromol. Sci-Chem., A3(5), 941 (1969) and Saari et al., J. Het. Chem., 23, 1253 (1986). In the method described by Saari, a methyl ester of an appropriate anthranilic acid is prepared by conventional means from the substituted anthranilic acid and the ester dizotized. The dizonium salt is then reacted with sulfur dioxide and cupric chloride to produce a sulfonyl chloride which is then reacted with concentrated ammonium hydroxide to produce the substituted saccharin derivatives of formula II. The latter, on reaction with formaldehyde in a lower-alkanol solvent affords the 2-hydroxymethylsaccharins of formula III, which, on reaction with a thionyl halide or a phosphotus trihalide, afford the corresponding 2-halomethylsaccharin derivatives. The approach is illustrated as follows:

5
$$R_4$$
 COOAIK R_4 NH₂

10 NaNO₂/HCI

15 R_4 COOAIK SO_2 CUCI₂ R_4 COOAIK R_4 CI

26 NH₄OH

27 NH₄OH

28 NH₄OH

39 NH₄CO

30 NH₄CO

31 NH₄CO

32 NH₄CO

33 NH₄CO

44 SOX₂ or PX₃

45 NH₄CH₂COOAIK

46 NH₄CO

47 NH₄CO

48 NH₄CO

49 NH₄CO

40 NH₄CO

40 NH₄CO

41 NH₄CO

42 NH₄CO

43 NH₄CO

44 NH₄CO

45 NH₄CO

46 NH₄CO

47 NH₄CO

48 NH₄CO

49 NH₄CO

40 NH₄CO

40 NH₄CO

41 NH₄CO

42 NH₄CO

43 NH₄CO

44 NH₄CO

45 NH₄CO

46 NH₄CO

47 NH₄CO

48 NH₄CO

49 NH₄CO

40 NH₄CO

40 NH₄CO

40 NH₄CO

41 NH₄CO

42 NH₄CO

43 NH₄CO

44 NH₄CO

45 NH₄CO

46 NH₄CO

47 NH₄CO

48 NH₄CO

49 NH₄CO

40 NH₄CO

40 NH₄CO

40 NH₄CO

41 NH₄CO

42 NH₄CO

43 NH₄CO

44 NH₄CO

45 NH₄CO

46 NH₄CO

47 NH₄CO

48 NH₄CO

49 NH₄CO

40 NH₄CO

40 NH₄CO

40 NH₄CO

41 NH₄CO

42 NH₄CO

43 NH₄CO

44 NH₄CO

45 NH₄CO

46 NH₄CO

47 NH₄CO

48 NH₄CO

48 NH₄CO

49 NH₄CO

40 NH₄CO

40 NH₄CO

40 NH₄CO

40 NH₄CO

40 NH₄CO

40 NH₄CO

41 NH₄CO

42 NH₄CO

43 NH₄CO

44 NH₄CO

45 NH₄CO

46 NH₄CO

47 NH₄CO

48 NH₄CO

48 NH₄CO

49 NH₄CO

49 NH₄CO

49 NH₄CO

40 NH₄CO

where R_3 and R_4 have the meanings given above and X is halogen.

The halomethylsaccharins of Formula I, where R_1 is halogen, R_2 is hydrogen, m and n are 0, and R_3 and R_4 have the meanings given above with respect to Formula I, can also he prepared by reaction of a corresponding 2-phenylthiomethylsaccharin with a sulfuryl halide in an inert organic solvent, for example MDC, ethylene dichloride (EDC) or carbon tetrachloride, at a temperature from around 0°C to around 30°C.

The compounds of Formula II where R_3 is either primary or secondary lower-alkyl, and which are useful as intermediates for the preparation of the compounds of Formula I as described above, are prepared by one of two methods. The compounds of Formula II where R_3 is primary lower-alkyl are prepared by reacting a 4-methyl- R_4 -2-unsubstituted-saccharin with two molar equivalents of a lower-alkyl lithium in an inert organic solvent, for example THF, and reacting the resulting lithium salt with one molar equivalent of a lower-alkyl halide, both reactions being carried out at a temperature in the range from about -50° C to -80° C.

The compounds of Formula II where R₃ is either primary or secondary lower – alkyl are prepared by reaction of a 2 - primary - lower - alkyl - R₄ - N,N - di - lower - alkylbenzamide with one molar equivalent of a lower - alkyl lithium in the presence of a tetra - lower - alkylethylenediamine in an inert organic solvent, for example THF and reaction of the resulting lithium salt with one molar equivalent of a lower - alkyl halide at a temperature in the range from about -50°C to -80°C. The resulting 2-primary - or secondary -loweralkyl - R₄ - N,N - di - lower - alkyl - benzamide is then reacted with one molar equivalent of a lower - alkyl lithium in the presence of a tetra - lower - alkylethylenediamine in an inert organic solvent, for example THF, and the resulting lithium salt reacted with sulfur dioxide at a temperature in the range from -50°C to -80°C followed by reaction of the product with hydroxylaminesulfonic acid in the presence of base. The resulting 2 - lower - alkyl - R₄ - 6 - amino - sulfonyl - N,N - di - lower - alklbenzamide is thereafter heated in an acid medium to effect cyclization of the latter to the desired 4 - primary or secondary lower - alkyl - R₄ -2 - unsubstituted - saccharin of Formula II. It is preferred to carry out the cyclization in refluxing glacial acetic acid. When the 2-lower-alkyl group in the 2-lower-alkyl-R₄-N,N-di-lower-alkyl-benzamide starting material is methyl, alkylation affords species where the 2-lower-alkyl group is either straight or branched depending upon whether a straight or branched chain lower - alkyl halide is used for the alkylation. On the other hand, when the 2 - lower - alkyl group in the starting material contains more than one carbon atom, alkylation takes place on the carbon atom adjacent the benzene ring and affords products having a sec. - lower - alkyl group at the 2 - position.

Access to certain of the required intermediates of Formula II in some cases requires building up the two rings making up the saccharin nucleus. Thus to prepare compounds where R_3 is lower – alkoxy and R_4 is 7–hydroxy, 3,3 – dithio – bispropionic acid is converted to the bis acid chloride by reaction of the acid with thionyl chloride, and the acid chloride is then reacted with two molar equivalents of benzylamine to produce the bis n – benzylamide. The latter, on reaction with sulfuryl chloride in an organic solvent, such as MDC, EDC or carbon tetrachloride, affords 5 – chloro – 2 – benzyl – 2H – isothiazol – 3 – one, which is oxidized with one molar equivalent of a per acid, such as perhenzoic acid or 3 – chloroperbenzoic acid, to 5 – chloro – 2 – benzyl – 2H – isothiazol – 3 – one – 1 – oxide. The latter, on heating under pressure with a 2 – lower – alkox – yfuran in an organic solvent, such as benzene, toluene or xylene, affords a 4 – lower – alkoxy – 7 – hydroxy – 2 – benzyl 1,2 – benzisothiazol – 2H – 3 – one – 1 – oxide. The 7 – hydroxy group can, if desired, then be reacted with a lower – alkyl halide or a lower – alkoxypoly,lower,alkoxy – lower – alkyl halide to give the corresponding 4,7 – di – lower – alkoxy or 4 – lower – alkoxy – 7 – lower – alkoxypoly – lower – alkyleneoxy – 2 – benzyl – 1,2 – benzisothiazol – 2H – 3 – one – 1 – oxide. Further oxidation of the product with one molar equivalent of a per acid as described before followed by catalytic debenzylation affords the corresponding 4 – lower – alkoxy – 7 – R_4 – 2 – unsubstituted saccharins.

Other simple chemical transformations which are conventional and well known to those skilled in the art of chemistry can be used for effecting changes in functional groups in the compounds of the invention. For example, catalytic reduction of nitro groups to produce the corresponding amino substituted compounds, acylation of amino—substituted species to prepare the corresponding amides or oxidation of sulfides or sulfoxides to prepare the corresponding, respective sulfoxides or sulfones as desired can be carried out. In standard biological test procedures, the compounds of Formula I have been found to possess human leukocyte elastase (HLE) and chymotrypsin inhibitory activities, and are thus useful in the treatment of degenerative diseases, such as emphysema, rheumatoid arthritis or pancreatitis.

The compounds of Formula I having basic functions can be converted to the acid – addition salt form by interaction of the base with an acid. In like manner, the free base can be regenerated from the acid – addition salt form in conventional manner, that is by treating the salts with cold, weak aqueous bases, for example alkali metal carbonates and alkali metal bicarbonates. The bases thus regenerated can be interacted with the same or a different acid to give back the same or a different acid – addition salt. Thus the bases and all of their acid – addition salts are readily interconvertible.

Likewise certain compounds of Formula I having acid, i.e. carboxylic acid, functions can be converted to salt forms thereof by reaction of the acid with a base, such as alkali metal or ammonium hydroxides or with organic bases such as alkyl, dialkyl or trialkylamines, and the acids can be regenerated from the salts by treatment of the salts with aqueous acids.

It will thus be appreciated that Formula I not only represents the structural configuration of the bases and acids of Formula I but is also representative of the structural entities which are common to all of the compounds of Formula I whether in the form of the free base, the free acids or in the form of the salts of the bases and acids. It has been found that, by virtue of these common structural entities the compounds of Formula I and their salts have inherent pharmacological activity of a type to be more fully described hereinbelow. This inherent pharmacological activity can be enjoyed in useful form for pharmaceutical purposes by employing the free bases or free acids themselves or the salts formed from pharmaceutically acceptable acids and bases, that is acids or bases whose anions or cations are innocuous to the animal organism in effective doses of the salts so that beneficial properties inherent in the common structural entity represented by the free bases and free acids are not vitiated by side effects ascribable to the anions or cations.

In utilizing this pharmacological activity of the salts, it is preferred, of course, to use pharmaceutically acceptable salts. Although water insolubility, high toxicity or lack of crystalline character may make some particular salt species unsuitable or less desirable for use as such in a given pharmaceutical application, the water – insoluble or toxic salts can be converted to the corresponding pharmaceutically acceptable bases by decomposition of the salts with aqueous base or aqueous acid as explained above, or alternatively they can be converted to any desired pharmaceutically acceptable salt by double decomposition reactions involving the anion or cation, for example by ion – exchange procedures.

Moreover, apart from their usefulness in pharmaceutical applications, the salts are useful as char-acterizing or identifying derivatives of the free bases or free acids or in isolation or purification procedures. Like all of the salts, such characterizing or purification salt derivatives can, if desired, be used to regenerate the pharmaceutically acceptable free bases or free acids by reaction of the salts with aqueous base or aqueous acid, or alternatively they can be converted to a pharmaceutically acceptable salt by, for example, ion – exchange procedures.

The novel feature of the compounds then resides in the concept of the bases and the cationic and anionic forms of the $4-R_3-R_4-2-$ substituted saccharins of Formula I and not in any particular acid or base moiety or acid anion or base cation associated with the salt forms of the compounds; rather, the acid or base moieties or the anions or cations which can be associated with the salt forms are in themselves neither novel nor critical and therefore can be any acid anion or base cation capable of salt formation with the bases or acids.

The compounds of Formula I of the invention can be prepared for pharmaceutical use by incorporating them in unit dosage form as tablets or capsules for oral administration either alone or in combination with suitable adjuvants such as calcium carbonate, starch, lactose, talc, magnesium stearate, gum acacia and the like. Still further, the compounds can be formulated for oral, parenteral or aerosol inhalation administration either in aqueous solutions of water soluble salts of the compounds or in aqueous alcohol, glycol or oil solutions or oil – water emulsions in the same manner as conventional medicinal substances are prepared.

The percentages of active component in such compositions may be varied so that a suitable dosage is obtained. The dosage administered to a particular patient is variable, depending upon the clinician's judgment using as criteria: the route of administration, the duration of treatment, the size and physical condition of the patient, the potency of the active component and the patient's response thereto. An effective dosage amount of the active component can thus only be determined by the clinician after a consideration of all criteria and using his best judgment on the patient's behalf.

The molecular structures of the compounds of the invention were assigned on the basis of study of their infrared and NMR spectra. The structures were confirmed by the correspondence between calculated and found values for elementary analyses for the elements.

The following examples will further illustrate the invention without, however, limiting it thereto. All melting points are uncorrected.

Example 1

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Powdered potassium hydroxide (7.4 g; 132 mmol; 2 equiv.) was admixed with dimethyl sulfoxide (DMSO) (100 ml), and the mixture was stirred for 5 minutes. 6 – Methylanthranilic acid (10.0 g; 66 mmol) was then added to the mixture and iodomethane (4.52 ml; 73 mmol; 1.1 equiv.) added dropwise. The reaction mixture was stirred for 30 minutes at room temperature, then diluted with ether (250 ml), washed with water (3 x 100 ml), dried (MgSO₄) and concentrated. The crude product was filtered through a pad of flash grade (32 – 63) silica gel and eluted with 1:9 ether:hexane to afford 4.23 g (39%) of methyl 6 – methylanthranilate as an oil. ¹H nmr (300 MHz, CDCl₂): 7.078 (1H, t, J=7.67 Hz); 6.529 (2H, d, J=7.79 Hz); 5.111 (2H, br s); 3.887 (3H, s); 2.424 (3H, s). IR (neat film, cm¹): 3480 (m), 3380 (m), 2950 (w), 1690

(s); 1605 (s).

The methyl 6 – methylanthranilate prepared in (a) (4.23 g; 25.6 mmol) was dissolved in acetic acid (25 ml) and the solution cooled to 0° C. Concentrated hydrochloric acid (45 ml) was added to produce a tan slurry. A solution of sodium nitrite (1.89 g; 27.4 mmol; 1.07 equiv.) in water (8 ml) was added dropwise with stirring, the resulting orange solution was stirred at 0° C for 1 hour and then added in 6 portions to a mixture of cupric chloride dihydrate (2.18 g; 12.8 mmol; 0.5 equiv.) and sulfur dioxide (6.3 g; excess) in acetic acid (33 ml) and water (6 ml) at 0° C. The dark green solution was stirred at room temperature overnight, poured into ice – water (300 ml), and the solid which separated was collected and dried by suction to provide 1.11 g of the sulfonyl chloride which was immediately added to ice cold arnmonium hydroxide (100 ml) and stirred at room temperature overnight. The solution was acidified to pH 1 with concentrated HCl and the resulting precipitate was collected and air – dried to provide 729 mg (12%) of 4 – methylsaccharin, mp 224 – 226° C. ¹H nmr (300 MHz, CD₃CN): 9.5 (1H, br s); 7.782 (2H, d, J = 4.35 Hz); 7.644 (1H, t, J – 4.20 Hz); 2.683 (3H, s). IR (KBr, cm¹): 3400 (w); 3100 (s); 3000 (s); 1720 (s); 1580 (m). FDMS: m/e 197 (M¹).

4-Methylsaccharin prepared in (b) (500 mg; 2.54 mmol) was dissolved in 2.53 ml of warm ethanol (steam bath). Once a homogeneous solution was achieved, formalin (37% in methanol; 1.76 ml excess) was added dropwise. The solution was allowed to cool to room temperature and then chilled to 0 °C for 4 days. The resulting solid was collected and air-dried to afford 476 mg (82%) of 2-hydroxymethyl-4-methylsaccharin, mp 196-198 °C. ¹H nmr (300 MHz, CDCl₃): 7.767 (1H, t, J=6.75 Hz); 7.732 (1H, d, J=7.72 Hz); 7.600 (1H, d, J=6.64 Hz); 5.361 (2H, d, J=8.00 Hz); 3.296 (1H, t, J=8.16 Hz); 2.793 (3H, s). IR (KBr, cm¹): 3505(s); 3070(w); 1735(s); 1580(m).

2 - Hydroxymethyl - 4 - methylsaccharin produced in (c) (76 mg; 0.33 ml) was admixed with acetic anhydride (1 ml; excess), and 2 drops concentrated sulfuric acid were added. The reaction mixture was stirred for 2 hours at room temperature, at which time a non - polar spot was observed by thin - layer chromatographic (tlc) analysis. The reaction mixture was diluted with MDC (50 ml) and washed with saturated sodium bicarbonate (2 x 15 ml). After drying (Na₂SO₄), the solvent was removed to afford 64 mg (72%) of 2 - acetoxymethyl - 4 - methylsaccharin, mp 198 - 205 °C (decomp.) ¹H nmr (300 MHz, CDCl₃): 7.8 (2H, m); 7.64 (1H, d, J = 6.18 Hz); 5.84 (2H, s); 2.82 (3H, s); 2.15 (3H, s). IR (KBr, cm¹): 2920 (w); 1745 (s); 1735 (s); 1630 (w). FDMS: m/e 269 (M¹).

30 Example 2

Using the procedure described above in Example 1, 6 – chloroanthranilic acid (5.00 g; 29.2 mmol) and iodomethane (2.75 ml; 44 mmol; 1.5 equiv.) were reacted in the presence of powdered potassium hydroxide (4.08 g; 72.7 mmol; 2.5 equiv.) to give 4.22 g (78%) of methyl – 6 – chloroanthranilate as an oil. 1 H nmr (300 MHz, CDCl₃): 7.077 (1H, t, J=8.06 Hz); 6.744 (1H, d, \overline{J} =6.7 Hz); 6.575 (1H, d, J=8.25 Hz); 4.871 (1H, br s); 3.929 (3H, s). IR (neat film, cm¹): 3480 (m); 3380 (m); 2950 (w); 1705 (s); 1610 (s).

4 - Chlorosaccharin was prepared by the same method as used for preparing 4 - methylsaccharin using methyl 6 - chlor - anthranilate (4.22 g; 22.7 mmol) in acetic acid (22 ml) and cone. HCl (40 ml) and sodium nitrite (1.68 g; 24.3 mmol) in water (7 ml) to prepare the diazonium salt which was added to cupric chloride dihydrate (1.93 g; 11.4 mmol; 0.5 equiv.) and sulfur dioxide (6.5 g; excess) in acetic acid (30 ml)/water (5 ml). The resulting sulfonyl chloride was treated with ammonium hydroxide (150 ml) as previously described to afford 3.07 g (62%) of 4 - chlorosaccharin as a pale yellow solid, mp 245 - 246 °C. ¹H nmr (300 MHz, CD₃CN): 7.918 (1H, dd, J=7.39, 1.91 Hz); 7.865 (1H, t, J=7.52 Hz); 7.829 (1H, br d, J=7.30 Hz). IR (KBr, CM¹): 3570 (s); 3520 (s); 2950 (s,b); 1735 (s); 1630 (m). FDMS: m/e 217 (M¹).

2-Hydroxymethyl-4-chlorosaccharin was prepared in the same maker as 2-hydroxymethyl-4-methylsaccharin, in Example 1, from 4-chlorosaccharin (1.00 g; 4.60 mmol) and formalin (37%; 3.22 ml; excess). All attempts to crystallize the viscous oily product resulted in decomposition to the starting material, and the product was thus used in the next step without characterization.

2-Acetoxymethyl-4-chlorosaccharin was prepared in the same manner as used for 2-acetoxymethyl-4-methyl-saccharin, in Example 1, from the crude 2-hydroxymethyl-4-chlorosacharin (0.34 g; 1.4 mmol) and acetic anhydride (2.5 ml) with 2 drops of sulfuric acid. In this case, after isolation, the product was purified by filtration through a pad of silica gel and elution with 1:1 ether:hexane to afford 2-acetoxymethyl-4-chlorosaccharin (35 mg, 95% yield) as a white solid, mp 138-142°C. ¹H nmr (300 MHz, CDCl₂): 7.921 (1H, dd, J=6.54, 2.63 Hz); 7.874 (1H, t, J=7.98 Hz); 7.842 (1H, dd, J=6.70, 2.20 Hz); 5.869 (2H, s); 2.172 (3H, s). IR (KBr, CM¹): 1745 (s); 1735 (m, shoulder); 1575 (w). Comb. anal.:

Theor.	C, 41,46;	H, 2.78;	N, 4.83;
Found	C, 41.17;	H, 2.81;	N, 4.75.

Example 3

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20

Crude 2 – hydroxymethyl – 4 – chlorosaccharin, from Example 2, (609 mg; 2.46 mmol max) was admixed with diethyl ether (5 ml), and thionyl chloride (3 ml; excess) was added. The resulting mixture was heated to effect complete solution, stirred at room temperature overnight, diluted with ether (20 ml) and filtered through a pad of celite topped with sand and eluted with ether. Removal of the solvent afforded 430 mg of crude chloromethyl derivative. A portion (225 mg) was removed for further reactions. The remainder (205 mg) was flash chromatographed on silica gel and eluted with 40% ether/ pentane to provide 137 mg of 2 – chloromethyl – 4 – chlorosaccharin, mp 135 – 136 °C. ¹H nmr (300 MHz, CDCl₃): 7.925 (1H, dd, J = 6.62, 2.26 Hz); 7.882 (1H, t, J = 8.18 Hz); 7.846 (1H, dd, J = 7.42, 2.36 Hz); 5.561 (2H, s). IR (KBr, CM¹): 3090 (w); 3050 (w); 1750 (s); 1575 (m). FDMS: m/e 265 (M¹).

Example 4

The chloromethyl derivative prepared in Example 3 (225 mg; 0.85 mmol) and sodium 1 phenyl – 5 – mercapto – 1H – tetrazole (200 mg; 1.01 mmol; 1.2 equiv.) were dissolved in acetone (5 ml) to give a tan solution. After about 10 minutes a precipitate was observed, and after stirring overnight at room temperature no 2 – chloromethyl – 4 – chlorosaccharin was present by tlc analysis. The reaction mixture was poured into water and extracted with MDC (3 x 25 ml). The combined extracts were dried (Na_2SO_4), concentrated and the residue flash chromatographed on silica gel and eluted with 1:1 ether:hexane. The major spot was collected to afford 122 mg of 4 – chloro – 2 – (1 – phenyl – 1H,tetrazol – 5 – ylthiomethyl)saccharin as a white solid, mp 175 – 177 °C. ¹H nmr (300 MHz, CDCl₃): 7.813 (3H, m); 7.515 (5H, s); 5.710 (2H, s). IR (Kbr, cm¹): 3080 (w); 1740 (s); 1590(w). FDMS: m/e 407 (M¹); 230 (M¹ – PMT); 178 (1%, PMT).

o Example 5

The chloromethyl derivative prepared as in Example 3, (337 mg crude; maximum 1.27 mmol) was dissolved (to the extent possible) in acetone (10 ml). Sodium 1 – phenyl – 5 – mercapt – 1H – tetrazole (304 mg; 1.52 mmol; 1.2 equiv.) was added, and the reaction mixture was stirred at room temperature for 3 days. The mixture was diluted with MDC (50 ml), washed with water (3 x 25 ml), dried (Na₂SO₄), concentrated and filtered through a pad of silica gel (1:1 ether:hexane elution). The material thus obtained was chromatographed on flash silica gel and eluted with 1:1 ether:hexane to afford 44 mg (8.5%) of 4 – chloro – 2 – (4 – phenyl – 5 – thioxotetrazolin – 1 – ylmethyl)saccharin mp 158 – 162 °C. ¹H nmr (300 MHz, CDCl₃): 7.981 (1H, d, J = 7.12Hz); 7.95 (2H, m); 7.887 (1H, t, J = 6.74 Hz); 7.864 (1H, d, J = 7.32Hz); 7.567 (3H, m); 6.392 (2H, s). IR (KBr, CM¹): 1745 (s); 1185 (s). FDMS: m/e 407 (M¹); 230 (M¹ – PMT).

Example 6

A mixture of 2-(chloromethyl)saccharin (0.98 g, 4.2 mmol), 1-(3-acetamidophenyl)5-mercapto-1H-tetrazole (1 g, 4.2 mmol) and potassium bicarbonate (0.84 g, 8.4 mmol) in methyl ethyl ketone (50 ml) was heated at 50 °C under nitrogen overnight. The reaction mixture was cooled, poured into dilute HCl/ice water (300 ml.), and the water was decanted from the semi-solid which solidified on trituration with hot ethyl acetate. The resultant white solid was recrystallized from acetonitrile (MeCN) with charcoaling, to afford 0.82 g of $2-[1-(3-acetamidophenyl)-1H-tetrazol-5-ylthiomethyl]saccharin as small white needles, mp <math>195-\overline{196}$ °C decomp. $\overline{}^{1}H$ nmr (90 MHz CDCl₃):2.05 (3H, s); 5.65 (2H, s). FDMS: m/e 430 (M⁺).

Theor.	C, 47.43;	H, 3.28;	N, 19.52;
Found	C, 47.02;	Н, 3.27;	N, 19.53.

Example 7

A mixture of 2-(bromomethyl)saccharin (2.7 g, 9.8 mmol), 1-(3-heptanamidophenyl)-5-mercapto-1H-tetrazole (3 g, 9.8 mmol) and potassium carbonate (3.4 g., 24.5 mmol) was heated under reflux in methyl ethyl ketone (50 ml) under nitrogen for 1 hour. The mixture was cooled and poured into a sodium bicarbonate/ice solution. The water layer was decanted from the resultant white semi-solid. The semi-solid was washed with water, then dissolved in hot acetonitrile, the solution treated with activated charcoal and filtered. The filtrate was freed of solvent under vacuum, and the resultant solid was chromatographed (silica gel-95:5 CH_2Cl_2 :acetone) to give a clear oil. The oil was crystallized from hot ethanol to afford 1.6 g of 2-[1-(3-heptanamidophenyl)-1H-tetrazol-5-ylthiomethyl]saccharin as a white solid, mp 146-147.5° C. 1H nmr (90 MHz, CDCl_3): 5.65 (2H, s). FDMS: m/e 500 (M+).

	C, 52.79;		
Found	C, 52.44;	Н, 4.75;	N, 16.64.

Example 8

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A mixture of 2-(bromomethyl)saccharin (3 g, 10.8 mmol) and the sodium salt of 5-mercapto-1-methyl-1H-tetrazole (1.49 g, 10.8 mmol) was heated under reflux in methyl ethyl ketone (75 ml) for 2 hours. The reaction mixture was cooled, poured into dilute sodium bicarbonate/ice solution and extracted with MDC (2x's). The combined organic extracts were dried (Na₂SO₄) and freed of solvent under vacuum. The crude product was chromatographed (silica gel-95:5 CH₂Cl₂: ether), and the resultant oil was crystallized from hot isopropanol to afford 2.7 g (80%) of 2-(1-methyl-1H-tetrol-5-ylthiomethyl)-saccharin as a white solid, mp 106-110°C. ¹H nmr (90 MHz, CDCl₃):5.55 (2H, s). FDMS: m/e 311 (M⁺).

Theor.	C, 38.58;	H, 2.91;	N, 22.49;
Found	C, 38,53;	Н, 2.79;	N, 22.60.

Example 9

A mixture of 2 – (chloromethyl)saccharin (3 g, 12.9 mmol), 1 – cyclohexyl – 5 – mercapto – 1H – tetrazole (2.37 g, 12.9 mmol) and potassium carbonate (4.45 g, 32.2 mmol) was heated under reflux in methyl ethyl ketone (50 ml) for 1 hour. The reaction mixture was cooled, poured into dilute sodium bicarbonate/ice solution and extracted twice with ethyl acetate. The combined organic extracts were washed with water, dried (Na₂SO₄) and freed of solvent under vacuum. Chromatography (silica gel; MDC) afforded 2 g of 2 – (1 – cyclohexyl – 1H – tetrazol – 5 – ylthiomethyl)saccharin as a white foam that was crystallized from hot cyclohexane, mp 103 – 105 °C. ¹H nmr (90 MHz, CDCl₃):5.65 (2H, s). FDMS: m/e 379 (M⁺).

Theor.	C, 47.48;	H, 4.52;	N, 18.46;
Found	C, 47.84;	H, 4.61;	N, 18.36.

Example 10

A mixture of m-chloroperbenzoic acid (0.43 g, 2.67 mmol) and 2-(1-phenyl-1H-tetrazol-5-ylthiomethyl)saccharin (1 g, 2.67 mmol) in MDC was stirred at room temperature for 24 hours. TLC (95:5 CH₂Cl₂:ether) revealed the presence of starting sulfide. Additional peracid (0.2 g) was added and the mixture stirred for an additional 2 days. The reaction mixture was washed with sodium bicarbonate solution, dried (Na₂SO₄) and freed of solvent under vacuum. Chromatography (silica gel-95:5 CH₂Cl₂:ether) afforded a foam that was crystallized from ether to afford 0.52 g of 2-(1-phenyl-1H-tetrazol-5-ylsulfinylmethyl)saccharin as white solid, mp 161-162°C. ¹H nmr (90 MHz, CDCl₃):5.5-6.0 (2H, q). FDMS:m/e 196 (M⁺ - PMT), 389 (M⁺).

Theor.	C, 46.27;	H, 2.85;	N, 17.98;
Found	C, 46.00;	H, 2.83;	N, 17.76.

Example 11

A mixture of 2-bromomethyl-5-nitrosaccharin (2 g, 6.2 mmol) and 1-phenyl-5-mercapto-1H-tetrazole sodium salt in methyl ethyl ketone (40 ml)/DMF (10 ml) was heated under reflux for 2 hours. The reaction mixture was cooled and poured into a dilute sodium bicarbonate/ice solution. The resultant white solid, isolated by filtration, was washed with water and air dried. The compound was sonicated with 50:50 MDC:acetone and filtered to remove soluble impurities. The remaining solid was recrystallized from 2:1 acetonitrile: ethanol to afford 1.5 g of 5-nitro-2-(1-phenyl-1H-tetrazol-5-ylthiomethyl)saccharin as an off-white solid, mp 189-190°C. ¹H nmr (90 MHz, DMSO-d₆):5.75 (2H, s). FDMS: m/e 418 (M⁺).

Theor.	C, 43.06;	H, 2.41;	N, 20.09;
Found	C, 42.29;	Н, 2.43;	N, 20.13.

Example 12

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A mixture of m-chloroperbenzoic acid (2.2 g, 12.8 mmol) and 2-(phenylsulfinylmethyl)saccharin (3.75 g, 11.6 mmol) in MDC (50 ml) was stirred at room temperature for 2 hours. An additional spatula of peracid was added, and stirring was continued for an additional 1 hour. m-Chlorobenzoic acid was removed by filtration and the solid was washed with a small amount of MDC. The filtrate was washed with sodium bicarbonate solution, dried (Na₂SO₄) and freed of solvent under vacuum. The resultant solid was recrystal-lized from 50:50 ethanol:acetonitrile to afford 2-(phenylsulfonylmethylsaccharin as a white solid, mp 169-171°C. ¹H nmr (90 MHz, DMSO-d₆, CDCl₃):5.15 (2H, s). FDMS: m/e 196 (M⁺-PMT).

Theor.	C, 49.84;	H, 3.29;	N, 4.15;
Found	C, 49.92;	H, 3.24;	N, 4.13.

Example 13

A mixture of m-chloroperbenzoic acid (0.9 g, 5.37 mmol) and 2-(2-pyrimidylthiomethyl)saccharin prepared by procedures similar to those of Examples 9 and 11 (1.5 g, 4.8 mmol) in MDC (75 ml) was stirred overnight at room temperature. The reaction mixture was washed with sodium bicarbonate solution, dried (Na₂SO₄) and freed of solvent under vacuum. Part of this crude product (0.5 g) was saved for direct conversion to the sulfone; the remaining material was chromatographed (silica gel – 95:5 CH₂Cl₂:acetone). Recrystallization from ethanol:acetonitrile afforded 0.95 g of 2 – (2 – pyrimidylsulfinylmethyl)saccharin as white crystals, mp 197 – 198 °C. decomp. 1 H nmr (90 MHz, CDCl₃, DMSO – d₆):5.1 – 5.5 (2H, q).

Theor.	C, 44.57;	H, 2.81;	N, 13.00;
Found	C, 44.67;	H, 2.84;	N, 12.97.

Example 14

A mixture of m-chloroperbenzoic acid (0.4 g, 2.3 mmol) and the sulfoxide prepared in Example 13 (0.75 g 2.3 mmol) in MDC (50 ml) was stirred at room temperature with TLC (95:5 MDC:acetone) monitoring. After 2 hours, some of the starting sulfoxide still remained; an additional spatula of peracid was added and the reaction was stirred overnight. Methylene chloride (100 ml) was added and the mixture was washed with sodium bicarbonate solution. The organic layer was dried (Na₂SO₄) and the solvent was removed under vacuum. Recrystallization of the residue from acetonitrile:ethanol afforded 0.95 g of 2-(2-

pyrimidylsulfonylmethyl)saccharin as a white solid, mp 225 - 227 °C decomp. ¹H nmr (90 MHz, DMSO - d₀):5.78 (2H, s). FDMS: m/e 196 (M+ - PMT), 339 (M+).

Theor.	C, 42.47;	H, 2.67;	N, 12.38;
Found	C, 42.20;	H, 2.62;	N 12.46.

Example 15

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A mixture of 2 – (chloromethyl)saccharin (3 g, 12.9 mmol) and sodium p – nitrophenoxide (2.55 g, 12.9 mmol) in THF (50 ml) was heated overnight at 50 °C and then refluxed for 45 minutes. The reaction mixture was cooled, poured into a dilute sodium bicarbonate/ice solution and extracted twice with ethyl acetate. The combined organic extracts were washed with sodium bicarbonate solution and water, dried (Na₂SO₄) and taken to dryness in vacuo. Chromatography (silica gel; MDC) afforded an oil that was crystallized from hot cyclohexane/ether. The resultant solid was recrystallized from ethanol to afford 0.92 g of 2 – (4 – nitrophenoxymethyl)saccharin as white shiny platelets, mp 162 – 164 °C. ¹H nmr (90 MHz, CDCl₃, DMSO – d_6):5.95 (2H, s). FDMS m/e 334 (M⁺).

Theor	C 50.30;	H, 3.02;	N, 8.38;
Found	C, 50.06;	H, 2.91;	N, 8.28.

25 Example 16

5 - Nitro - 2 - (1 - phenyl - 1H - tetrazol - 5 - ylthiomethyl)saccharin (4 g, 9.56 mmol) (Example 11) was dissolved in THF (250 ml) and placed in a Parr shaker bottle. Two spatulas of 10% palladiumon - charcoal catalyst were added under nitrogen, and the mixture was shaken under hydrogen (55 psi) for 2.5 days. The reaction mixture was filtered through Celite diatomaceous earth and filtrate mixed with water and extracted with MDC. The organic layer was dried (Na₂SO₄) freed of solvent under vacuum, and the resultant yellow foam was sonicated with warm ethanol, cooled and filtered. The desired corresponding 5 - aminosaccharin derivative, 0.5 g, was isolated as a cream - colored solid. FDMS: m/e 388 (M⁺).

A mixture of the above 5 – aminosaccharin derivative (0.5 g, 1.29 mmol) and 3,3 – diphenylpropanoyl chloride (0.315 g, 1.29 mmol) in acetonitrile (50 ml) was heated under reflux for 2.5 hours. TLC (95:5 methylene chloride:acetone) analysis revealed the presence of some starting amine. A small additional amount of acid chloride was added, refluxing was continued for an additional 1.5 hours and the reaction mixture cooled and poured into ice water (400 ml). After 30 minutes, the mixture was filtered, and the resultant tan colored solid was washed with water and air dried. Chromatography on silica gel (95:5 MDC:ether) produced a foam that was crystallized from hot ethanol to yield 0.68 g of $\frac{5 - (3,3 - \text{diphenyipropionamido}) - 2 - (1 - \text{phenyl} - 1\text{H} - \text{tetrazol} - 5 - \text{ylthiomethyl})\text{saccharin}$ as a white solid, $\frac{1}{1}$ mp 92 - $\frac{1}{1}$ mp $\frac{1}{1}$ mmr(90 mHz, $\frac{1}{1}$ mmr(9

Theor. for $C_{30}H_{24}N_6O_4S_2 + 2C_2H_5OH$:				
C, 59.28; H, 5.27; N, 12.2; Found: C, 58.09; H, 5.15; N, 12.09				

Example 17

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Methyl 2 - chlor - 2 - phenylthioacetate was prepared as reported in the literature: I. Fleming and J. lqbal, Tetra. Lett., 24, 327 (1983); M. Campbell, et al., Tetra. Lett., 21, 3305 (1980).

Saccharin (10 g, 54.6 mmol) was dissolved in ethanol (500 ml) with slight warming. Thallous ethoxide (13.6 g, 54.6 mmol) was added dropwise, and the resultant heterogeneous mixture was stirred at room temperature for 2 hours, then cooled, filtered and the solid washed with cold ethanol. The greyish white,

crystalline solid was dried under vacuum in a desiccator to yield 19.4 g (92%) of the thallium salt of saccharin.

A mixture of the thallium salt of saccharin (1.78 g, 4.6 mmol) and methyl 2-chloro -2-phenyl-thioacetate (1 g, 4.6 mmol) in DMF (25 ml) was stirred at 60° C for 7 hours. The mixture was cooled and poured into ice water (400 ml). After 30 minutes, the mixture was filtered, and the solid was washed with water and air-dried. Chromatography on silica (MDC) afforded a clear oil that was crystallized from hot ethanol to yield 0.87 g (51%) of white needles of methyl-2-phenylthio-2-(2-saccharinyl)acetate, mp $144-146^{\circ}$ C. 1 H nmr (90 MHz, CDCl₃): 3.8 (3H, s); 5.95 (1H, s); 7.2-8.15 (9H, m). FDMS: m/e 363 (M⁺).

A solution of methyl 2-phenylthio -2-(2-saccharinyl)acetate (2 g, 5.5 mmol) and sulfuryl chloride (0.74 g, 5.5 mmol) in MDC (50 ml) was stirred at room temperature for 2 hours. Solvent was removed under vacuum, and the yellow oil was crystallized from warm ethanol to give 0.94 g of product. NMR revealed greater than 50% starting material. An additional amount of starting material (1 g, 2.75 mmol) was added to the crude product mixture and it was redissolved in methylene chloride. Sulfuryl chloride (0.5 ml) was again added and the mixture was stirred at room temperature for about 12 hours. Work up as above afforded 0.66 g of crude methyl 2-chloro -2-(2-saccharinyl)acetate that was used immediately in the next step.

A mixture of this chloride (0.66 g crude mixture) and 1-phenyl-5-mercapto-1H-tetrazole in the form of the sodium salt (0.44 g, 2.2 mmol) was heated under reflux in methyl ethyl ketone (25 ml) for 4 hours. After stirring at room temperature for 2 days, the reaction mixture was poured into ice water. The tan solid isolated by filtration was washed with water and air-dried. Chromatography on silica gel (MDC) yielded an off-white foam that was crystallized from ethanol to yield 0.36 g of methyl $2-(1-phenyl-1H-tetrazol-5-ylthio)-2-(2-saccharinyl)acetate as a white crystalline solid, mp <math>160-162 \, ^{\circ} \, \text{C}$. $^{1} \, \text{H}$ nmr (90 MHz, CDCl₃):3.8 (3H, s); 7.05 (1H, s); 7.4-8.1 (9H, m). FDMS: m/e 431 (M⁺).

Theor. for C ₁₇ H ₁₃ N ₅ O ₅ S ₂ :									
Found:	C, 47.33; C, 47.15;	H, 3.04; H, 3.09;							

Example 18

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To a suspension of 6.0 g (0.03 mol) of cuprous iodide in 100 ml of THF was added 25 ml of dimethyl sulfide, and the resulting yellow solution was cooled to -78° C and treated dropwise with a solution of 23 ml (0.06 mol) of a 3.0 M solution of phenyl magnesium bromide in diethyl ether. The resulting pale yellow range solution was stirred at -78° C under nitrogen for one hour and then treated with 3.02 g (0.03 mol) 2-cyclohexenone in 10 ml of THF. The resulting mixture was allowed to warm to 0° C over a two hour period, recooled to -78° C, treated with 15 ml hexamethyl-phosphoramide stirred for thirty minutes, treated with 8.0 g (0.09 mol) of methyl cyanoformate and allowed to warm to ambient temperature overnight. The reaction mixture was poured into 100 ml of 2N hydrochloric acid, and the organic phase was separated and the aqueous phase back extracted with MDC. The combined organic extracts were taken to dryness in vacuo, and the residue triturated with saturated ammonium chloride, then with water, then with brine and taken to dryness once again to give 3.2 g of methyl 2-phenylcyclohexan-6-one carboxylate as an oil.

The latter (3.0 g, 0.013 mol), 4.8 g (0.039 mol) benzyl mercaptan and 1.0 g Amberlyst(®) – 15 resin (Rohm and Haas) in chloroform was heated under reflux for twenty hours, the mixture treated with an additional 1.5 g resin and heated for an additional four hours. The mixture was then cooled to ambient temperature, filtered, the filtrate taken to dryness in in vacuo, the residue triturated with hexane and the solid collected by filtration to give 0.85 g (19%) of a mixture of methyl 2 – benzylthio – 6 – phenylcyclohex – 2 – ene carboxylate and methyl 2 – benzylthio – 6 – phenylcyclohex – 1 – ene carboxylate, 0.6 g (0.0018 mol) of which was heated with 2.0 g 2,3 – dichloro – 5,6 – dicyano – benzequinone in 25 ml toluene with stirring under nitrogen for twenty – four hours. The mixture was filtered through a pad of silica gel, eluting with 2:1 MDC:hexane and the eluate taken to dryness to give 0.3 g (67%) of methyl 2 – benzylthio – 6 – phenylben – zoate.

The latter 0.52 g (0.0016 mol) dissolved in 10 ml of MDC was diluted with 20 ml of acetic acid and 5 ml of water, the mixture cooled to -10°C, and chlorine gas was bubbled through the mixture until the exothermic reaction subsided. The mixture was then stirred for ten minutes and taken to dryness in vacuo, to give 0.41 g (85%) of methyl 2 - chlorosulfonyl - 6 - phenylbenzoate which was dissolved in 10 ml THF and added to 25 ml of a solution of concentrated ammonium hydroxide while cooling in an ice/acetone bath.

The reaction mixture was extracted with MDC, the organic phase discarded, and the aqueous layer acidified to pH 1 with concentrated hydrochloric acid and extracted with MDC. The organic extracts, on washing with brine, drying and evaporation to dryness, afforded 0.33 g (97%) of 4 – phenylsaccharin.

Following a procedure similar to that described in Example 21, the latter (0.33 g, 0.0012 mol) was reacted with 0.3 g (0.0019 mol) chloromethyl phenyl sulfide in 15 ml toluene in the presence of 0.08 g (0.0025 mol) tetrabutylammonium bromide and the product, 2-phenylthiomethy-4-phenylsaccharin (0.48 g, 100%), treated with sulfuryl chloride in MDC to give 0.36 g (95%) of 2-chloromethyl-4-phenylsac-charin.

Example 19

A solution of 2.2 g (0.0071 mol) 2-(phenylchloromethyl)-saccharin and 1.4 g (0.0071 mol) of the sodium salt of 1-phenyl-5-mercaptotetrazole in 30 ml of DMF was heated at 55° for three and one half hours, then stirred at ambient temperature for about sixteen hours and poured into ice water containing dilute sodium bicarbonate. The solid which separated was collected, washed with water, air dried and chromatographed on silica gel, eluting with 98:2 MDC: diethyl ether, to give 2 g (63%) of 2-(1-phenyl-1+tetrazol-5-ylthiophenylmethyl)saccharin, mp 192-193°C.

Example 20

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A solution of 4.53 g (0.022 mol) of the sodium salt of saccharin and 5 g (0.022 mol) 1 – phenyl – 4 – chloromethyl – tetrazolin – 5 – thione in 50 ml DMF was heated at 130° for four hours, then cooled and poured into ice water. The solid which separated was collected, washed with water, dried and chromatog – raphed on silica gel, eluting with MDC, to give 4.8 g (58%) of 2 - (1 - phenyl - 5 - thioxotetrazolin - 4 - ylmethyl)saccharin, mp 140 – 142°C.

Example 21

A mixture of 3.27 g (0.012 mol) 4 – bromosaccharin [Japanese Pat. Publich. 58/79,034, C.A. 100, 7773w (1984)], 1.63 g (0.015 mol) potassium t – butoxide, 0.39 g (0.0012 mol) tetrabutylammonium bromide and 3.0 ml (0.022 mol) chloromethyl phenyl sulfide in 100 ml toluene was heated under reflux under a nitrogen atmosphere for eight hours and then at ambient temperature for about sixteen hours. The reaction mixture was then cooled, diluted with ethyl acetate and the organic layer washed with bicarbonate, water and brine and then dried over magnesium sulfate and taken to dryness in vacuo. The residual solid was recrystallized from toluene – hexane to give 3.86 g (84%) of 4 – brom – 2 – phenylthiomethylsaccharin, mp 174.5 – 178° C.

To a solution of the latter (3.27 g, 0.0085 mol) in 85 ml MDC was added, dropwise with stirring, 1.02 ml (0.0127 mol) sulfuryl chloride. The mixture was stirred at ambient temperature for an hour and a half, concentrated in vacuo and the residue triturated with hexane and filtered to give 2.61 g of crude product which was recrystallized from toluene – hexane to give 2.24 g (85%) of 2 – chloromethyl – 4 – bromosac – charin, mp 157 – 159 °C.

Example 22A

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To a solution of 8.0 ml (0.053 mol) tetramethylethylenediamine (TMEDA) in 350 ml THF at -70° C was added 42 ml (0.055 mol) of a 1.3 M solution of s – butyl lithium in hexane and the mixture was stirred for fifteen minutes. To the solution was added dropwise with stirring a solution of 10.36 g (0.050 mol) 2 – methoxy – N,N – diethylbenzamide in 150 ml THF while maintaining the temperature at -60° C or below and sulfur dioxide then bubbled into the reaction mixture, keeping the reaction temperature below -50° C until the reaction mixture was acid to wet litmus paper. The mixture was then stirred at ambient temperature for two hours, diluted with 450 ml hexane, and the solid material which had separated was collected, dissolved in 200 ml water and the mixture treated with 65 g sodium acetate, and 21.5 g (0.19 mol) hydroxylamine – O – sulfonic acid was added in portions with stirring. The white solid which separated was collected and dried to give 7.04 g (49%) of $\frac{2-\text{aminosulfonyl}-6-\text{methoxy}-\text{N,N}-\text{diethylbenzamide}}{2-\text{aminosulfonyl}-6-\text{methoxy}-\text{N,N}-\text{diethylbenzamide}}$, mp $\frac{190-194.5^{\circ}\text{C}}{2-\text{methoxy}-194.5^{\circ}\text{C}}$

A mixture of the product (4.3 g, 0.015 mol) in 75 ml dioxane and 25 ml concentrated hydrochloric acid was heated on a steam bath for 70 hours, then cooled, concentrated in vacuo, diluted with water and ice and rendered strongly basic with concentrated sodium hydroxide. Extraction of the mixture with MDC and isolation of the product from the organic extracts afforded 1.29 g (40%) of 4-methoxysaccharin. In an

alternative, and preferred, procedure, cyclization of 2 - aminosulfonyl - 6 - methoxy - N,N - diethylbenzamide to 4 - methoxy - saccharin in 65% yield was carried out in refluxing glacial acetic acid for six and a half hours.

Following a procedure similar to that described in Example 21 above, 1.14 g (0.0053 mol) of the latter was reacted with 1.31 ml (0.0097 mol) chloromethyl phenyl sulfide in toluene in the presence of 0.72 g (0.00064 mol) potassium t – butoxide and 174 mg (0.00054 mol) tetrabutylammonium bromide to give 1.23 g (69%) of $\frac{4 - \text{methoxy} - 2 - \text{phenylthiomethylsaccharin}}{2 - \text{methoxy}}$, mp $\frac{152.5 - 154.5 \text{ °C}}{2 - \text{chloromethyl}}$ (from ethyl acetate – hexane), 1.02 g (0.003 mol) of which was treated with 0.36 ml (0.0045 mol) sulfuryl chloride in MDC to give 282 mg (36%) of 2 – chloromethyl – 4 – methoxysaccharin, mp $\frac{169 - 174 \text{ °C}}{2 - \text{chloromethyl}}$

Example 22B

To a solution of 4.74 ml (0.031 mol) tetramethylethylenediamine in 300 ml THF (passed through alumina prior to use) was added 5.8 g (0.03 mol) of 2-ethyl-N,N-diethylbenzamide. The solution was cooled to -78°C and treated with a solution of 34.9 ml (0.031 mol) 0.9 M solution of s-butyl lithium in cyclohexane. When addition was complete, the mixture was stirred for twenty minutes and then treated with a solution of 3.2 ml (0.04 mol) ethyl iodide while maintaining the temperature at -78°C. The temperature was then allowed to rise to ambient temperature and the mixture stirred for about sixteen hours and then poured into water. The resulting oil was separated and chromatographed on silica gel, eluting with 10% ethyl acetate/hexane to give 2.86 g (43%) of 2-sec. -butyl-N,N-diethylbenzamide as a yellow oil.

Following a procedure similar to that described in Example 22A above the latter (10.45 g, 0.045 mol), dissolved in 70 ml THF, was added to a solution of 39.2 ml (0.047 mol) 1.2 M solution s – butyl lithium in cyclohexane and 7.1 ml (0.047 mol) tetramethylethylenediamine in 250 ml THF while maintaining the temperature at –78°C. When addition was complete the mixture was stirred for an additional one half hour at –78°C and then treated with sulfur dioxide at –70°C and then allowed to warm to room temperature. The mixture was taken to dryness in vacuo, and the residue was dissolved in water and added with stirring to a cold solution 15.2 g (0.134 mol) of hydroxylamine – sulfonic acid and 15.4 ml (0.134 mol) 35% sodium hydroxide to give 10.1 g (72%) of 2 – aminosulfonyl – 6 – sec. – butyl – N,N – diethylbenzamide.

The latter (6.83 g, 0.22 mol) was dissolved in 100 ml glacial acetic acid and the solution heated under reflux for thirteen hours and then taken to dryness. The residue was triturated with diethyl ether and collected by filtration to give 5.7 g (83%) of the diethylammonium salt of 4 – sec. – butylsaccharin.

The latter (3.0 g, 0.0096 mol), on reaction with 1.13 ml (0.012 mol) chloromethyl phenyl sulfide in toluene, afforded 3.47 g (100%) of 2 – phenylthiomethyl – 4 – sec. – butylsaccharin.

Reaction of the latter (3.2 g, 0.0097 mol) with 2.3 ml (0.029 mol) sulfuryl chloride in 20 ml of MDC afforded 2.4 g (87%) of 2 - chloromethyl - 4 - sec. - butylsaccharin.

Example 22C

To a solution of 9.3 ml (0.058 mol) tetramethylethylenediamine in 340 ml THF at -78° C was added 52 ml 1.1 M solution (0.057 mol) s - butyl lithium in THF. The solution was then treated with a solution of 11.37 g (0.052 mol) 2-propyl-N,N-diethylbenzamide in 75 ml of THF at -78° C and the solution stirred for fifteen minutes and then treated with a solution of 8.3 ml (0.104 mol) of ethyl iodide in THF. The solution was stirred for an hour and a half at -78° C and then quenched by the addition of saturated ammonium chloride added dropwise at -78° C. The mixture was then allowed to warm to ambient temperature, diluted with diethyl ether, washed first with dilute hydrochloric acid, then with water, then with saturated sodium bicarbonate, then with brine dried and taken to dryness to give 12.91 g of crude product which was chromatographed on silica gel, eluting with 10% ethyl acetate/hexane to give 3.23 g (25%) of 2-(3-pentyl)-N,N-diethyl-benzamide as a yellow oil.

Following a procedure similar to that described in Example 22A above, the latter (3.05 g, 0.0115 mol) in THF was reacted with 10.5 ml (0.126 mol) 1.2 M solution of s – butyl lithium in THF in the presence of 2.1 ml (0.014 mol) tetramethylethylenediamine. The resulting lithium salt was then reacted first with sulfur dioxide and then with sodium hydroxylamine – sulfonate to give 1.97 g (52%) of 2 – aminosulfonyl – 6(3 – pentyl) – N,N – diethyl – benzamide as pale yellow crystals, mp 118 – 120 °C (soft 102°), 1.84 g (0.0056 mol) of which was cyclized in 22 ml refluxing glacial acetic acid to give 1.28 g (70%) of the diethylammonium salt of 4 – (3 – pentyl)saccharin, mp 107.5 – 109.5 °C.

The latter (0.0037 mol), on reaction with 0.74 ml (0.0055 mol) chloromethyl phenyl sulfide in the presence of 116 mg (0.0004 mol) tetrabutylammonium bromide in 45 ml toluene, afforded 1.93 g of 2-phenylthiomethyl-4-(3-pentyl)- saccharin as a pale yellow oil, 1.93 g (0.0037 mol) of which, on reaction

with 0.59 ml (0.0073 mol) sulfuryl chloride in 37 ml MDC, afforded 1.2 g of 2 - chloromethyl - 4 - (3 - pentyl)saccharin as a pale yellow oil.

Examples 22D - 22N

Following a procedure similar to that described above in Example 22A, substituting for the 2-methoxy-N,N-diethylbenzamide used therein an appropriate $2-R_3-R_4$ -substituted-N,N-diethylbenzamide, the following 2-halomethyl-4-R₃-R₄-substituted saceharins listed in TABLE A were prepared via the corresponding 2-phenylthiomethylsaccharins. Wherever available, the melting point, recrystallization solvent and yield are given for each of the 2-unsubstituted saccharins, the 2-phenyl-thiomethylsaccharins and the 2-chloromethylsaccharins in columns headed "mp/Solv." and "Yield". In all instances, the intermediate 2-phenylthiomethylsaccharins were used directly in the subsequent step without further characterization or purification.

5		cc.	Yield 41	20	76		64	\$	
10		2-CICH ₂ -Sacc.	mp/Solv158.0 - 160.0 i-PrOH	96.0 - 98.0 i-PrOH-Cyc. hex.	190.0 - 192.0	186.0 - 187.0	139.0 - 140.0	142.5 - 143.5	
15									_
20		H2-Sacc.	Xield 100	100	100	65	95	75	88
20	TABLEA	-C ₆ H ₅ SC	mp/Solv. Yield	ı	ŀ	:	!	:	ŀ
25		7	=						
30			Yield 93	88	\$	76	%	87	94
35		Sacc	_mp/Solv. 260 - 262	177.0 - 178.0 MeOH	(a)	(b) EtOAc-hex.	(a)		(a)
40			R3/R4 H 7 - CI	СН (СН ₃) ₂ Н	сн ₃ 0 5-СН ₃ 0	соосн ₃ Н	C ₂ H ₅ O H	(СН3) ₂ СНО Н	сн ₃ о 6-сн ₃ о
45			Ex. 22D	22E	22F	22G	22H	221	22J

	Yield 87	86	001	00	lphthalate in
,	2-Saco		-	Ē.	aminosulfonyl
	2-CICH.mp/Solv.	163 - 168 hexane	io	ŀ	salt. of dimethyl 3-
,	'2-Sacc. Xield 100	52	100	100	ammonium cyclization
TABLEA (cont'd)	2-C ₆ H ₅ SCH ₂ -Sacc. mp/Solv. Yield 100	;	ı	ı	p as the diethylvas prepared by
IAI	Yield 83	<i>L</i> 9	70		the next ste- saccharin v
1	Saccmp/Solv.	240 - 243 i-ProH	2 107.5 - 109.5 Me t-Bu ether: hex.	(c)	solated and used in the next step as the diethylammonium salt. The 2-unsubstituted-saccharin was prepared by cyclization of dimethyl 3-aminosulfonylphthalate in
) i		C ₂ H ₅ 5,7-(CH ₃ O) ₂	CH (C ₂ H _S) ₂ H	C ₆ H ₅ H	(a) Isola (b) The
	Ex. 22K	22L	22M	22N	

by diazotization of dimethyl 3-aminophthalate, decomposition of the diazonium salt with sulfur dioxide

in the presence of cupric chloride and reaction of the resulting dimethyl 2-chlorosulfonylphthalate with

See Example 21B for preparation of 2-unsubstituted-saccharin.

ammonia. (84% yield overall).

Example 23

50

Following a procedure similar to that described in Example 1, reaction of 18.3 g (0.1 mol) saccharin with 70 ml 37% formalin in ethanol afforded 3.58 g (70%) of 2-hydroxymethylsaccharin, 25 g (0.117 mol) of which was reacted with 63.3 g (0.234 mol) phosphorus tribromide in diethyl ether to give 29.8 g (92%) of 2 - bromomethylsaccharin, mp 155 - 157°C.

Example 24

To a solution of 4 g (0.0175 mol) 6 – nitrosaccharin in 240 ml ethanol was added 4.4 g (0.0175 mol) thallium ethoxide, and the mixture was allowed to stand at room temperature for one hour, cooled for ahout 16 hours and the precipitated solid collected and dried to give 7.6 g (100%) of the thallium salt of 6 – nitrosaccharin. The product was suspended in 50 ml DMF and the mixture treated with 3.07 g (0.0194 mol) chloromethyl phenyl sulfide, the mixture warmed at about 63°C for five hours, allowed to stand at ambient temperature for about 16 hours, and then poured into ice water. The crude product, obtained by filtration, was stirred in MDC and filtered to remove thallium salts. The filtrate was freed of solvent, and the resultant pale yellow solid was sonicated with warm ethanol and once again collected and dried to give 4.6 g (75%) of 6 – nitro – 2 – phenylthiomethylsaccharin, mp 161 – 163°C. The latter, on reaction with sulfuryl chloride in MDC using the procedure described above in Example 17 afforded3.7 g of 2 – chloromethyl – 6 – nitrosac – charin.

15 Example 25A

A solution of 49.8 g (0.199 mol) 2-hydroxy-5-(1,1,3,3-tetramethylbutyl)benzoic acid in 200 ml methanol was heated to 50°C and then treated dropwise with about 80 g sulfuric acid at a rate to maintain the reaction under reflux. The reaction mixture was heated under reflux for an additional 11 hours, then cooled and partitioned between water and ethyl acetate. The organic layer was washed with saturated sodium bicarbonate, then with brine, dried over sodium sulfate and taken to dryness to given 48.6 g (92%) of methyl -2-hydroxy-5-(1,1,3,3-tetramethylbutyl)benzoate.

The latter dissolved in 250 ml DMF was treated first with 40.4 g (0.36 mol) 1,4 – diazabicyclo[2.2.2] – octane followed by 3.4 g (0.27 mol) dimethyl thiocarbamoyl chloride and 100 ml DMF. The reaction mixture was heated at 45 °C for about eight hours, cooled, poured into ice/water and concentrated hydrochloric acid and then extracted with ethyl acetate. The combined organic extracts were washed with dilute hydrochloric acid, then with sodium bicarbonate and then with brine, dried and taken to dryness to give 48.2 g (76%) of methyl 2 – (N,N – dimethylthiocarbamyloxy) – 5 – (1,1,3,3 – tetra – methylbutyl)benzoate which was heated at 220 °C for 15 hours, then cooled, dissolved in toluene and chromatographed on silica, eluting with 1:9 ethyl acetate:toluene, to give 3.6 g (14%) of methyl 2 – (N,N – dimethyl – carbamylthio) – 5 – (1,1,3,3 – tetramethylbutyl)benzoate.

A solution of the latter (0.025 mol) in 40 ml MDC was treated, with stirring, with 80 ml glacial acetic acid, followed by 16 ml water. The reaction mixture was cooled to 0°C, and chlorine was bubbled through the reaction mixture for about five minutes while maintaining the temperature between 5 and 24°C. The reaction was stirred for an additional 30 minutes, concentrated in vacuo, and the remaining solution poured into ice water. Extraction of the mixture with ethyl acetate and isolation of the product from the combined organic extracts afforded 6.8 g (78%) of methyl 2-chlorosulfonyl-5-(1,1,3,3-tetramethyl-butyl)-

The product (9.0 g, 0.026 mol) was dissolved in THF and added to 100 ml concentrated ammonium hydroxide with cooling in an ice bath. The resulting solution was stirred for about 16 hours, then concentrated in vacuo and the concentrated solution acidified to pH 3 with concentrated hydrochloric acid. The mixture was stirred for several hours, and the separated solid collected, washed with water and dried to give 9.0 g of 5 – (1,1,3,3 – tetramethylbutyl)saccharin, mp 213 – 215 °C.

Following a procedure similar to that described in Example 17, 9.0 g (0.30 mol) of the product was reacted with thallium ethoxide in ethanol and the resulting thallium salt reacted with 3.33 g (0.021 mol) chloromethyl phenyl sulfide in DMF to give 5.76 g (66%) of 2-phenylthiomethyl-5-(1,1,3,3-tetramethylbutyl)saccharin, 3.3 g (0.007 mol) of which was treated with 0.944 g sulfuryl chloride in MDC to give 1 g (41%) of 2-chloromethyl-5-(1,1,3,3-tetramethylbutyl)saccharin

Example 25B

Following a procedure similar to that described in Example 25A above, 15.5 g (0.086 mol) ethyl 2-hydroxy-6-methylbenzoate was reacted with 15.9 g (0.129 mol) N,N-dimethylchlorothiocarbamate in the presence of 19.3 g (0.172 mol) 1,4-diazabicyclo [2.2.2] octane in DMF to give 22.1 g (96%) of ethyl 2-(N,N-dimethylthiocarbamyloxy)-6-methylbenzoate which was heated at 220°C for about 10 hours. The product was purified by chromatography on silica gel in MDC to give ethyl 2-(N,N-dimethylcarbamyl-thio)-6-methylbenzoate as a red-brown oil.

A solution of the latter (22.6 g, 0.0884 mol) in 170 ml MDC was treated with 340 ml glacial acetic acid and 68 ml water while cooling in an ice/acetone bath, and chlorine was bubbled through the reaction mixture for 10 – 15 minutes. The reaction vessel was evacited to remove excess chlorine and MDC and the mixture poured into water and partitioned between MDC and water. The organic layer, on drying and evaporation to dryness, afforded 19 g of ethyl 2 – chlorosulfonyl – 6 – methylbenzoate, 5 g (0.019 mol) of which was reacted with concentrated ammonium hydroxide in THF to give 6.1 g (67%) of 4 – methylsaccharin.

Following a procedure similar to that described in Example 17 above, the product (10.1 g, 0.0512 mol) was converted to the thallium salt by reaction with 12.8 g (0.0512 mol) thallium ethoxide in ethanol and the thallium salt reacted with 6.7 g (0.0427 mol) chloromethyl phenyl sulfide in DMF to give 6.85 g (50%) of 2-phenylthiomethyl - 4 - methylsaccharin.

Reaction of the latter (6.7 g, 0.021 mol) with sulfuryl chloride in MDC afforded 4.9 g (95%) of 2 – chloromethyl – 4 – methylsaccharin.

Example 26A

15

A mixture of 75 g (0.36 mol) 3,3 – dithiobispropionic acid,102 ml thionyl chloride and a catalytic amount of pyridine was stirred for about 24 hours and then evaporated to dryness $\underline{\text{in vacuo}}$. The residue was treated with MDC and evaporated to dryness again to remove residual thionyl chloride and pyridine to give 87 g (98%) of the corresponding bis acid chloride, 44.8 g (0.18 mol) of which was dissolved in THF and added dropwise to a solution of 77.16 g (0.72 mol) of benzylamine in THF. The mixture was stirred for two hours at $40-45^{\circ}$ C, cooled and the precipitated solid collected, washed with water and dried to give 59 g (84%) of 3,3 – dithiobispropionic acid N,N' – dibenzyl – carboxamide, mp $162-165^{\circ}$ C.

Reaction of 7.0 g (0.018 mol) of the latter with 10.25 g (0.076 mol) sulfuryl chloride in MDC gave a mixture of 2-benzyl-2H-isothiazol-3-one and 5-chloro-2-benzyl-2H-isothiazol-3-one which were largely separated from one another by sonication in MDC (which solubilized most of the former). The insoluble material was collected by filtration and chromatographed on silica gel with MDC. There was thus obtained 5-chloro-2-benzyl-2H-isothiazol-3-one, mp 58-68°C.

A solution of 10 g (0.044 mol) of the latter in MDC was cooled to 0°C and the solution treated with 7.6 g (0.044 mol) 3-chloroperbenzoic acid, the mixture stirred for 10 minutes and then treated with a second 7.6 g portion of the perbenzoic acid. The reaction mixture was filtered, the filter washed with MDC and the filtrate washed with saturated sodium bicarbonate, then with brine, dried over sodium sulfate and taken to dryness and the residue chromatographed in MDC on silica gel, the product being eluted with 50:50 hexane:MDC, to give 7.15 g (46%) of 5-chloro-2-benzyl-2H-isothiazol-3-one-1-oxide.

A solution of 1.1 g (0.0045 mol) of the latter in 8 ml benzene was treated with 0.55 g (0.0051 mol) 2 – methoxyfuran and the solution heated in a pressure bottle at 70 °C for 1 1/2 hours and then cooled and the solid collected, washed with benzene and dried to give 2 – benzyl – 7 – hydroxy – 4 – methoxybenzisothiazol – 3 – one – 1 – oxide,mp 235 – 237 °C.

A mixture of the product (1.85 g, 0.006 mol), 2.48 g (0.018 mol) potassium carbonate and 1.70 g (0.012 mol) methyl iodide in acetone was heated under reflux for 1 1/2 hours and then cooled and poured into water. The solid which separated was collected by filtration, washed with water and dried to give 1.70 g (89%) of 2 - benzyl - 4,7 - dimethoxybenziso - thiazol - 3 - one - 1 - oxide, 1.13 g (0.0035 mol) of which was oxidized with 1.20 g (0.007 mol) 3 - chloroperbenzoic acid in MDC using the procedure described above to give 1.03 g (88%) of 2 - benzyl - 4,7 - dimethoxysaccharin.

A mixture of 2.07 g (0.0062 mol) of the product, 1.37 g (0.02 mol) ammonium formate and 1.5 g 10% palladium – on – charcoal catalyst in 80 ml methanol was heated under reflux for one hour, then cooled and filtered, and the filtrate taken to dryness to give 0.92 g (57%) of the ammonium salt of 4,7 – dimethoxysac – charin.

A solution of 1.11 g (0.0042 mol) of the ammonium salt was dissolved in DMF, 0.67 g (0.0042 mol) chloromethyl phenyl sulfide was added, and the solution heated under reflux for eight hours and then cooled and poured into ice water. The solid which separated was collected, washed with water and dried to give 0.50 g (33%) of 2 – phenylthiomethyl – 4,7 – dimethoxysaccharin.

Reaction of the latter (0.5 g, 0.0013 mol) with sulfuryl chloride in MDC using the procedure described above in Example 17 afforded 0.22 g (58%) of 2 – chloromethyl – 4,7 – dimethoxysaccharin.

Examples 26B and 26C

Following a procedure similar to that described in Example 26A, other 2-chloromethylsaccharin derivatives were prepared as follows:

Example 26B

Reaction of 5.8 g (0.024 mol) 5-chloro-2-benzyl-2H-isothiazol-3-one with 3.76 g (0.0335 mol) 2-ethoxyfuran afforded 3.05 g (40%) 2-benzyl-4-ethoxy-7-hydroxybenz isothiazol-3-one-1-oxide, 5.7 g of which was reacted with 3.6 g (0.0197 mol) 2-[2-methoxyethoxy] ethyl bromide in the presence of 4.95 g (0.0358 mol) potassium carbonate in 125 ml methylethyl ketone and 25 ml DMF to give 7.0 g (93%) of 2-benzyl-4-ethoxy-7-[2-(2-methoxyethoxy)ethoxy]benzisothiazol-<math>3-one-1-ox-ide, which was oxidized as before with 3-chloroperbenzoic acid in MDC to give 2-benzyl-4-ethoxy-7-[2-(2-methoxyethoxy)ethoxy]saccharin.Debenzylation of 6.6 g (0.015 mol) of the latter with <math>3.34 g (0.053 mol) ammonium formate in the presence of 6.4 g 10% palladium-on-charcoal catalyst in methanol afforded the ammonium salt of 4-ethoxy-7-[2-(2-methoxyethoxy)ethoxy]saccharin, which was reacted with 2.38 g $\overline{(0.015 mol)}$ chloromethyl phenyl sulfide in 100 ml DMF to give 1.46 g (21%) of 2-benzyl-4-ethoxy-7-[2-(2-methoxyethoxy)ethoxy]-saccharin, mp 73-75°C (from isopropanol). Treatment of 1.4 g (0.0029 mol) of the product with 0.4 g (0.0029 mol) sulfuryl chloride in MDC afforded 1.16 g (100%) 2-chloromethyl-4-ethoxy-7-[2-(2-methoxy-ethoxy-ethoxy-ethoxy)ethoxy]-saccharin.

Example 26C

Reaction of 3.03 g (0.01 mol) 2-benzyl-7-hydroxy-4-methoxybenzisothiazol-3-one-1-oxide (Example 26A) with 2.01 g (0.011 mol) 2-(2-methoxyethoxy)ethyl bromide in methyl ethyl ketone in the presence of 2 g (0.015 mol) potassium carbonate afforded 2.58 g (64%) of 2-benzyl-4-methoxy-7-[2-(2-methoxyethoxy)ethoxy]benzisothiazol-3-one-1-oxide, which, on oxidation with 1.1 g (0.0063 mol) 3-chloroperbenzoic acid in MDC, gave <math>2-benzyl-4-methoxy-7-[2-(2-methoxyethoxy)ethoxy]-saccharin. Debenzylation of 0.25 g (0.0006 mol) of the product with 0.13 g (0.0021 mol) ammonium formate in methanol in the presence of 0.25 g 10% palladium-on-charcoal gave 0.21 g (100%) of the ammonium salt of 4-methoxy-7-[2-(2-methoxyethoxy)ethoxy]saccharin. Reaction of 1.4 g (0.004 mol) of the ammonium salt with 0.63 g (0.004 mol) chloromethyl phenyl sulfide in DMF afforded 2-phenylthiomethyl-4-methoxy-7-[2-(2-methoxyethoxy)ethoxy]saccharin, which, on reaction with sulfuryl chloride in MDC, afforded 0.53 g (35%) of <math>2-chlor-methyl-4-methoxy-7-[2-(2-methoxyethoxy)ethoxy]saccharin.

Example 27

30

A solution of 1.89 g (0.011 mol) diethylamino sulfur trifluoride (DAST) in 20 ml MDC was added to a suspension of 2.13 g (0.01 mol) 2-hydroxymethylsaccharin in 25 ml MDC while maintaining the reaction mixture at -78° C.

The reaction mixture was stirred at -78°C for one hour, the temperature was then allowed to slowly rise to ambient temperature and the mixture stirred for 16 hours and then poured into ice-water. The organic layer was separated and washed with water, dried over magnesium sulfate and taken to dryness to give 2.2 g of product which was recrystailized from ethyl acetate to give 1.6 g (74%) of 2-fluoromethyl-saccharin, mp 96-98°C.

Example 28A

To a solution of 0.5 g (0.0025 mol) 4 – methyl – saccharin in THF cooled to –78°C by a dry ice/acetone bath was added, dropwise with stirring, a solution of 5.2 ml 1.3 M solution of s – butyl lithium in THF. The mixture was stirred an additional hour at –78°C and then treated with 0.16 ml (0.025 mol) methyl iodide over a 1 1/2 hour period. The mixture was stirred for an hour and 45 minutes, quenched in 25 ml 1N hydrochloric acid, the reaction mixture rendered basic, the aqueous mixture extracted with chloroform and then acidified and extracted with ethyl acetate. The combined organic extracts were washed with 10% sodium thiosulfate, then with brine, dried over sodium sulfate and taken to dryness to give a product, whose PMR spectrum indicated a mixture consisting of 74% of 4 – ethylsaccharin and 21% of 4,7 – dimethylsac – charin.

Following a procedure similar to that described in Example 17 above, the crude material (0.47 g, 0.022 mol) was reacted with 0.24 ml (0.0028 mol) chloromethyl phenyl sulfide in toluene in the presence of tetrabutylammonium bromide, and the product chromatographed on silica gel, eluting with MDC, 5 ml fractions being collected. The first 420 ml of eluate were discarded. The next 20 fractions, on evaporation, afforded 0.07 g of material, predominantly the 4,7-dimethylsaccharin, which was set aside. The next 25 fractions afforded 0.37 g of 2-phenylthiomethyl-4-ethyl-saccharin, which was reacted with sulfuryl

chloride in MDC to give 0.19 g (66%) of 2-chloromethyl-4-ethylsaccharin.

Example 28B

5

Following a procedure similar to that described in Example 28A 10 g (0.051 mol) 4-methylsaccharin was reacted with 86 ml (0.10 mol) 1.18 M solution of s-butyl lithium in THF and the resulting solution treated with 4.5 ml (0.050 mol) ethyl iodide to give 10.15 g (89%) of 4-propylsaccharin, which on reaction with 5.32 ml (0.056 mol) chloromethyl phenyl sulfide in toluene in the presence of tetrabutylammonium bromide afforded a 65% yield of 2-phenylthiomethyl-4-propylsaccharin as an oil, 1.8 g (0.0052 mol) of which, on reaction with 1.25 ml (0.016 mol) sulfuryl chloride in MDC afforded 0.94 g (66%) of 2-chloromethyl-4-propylsaccharin.

Example 29

The 0.07 g sample of material obtained in the early fractions from the chromatographic separation described above in Example 28A was reacted with 0.05 ml sulfuryl chloride in MDC and the product recrystallized from cyclohexaneethyl acetate to give 20 mg (51%) 2-chloromethyl-4,7-dimethyl-saccharin, mp 107-108°C.

Examples 30A - 30BK

Following a procedure similar to that described in Example 4 above, substituting for the 2-chloromethyl-4-chlorosaccharin and the sodium 1-phenyltetrazole salt used therein, molar equivalent amounts of a respective appropriate 2-halomethyl-4-R₃-R₄-substituted saccharin and an appropriate L_nR₁ moiety, the following compounds in TABLE B below were prepared. The identities of the halogen moiety in the 2-halomethylsaccharin and the base used to catalyse the reaction, i.e., either the sodium or thallium salt of the L_nR₁ reactant or the added basic catalyst, i.e., potassium carbonate, triethylamine (TEA), ethyldiisopropylamine (EDIPA) or sodium methoxide, are given in the column headed "X/Base". The reaction solvent used (DMF, THF, MDC, MEK or acetone) is given in the column headed "Sol", and the melting point (mp) and the solvent used for recrystallization are given in the column headed "mp/From". In TABLE B, and in the other tables which follow, various heterocyclic or other groups, R₁, are abbreviated as follows:.

tet. tetrazolyl
triaz. triazolyl
Mor. morpholinyl
thiadiaz. thiadiazolyl
imidaz. imidazolyl
benzthiaz. benzothiazolyl
benzoxaz. benzoxazolyl

45

40

35

50

		Yield	<i>L</i> 9		8		14		42		9
5		mp/Erom	168-170 EtOH	Foams-no mp	187-189	EtOH-CH ₃ CN	201-202	i-ProH-CH ₃ CN	126-129	Etoh-CH ₃ CN	170-172 EtOH-CH ₃ CN
15		Solv.	DMF	THF	DMF		MDC		MEK/DMF		MEK
20		X/Base	Cl Na Salt	Br TEA	Вr	TEA	Ō	TEA	ರ	K2CO3	Br K ₂ CO ₃
25											
30	TABLEB	T/u	1 ;C (CH ₃) ₂ S	0	1	S		S		S	- v
35		R3/R4	н 5- (СН ₃) ₃ ССН ₂ С (СН ₃₎₂	нн	н	π	Ξ	н	I	π	ππ
45		R1/R2	1-C ₆ H5-5tet. H	4-Mor.SO ₂ C ₆ H ₄ H	1-(3-NH ₂ C ₆ H ₄)- 5-tet.	I	1-(4-HOOCC ₆ H ₄)- 5-tet.	H	1-[3,5-(MeOOC) ₂ - C ₆ H ₃]-5-tet.	Д S	2-pyrimidinyl H
50		Ēx.	30A	30B	30C		30D		30E		30F

		Yield	15		68								74	
10		mp/Erom	233-234		162-164				oil		oil		94-96	
15		Solv.	DMF		Acet.	MDC	DMF		MDC		MDC		MEK	
20	(þ.	X/Base	Br	TIOE	ប	Na Salt	Br	TIOEt	ū	TEA	Br	TEA	ប៊	Na Salt
25	TABLEB (cont'd)	n/L	-	0	-	S	1	S	1	S	-	S	-	S
35	IA	R3/R4	н	Ħ	CH ₃	H	Ξ	H	C ₂ H ₅	н	π	Ξ	C ₂ H ₅ O	$7-CH_3(OCH_2CH_2)_2O$
40 45		R1/R2	6-(1-oxophen-	H	1-C ₆ H ₅ -5-tet.	H	1-C ₆ H ₅ -2-(1,3,4- triaz.)	五	1-(4-Mor. CH ₂ - CH ₂)-5-tet.	Ξ	1-(4-Mor.CH ₂ -CH ₂)-5-tet.	H	-tet.	.7
50		Ēx.	30G		30H		301		30J		30K		30L	

		Yield		8				61		53	
5		mp/Erom	104-106	198-200	EtOH-CH ₃ CN	Dec. 110	і-Р•ОН	188.5-190.5	Etoh	149.0-150.0	EtOH-CH ₃ CN
,,		1	N.	(1				Į,		Į,	
15		Solv	MEK	DMF		MDC		DMF		DMF	
20	ťd)	X/Base	C1 Na Salt	Br	TEA	Br	TEA	Br	TEA	Br	TEA
25	TABLE B (cont'd)	n/L	- v	-	S	0	ł	-	S	1	S
30	Ŧ	ŀ) 42)20								
35		R3/R4_	CH3O 7-CH3(OCH2CH2)2O	Ξ	H	I	н	H	Ξ	н	Ή
40		R1/R2		o- -triaz.		OC ₆ H ₄)- -1-tet.		5-(CyclohexNH)-	ullaula <i>2.</i> .)	dyl)-	
45			1-C ₆ H ₅ -5-tet. H	3-(pyrido- [2,1-c]-s-triaz	Œ	4-(3-HOOC ₆ H ₄)- 5-thioxo-1-tet.	Ħ		H	I-(3-pyridyl)- 5-tet.	Ξ
50		Ex	30M	30N		30-0		30P		300	

		Yield	∞		42				73		69		15	
5		mp/From	108.0-110.0	EtOAc	225-227	CH ₃ CN	192-193		164-165	EtOH-CH ₃ CN	185.5-188.0	EtOH-CH ₃ CN	133.5-135.0	EtOAc-hex
15		Solv.	DMF		DMF		DMF/MEK		DMF		DMF		снзон	
20	(р	X/Base	Br	NaH	ਹ	K ₂ CO ₃	ū	Na Salt	Ü	Na Saft	ច	Na Salt	В	NaOMe
25	TABLE B (cont'd)	n/L	0	1	-	S	-	S	-	S	-	S	0	:
30	TABI													
35		R3/R4	Ξ	Ξ	CH3	н	CH ₃ O	7-CH ₃ O	CH ₃ O	Ξ	Br	Ξ	н	Ι
40		1	ند ،		HZ .;								٨	
45		R ₁ /R ₂	4-(3-pyridyl)- 5-thioxo-1-tet.	I	4-(3-CH ₃ CONH- C ₆ H ₄)-5-triaz.	Ξ	1-C ₆ H ₅ -5-tet.	I	$1-C_6H$	Η	1-C ₆ H ₅ -5-tet.	Η	2-Me-5-thioxo-1-tet.	Н
50		Ex.	30R		308		30T		30U		30V		30W	

		Yield	17	39		11		52	75		88	
5		mp/Erom	182.0-183-0 EtOAc-hex	138.5-140.5	БОН	157.0-159	Еюн	88.0-90.0 Et ₂ O-hex	114-116	і-Рюн	187-189	EtOH
10										-		
15		Solv.	СН3ОН	DMF		DMF		DMF	DMF		DMF	
20	ont'd)	X/Base	Br NaOMc	ច	TEA	Ö	TEA	Cl Na Salt	Br	TEA	Br	TEA
25	TABLE B (cont'd)	n/L	- v	-	S	0	1	- v		S	-	S
30	H	ı										
35		R3/R4	нн	н	Ή	(CH ₃) ₂ CH	Ħ	(C ₂ H ₅) ₂ CH H	Ŧ	Ή	н	н
40		İ										
45		R1/R2	2-Me-5-tet. H	1-(3-pyridyl)- 5-tet.	H	4-(3-pyridyl)- 5-thioxo-1-tet.	Ξ	30AA 1-C ₆ H5-5-tet. H	5-Me-2-(1,3,4-thiadiaz.)	н	30AC 1-Me-2-(1,3,4- triaz.)	H
50		Ex.	30X	30Y		305		30AA	30AB		30AC	

		Yield	88		43	33		30	32	
5		mp/From	132-134	EtOAc-hex	180.0-182.0 EtOAC-CHCl ₃	117.0-119.0	і-Рюн	166.0-168.0 CH ₃ CN	201.0-203.0	EtOH
10							· -	_ 0	.,	_
15		Solv.	DMF		DMF	DMF		DMF	БЮН	
20 .	(p,	X/Base	Br	TEA	C! Na Salt	Br	TEA	C! Na Salt	Br	NaOMe
25	TABLEB (cont'd)	n/L	0	ı	- v	-	S	- S		S
30	H									
35		R3/R4	Ħ	Н	C ₂ H ₅ H	Ξ	Н	H 7-Cl	Ξ	H
40		-R ₁ (R ₂	thioxo- hiadiaz.)		-tet.	- iadiaz.)	Н	-tet. 1	1,3,4-	_
45		\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	2-MeS-5-thioxo- 4-(1,3,4-thiadiaz.)	Ξ	1-C ₆ H5-5-tet. H	4-C ₆ H ₅ -2- (1,3,5-thiadiaz.)	, L	30AG 1-C ₆ H5-5-tet. H	5-HS-2-(1,3,4-thiadiaz.	H
50		Ex.	30AD		30AE	30AF		30AG	30AH	

		Yield		16			24	74		51
5		mp/Erom	158.0-160.0 EtOH	178.0-179.5	БОН	140.0-141.0 EtOH	109.0-111.0 EtOH-CH ₃ CN	72-74		137.0-139.0 EtOH-H ₂ O
10										
15		Solv.	DMF	DMF		DMF	DMF	DMF		DMF
20	ф)	X/Base	CI Na Salt	Ū	Na Salt	CI Na Salt	Cl Na Salt	ō	TEA	CI TEA
25	TABLE B (cont'd)	n/L	- v	0	ŀ	- v	- v	-	S	- v
30	IAB			Ħ.		СН		усн		усн
35		R3/R4	CH ₃ O 5-CH ₃ O	(CH ₃) ₂ CH	Н	(CH ₃) ₂ CH H	n-С3Н7 Н	(СН3)2СН	Ħ	(СН ₃₎₂ СН Н
40		-R1/R2	-tet.	. ಕ		itet.	tet.	,1,3,-)		ë:
45		RI	1-C ₆ H5-5-tet. H	4-C ₆ H ₅ -5- thioxo-1-tet.	H	30AK 1-C ₆ H5-5-tet. H	30AL 1-C ₆ H5-5-tet. H	30AM 5-CH3-2-(1,3,- 4-thiadiaz.)	I	30AN 1-CH ₃ -5-tet. H
50		Ex.	30AI	30AJ		30AK	30AL	30AM		30AN

		Yield	52	53		8		92		. 6	
5		mp/Erom	120.0-122.0 EtOAc-hex	142.0-143.0	MDC-hex	148.0-150.0		138.0-139.5		178.5-179.5	
15		Solv.	DMF	DMF		MDC		MDC		MDC	
20		X/Base	Cl Na Salt	Br	TEA	Br	EDIPA	Вг	EDIPA	Br	EDIPA
25	TABLE B. (cont'd)	T/u	- v	-	S	1	S		S	-	S
30	TABL		СН₃СНС₂Н₅ Н								
35		R3/R4_	СНЗСІ	I	H	Ξ	H	Ħ	Ħ	H	H
40						42)-	-	ó.	•	12-	
45 .		R ₁ (R ₂	30AO 1-C ₆ H5-5-tet. H	30AP 1-EtO ₂ CCH ₂ - 5-tet	H	30AQ 1-(HOCH ₂ CH ₂)- 5-tet.	H	30AR 1-C6H5-4-COO-	H	30AS 1-Me ₂ NCOCH ₂ - 5-tet.	H
50		Ex.	30AO	30AP		30AQ		30AR		30AS	

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		Yield	62	\$	80	82	72
5		mp/From	139.5-140.5	124.5-125.5	146.5-148	180.0-181.5	146.0-147.0
10							
15		Solv	DMF	DMF	MDC	MDC	MDC
20	(X/Base	CI Na Salt	CI Na Salt	CI EDIPA	CI EDIPA	CI TEA
25	TABLEB (cont'd)	T/u	- v	- v	- v	- v	- s
35	IA	R3/R4	C ₂ H ₅ O H	(СН3) ₂ СНО Н	C ₂ H ₅ H	(СН ₃)2СН Н	C ₂ H ₅ O H
40					12-	-5-	12-
45		R ₁ /R ₂	30AT 1-C ₆ H5-5-tet.	30AU 1-C ₆ H5-5-tet. H	1-Me ₂ NCOCH ₂ . 5-tet. H	30AW 1-Me2NCOCH2- 5-tet. H	30AX 1-Me ₂ NCOCH ₂ - 5-tet. H
50		Ex	30AT	30AU	30AV	30AW	30AX

	Yield		4		51		70	8		
5	mpÆrom		164.0-165.0	CH ₃ CN	224.0-226.0	EtOAc	175.0-177.0	170.0-172.0	185-186	CH ₃ CN
15	Solv.	DMF	Еюн		DMF		DMF	DMF	MEK	
20	r'd) X/Base	Cl Na Salt	Br	NaOMe	ប	Tl Salt	Br TI Salt	Cl Na Salt	Br	K ₂ CO ₃
25 30	TABLEB_(cont'd) n/L X	- v		S	-	0	1 0	- S	1	S
35	R3/R4	СН ₃ О 6-СН ₃ О	Ή	н	(CH ₃) ₂ CH	π	нн	C ₆ H ₅ H	н	н
40				•						
45	R1/R2	30AY 1-C ₆ H5-5-tet. H	5-NH ₂ -2-(1,3,-	4-thiadiaz.) H	30BA 1-oxo-6-phen-	alenyl H	2,6-Cl ₂ C ₆ H ₄ H	30BC 1-C ₆ H5-5-tet. H	30BD 6-NO ₂ -2-	benzthiaz. H
50	Ex.	30AY	30AZ		30BA		30BB	30BC	30BD	

	Yield	88	33	74	74	•	8
5	mp/Erom	161-163 Ch.CN		EtOH-CH ₃ CN 110-113	EtOH 153-155	ЕЮН	
15	Solv	DMF	DMF	MDC	MEK	ļ	MDC
20	d) X/Base	Br	II Salt Br	K Salt Br	TEA Br	K ₂ CO ₃	CI TEA
25	TABLEB (cont'd) n/L X/J	- 0	n o	: -	o o	:	. v
30							(СН ₂₎₂ СН Н
35	R3/R4	н :	ш ш	ж ж	н н	н	(CH ₂
40	R1/R2		nidyl	.CH2-	it. -oxo-		.CH ₂ -
45	R	30BE 6-NO ₂ -2- benzoxaz.	H 30BF 2-phthalimidyl	H 30BG 1-(4-Mor.CH ₂ -	CH ₂)-5-tet. H 30BH 4-C ₆ H ₅ -5-oxo-	1-tet.	1 -(4-Mor.CH ₂ - CH ₂)-5-tet. H
50	Ex	30BE	30BF	30BC	308		30BI

Yield		
mp/Erom 170-172 EtOH-CH ₃ CN	162-164 FtOH-H-O	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\
MEK Solv	DMF	
r'd) X/Base Br K2CO3	D S	ואם ספון
TABLEB (cont'd) n/L X s S K	- <i>u</i>	י
R ₃ /R ₄	C ₂ H ₅	ごうじょうごう
40	et.	
Ex. R ₁ /R ₂ H	30BK 1-C ₆ H5-5-tet.	C
30BJ	30BK	

Examples 31A - 31C

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Following a procedure similar to that described above in Example 17, substituting for the saccharin and the methyl 2 – chloro – 2 – phenylthioacetate used therein molar equivalent amounts of a respective appro – priate $4 - R_3 - R_4$ – substituted saccharin and an appropriate $CI - CHR_2 - S - R_1$ moiety, the following com –

pounds shown in TABLE C, were similarly prepared where in each instance, n is 1 and L is -S-. In each case, the thallium salt of the saccharin derivative was used, and the reactions were carried out in DMF.

5		Yield	15		8							74
10		mp/Erom	233-234		162-164			oil		·		94-96
15												
20		Solv	DMF		Acet. MDC	DMF		MDC		MDC		MEK
25	ıt'd)	X/Base	B	TIOEt	Cl Na Salt	Вr	TIOE	ರ	TEA	Br	TEA	CI Na Salt
30	TABLEB (cont'd)	n/L		0	- S	-	S	-	Ø	-	S	- v
35	.,	R3/R4	_		СН3	н	Ι	C2H5	H	π	н	C ₂ H ₅ O 7.CH ₃ (OCH ₂ CH ₂) ₂ O
40		R3	I	н	O	-		S	_) 7-CH ₃ (OC
45		R1/R2	6-(1-oxophen-	alenyl) H	1-C ₆ H5-5-tet. H	1-C ₆ H ₅ -2-(1,3,4-	tnaz.) H	1-(4-Mor.CH ₂ -	CH ₂)-5-tet. H	1-(4-Mor.CH ₂ -	CH ₂)-5-tet. H	30L 1-C ₆ H5-5-tet. H
		Ex	30G		30H	30I		303		30K		30L

Example 32A

A solution of 0.28 g (0.00067 mol) 2-(2,6-dichlorophenylthiomethyl)saccharin in 5 ml MDC was treated with about 0.3 g (0.0017 mol) 3-chloroperbenzoic acid with stirring, and the mixture was stirred for about 16 hours and quenched with aqueous 10% sodium bisulfite solution. The reaction mixture was diluted with MDC, the layers separated and the organic layer washed sequentially with water, saturated sodium bicarbc)nate, and saturated ammonium chloride, dried over sodium sulfate and evaporated to dryness in vacuo and the residue chromatographed on silica gel with 10:1 MDC:diethyl ether. There was thus obtained $\overline{0.1 \text{ g}}$ (10%) of 2-(2,6-dichlorophenylsulfonylmethyl)saccharin, mp 201.0 – 203.0 °C.

Example 32B

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Following a procedure similar to that described in Example 32A, 0.75 g (0.0023 mol) 2-(2-pyrimidinylsulfinyl-methyl)saccharin was oxidized with 0.4 g (0.0023 mol) of 3-chloroperbenzoic acid in 50 ml MDC and the product recrystallized from 75:25 acetonitrile:ethanol to give 2-(2-pyrimidinylsulfonyl-methyl)saccharin, mp 225-227°C.

Example 33A

To a solution of 0.345 g (0.001 mol) 2-(5-mercapto-1,3,4-thiadiazol-2-ylthiomethyl)saccharin and 0.46 ml (0.003 mol) triethylamine in 2 ml DMF was added 0.37 g (0.002 mol) 4-(2-chloroethyl)morpholine hydrochloride. The reaction mixture was stirred at ambient temperature for about 24 hours, quenched by pouring into water and extracted with ethyl acetate. The organic layer was washed with water, then with brine and taken to dryness to give a yellow oil which was taken into chloroform and chromatographed on silica gel, eluting with ethyl acetate. There was thus obtained 0.225 g (49%) of 2-(5-2-(4-mor-pholinyl))ethylthio -1,3,4 -thiadiazol -2 -ylthiomethyl)saccharin, mp 129 -131 °C.

Example 33B

Following a procedure similar to that described in Example 33A above, 1.72 g (0.005 mol) 2-(5-mercapto-1,3,4-thiadiazol-2-ylthiomethyl)saccharin was reacted with 1.44 g (0.01 mol) 2-dimethylaminoethyl chloride hydrochloride in 10 ml DMF in the presence of 1.72 g (0.017 mol) triethylamine to give 1.2 g (58%) of 2-(5-[2-(N,N-dimethylamino)ethylthio]-1,3,4-thiadiazol-2-ylthiomethyl)-saccharin, mp 90.5-91.5°C (from ethyl acetate).

Example 33C

Following a procedure similar to that described above in Example 33A, 0.69 g (0.002 mol) 2(5-mercapto-1,3,4-thiadiazol-2-ylthiomethyl)saccharin was reacted with 0.74 g (0.004 mol) 1-(2-chloroethyl)-piperidine hydrochloride in 4 ml DMF in the presence of 0.4 g triethylamine to give 0.55 g (60%) of 2-(5-[2-(1-piperidinyl)ethylthio]-1,3,4-thiadiazol-2-ylthiomethyl)saccharin, mp $100.0-101.0^{\circ}$ C (from ethyl acetate).

Example 33D

Following a procedure similar to that described in Example 33A, 2-(5-mercapto-1,3,4-thiadiazol-2-ylthiomethyl)-saccharin is reacted with <math>2-chloroethyldiethylamine hydrochloride in DMF in the presence of triethylamine to give <math>2-(5-2-(diethylamino)ethylthio-]1,3,4-thiadiazol-2-ythiomethyl)-saccharin, mp 93.0-94.5°C (from cyclohexane:ethyl acetate).

Example 33E

Following a procedure similar to that described in Example 33A, 1.72 g (0.005 mol) 2-(5-mercapto-1.3.4-thiadiazol-2-ylthiomethyl)saccharin was reacted with 1.84 g (0.01 mol) 1-(2-chloroethyl)-piperidine hydrochloride in 10 ml DMF in the presence of 2.4 ml triethylamine to give 1.08 g (47%) of 2-(5-[2-(1-piperdinyl)ethythio]-2-thioxo-1.3.4-thiadiazolin-3-ylmethyl)saccharin, mp $89.0-92.0^{\circ}$ C, in which the 1.3.4-thiadiazol-2-ylthiomethyl group underwent rearrangement during the reaction.

Example 34

To a solution of 0.44 g (0.0013 mol) 2-[1-(2-hydroxyethyl)-1H-tetrazol-5-ylthiomethyl]saccharin in 10 ml acetone was added chromic acid prepared from dilute sulfuric acid and sodium dichromate dihydrate (Jones' reagent) while maintaining the reaction mixture at 0°C until a persistent orange-brown color remained in the solution. The mixture was stirred for 1 hour at 0°C, then at ambient temperature for about 6 hours, diluted with water and extracted with ethyl acetate. The combined organic extracts were washed with water, then with brine, dried and evaporated to dryness, and the residue chromatographed on silica gel, eluting with <math>5-15% methanol-MDC, to give 0.29 g (63%) of 2-[1-(2-carboxymethyl)-1H-tetrazol-5-ylthiomethyl]saccharin,mp 173-175°C.

Example 35

A solution of 1 g (0.0026 mol) 2-[1-(3-aminophenyl)-1H-tetrazol-5-yl-thiomethyl]saccharin and 0.26 g (0.0026 mol) succinic anhydride in 50 ml dioxane was stirred at ambient temperature for two hours, heated under reflux for six hours, stirred at ambient temperature for two days, then heated to reflux again and additional small amounts of succinic anhydride added over a period of a few hours until TLC analysis indicated the absence of starting material in the reaction mixture. The reaction was cooled, poured into dilute hydrochloric acid and ice water, and the solid which separated was collected, dried and recrystallized from 50:50 ethanol: acetonitrile to give 0.5 g (39%) of 2-[1-(3-succinoylaminophenyl)-1H-tetrazol-5-ylthiomethyl]saccharin, mp 197-199°C.

Example 36

5-Nitro-2-(1-phenyl-1H-tetrazol-5-ylthiomethyl)-saccharin (1 g, 0.0024 mol) was reduced over 3 spatulas of Raney nickel (washed with THF prior to use) in 150 ml THF under 50 psi hydrogen pressure. When reduction s complete (in about 5 hours) the reaction mixture was filtered, the filter pad washed with THF, and the filtrate evaporated to dryness to give a pale yellow, cloudy oil which was taken into hot ethanol and filtered. On cooling, the product separated and was collected to give 0.4 g (43%) of 5-amino-2-(1-phenyl-1H-tetrazol-5-ylthiomethyl)saccharin as yellow crystals.

Example 37

A suspension of 0.4 g 5-amino -2-(1-phenyl-1H-tetrazol-5-yl-thiomethyl)saccharin in 25 ml acetonitrile was treated with 0.08 g (0.001 mol) acetyl chloride, the mixture was heated under reflux for 30 minutes, treated with an additional drop of acetyl chloride and refluxing continued for another 30 minutes. Evaporation of the mixture to dryness afforded a white foam which was chromatographed on silica gel with 95:5 MDC:ethyl acetate to give 200 mg (43%) of $\frac{5-acetylamino-2-(1-phenyl-1H-tetrazol-5-ylthiomethyl)-saccharin.}{5-acetylamino-2-(1-phenyl-1H-tetrazol-5-ylthiomethyl)-saccharin.}$

Example 38

A solution of 5 g (0.025 mol) of the sodium salt of 1-phenyl-5-mercaptotetrazole in 50 ml methylethylketone was added dropwise with stirring to a warm solution of 3-chloro-1-iodopropane in 300 ml MEK. The mixture was stirred at 40 °C for six hours, allowed to stand at ambient temperature for about two days and the reaction mixture taken to dryness in vacuo. The residue was dissolved in MDC, the solution washed with water and the aqueous washings back-extracted with MDC. The combined organic extracts were dried and taken to dryness to give a yellow oil which was chromatographed on silica gel, eluting with MDC. There was thus obtained 3.6 g (57%) of 1-phenyl-5-(3-chloropropylthio)-1H-tetrazole as a pale yellow oil. (In another run, the same material was obtained as a white crystalline solid, mp 33-34 °C.)

The product (3.5 g, 0.014 mol) was dissolved in 100 ml MEK, 1.7 g (0.014 mol) of the sodium salt of thiophenol was added, the mixture warmed at 40° C for three hours, poured into potassium bicarbonate solution and the mixture extracted with MDC. The combined organic extracts, on drying and evaporation to dryness, afforded a pale yellow oil which was chromatographed on silica gel with MDC to give 3.85 g (86%) of 1-phenyl-5-[3-(phenylthio)propylthio]-1H-tetrazole. (In another run this same material was obtained as white crystals,mp $57-59^{\circ}$ C.)

A solution of 3.8 g (0.012 mol) of the product in 100 ml carbon tetrachloride was treated with 1.5 g (0.012 mol) N-chlorosuccinimide, the mixture allowed to stand at ambient temperature for one hour, then filtered and the filtrate evaporated to dryness to give 4.3 g of $\frac{1-\text{phenyl}-5-[3-\text{chlor}-3-(\text{phenylthio})-\text{propylthio}]-1H-\text{tetrazole}$.

The product (0.012 mol) and 4.58 g (0.12 mol) of the thallium salt of saccharin dissolved in 75 ml DMF was heated at 50 °C for three hours, allowed to stand at ambient temperature for two hours, filtered and the filter pad washed with DMF. The combined filtrate was poured into water, the mixture extracted with MDC, and the combined organic extracts were washed with brine and concentrated to dryness to give a pale yellow oil which was chromatographed on silica gel with MDC to give 2.45 g (41%) of 2-[1-phenylthio-3-(1-phenyl-1H-tetrazol-5-ylthio)propyl]saccharin.

The product (1 g, 0.002 mol) was dissolved in MDC and oxidized with 0.34 g (0.002 mol) 3 – chloroperbenzoic acid according to the procedure described in Example 32A above. There was thus obtained 0.8 g (76%) of 2 – [1 – phenyl – sulfinyl – 3 – (1 – phenyl – 1H – tetrazol – 5 – ylthio)propyl]saccharin.

The product (1.6 g, 0.003 mol) was heated in 130 ml diethylene glycol dimethyl ether at 120°C for 45 minutes and the mixture cooled and poured into water. The solid which separated was collected, dried and dissolved in MDC and the solution chromatographed on silica gel in MDC. There was thus obtained 1.2 g of trans – 2 – [3 – (1 – phenyl – 1H – tetrazol – 5 – ylthio) – 1 – propenyl]saccharin, mp 191 – 193°C.

Example 39

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Other 2 – unsubstituted saccharins of formula II useful as intermediates for the preparation of the compounds of formula I can be prepared as follows.

Reaction of 3-trifluoromethylbenzoic acid with thionyl chloride affords 3-trifluoromethylbenzoyl chloride, which, on reaction with diethylamine, affords 3-trifluoromethyl-N,N-diethylbenzamide. Following a procedure similar to that described in Example 22A, reaction of the latter with s-butyl lithium and reaction of the resulting lithium salt with sulfur dioxide followed by sodium hydroxylaminesulfonate affords 3-trifluoromethyl-2-aminosulfonyl-N,N-diethylbenzamide, which, on heating in glacial acetic acid, affords 7-trifluoromethyl saccharin.

Similarly, reaction of 4-cyclohexylbenzoic acid with thionyl chloride affords 4-cyclohexylbenzoyl chloride, which, on reaction with diethylamine, affords 4-cyclohexyl-N,N-diethylbenzamide. Following a procedure similar to that described in Example 22A, reaction of the latter with s-butyl lithium and reaction of the resulting lithium salt with sulfur dioxide followed by sodium hydroxylaminesulfonate affords 4-cyclohexyl-2-amino-sulfonyl-N,N-diethylbenzamide, which, on heating in glacial acetic acid, affords 6-cyclohexylsaccharin.

Reaction of 6 – aminosaccharin with methanesulfonyl chloride or trifluoromethylsulfonyl chloride in MDC in the presence of pyridine affords, respectively, 6 – methylsulfonylaminosaccharin or 6 – trifluoromethylsul – fonylaminosaccharin.

Diazotization of 6-aminosaccharin with nitrous acid in an acid medium and decomposition of the resulting diazonium salt in the presence of cupric cyanide or cupric chloride and sulfur dioxide, or cupric chloride and an alkali metal salt of methyl mercaptan or trifluoromethyl mercaptan affords, respectively, 6-cyansaccharin, 6-chlorosulfonylsaccharin, 6-methylthiosaccharin or 6-trifluoromethylthiosaccharin. Reaction of the 6-chlorosulfonylsaccharin in situ with ammonia or methanesulfonylamide affords, respectively, 6-aminosulfonylsaccharin and 6-methanesulfonylaminosulfonyl-saccharin. Oxidation of 6-methyl-thiosaccharin and 6-tri-fluoromethylthiosaccharin with two molar equivalents of 3-chloroperbenzoic acid affords 6-methanesulfonylsaccharin and 6-trifluoromethylsulfonylsaccharin, respectively.

Hydrolysis of 6-cyanosaccharin by heating with aqueous sodium hydroxide affords saccharin-6-carboxylic acid. Reaction of 6-cyanosaccharin by heating with a catalytic amount of sulfuric acid in ethanol solution affords ethyl saccharin-6-carboxylate, which, on reduction with lithium borohydride, affords 6-hydroxymethylsaccharin. Oxidation of the latter with pyridine:chromium trioxide (2:1) complex (Collins reagent) in MDC affords 6-formylsaccharin, which, on reductive annination with ammonia and sodium cyanoborohydride, affords 6-aminomethylsaccharin.

Reaction of each of the 2-unsubstituted saccharins so-prepared with chloromethyl phenyl sulfide in the presence of potassium t-butoxide and tetrabutylammonium bromide, and reaction of the resulting 2-phenylthiomethylsaccharins with sulfuryl chloride in MDC affords the R_4 -2-unsubstituted saccharins of formula I listed in TABLE D where, in each instance, m and n are 0, R_1 is CI and R_2 and R_3 are both hydrogen.

Example

39A

39B 39C

39D

39E

39F

39G 39H

391

39J

39K

39L

39M

TABLE D

6-CN

6 - NH2 SO2

6 - CH₃SO₂

6-CF3SO2

6 - HOOC 6 - HOCH₂

6-OHC

6 - NH2 CH2

6 - CH₃SO₂NHSO₂

6 - cyclohexyl

6 - CH₃SO₂NH6 - CF₃SO₂NH

R₄
7 – CF₃

)	ł	

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Example 40

Following a procedure similar to that described in Example 4 above, substituting for the 2-chloromethyl-4-chlorosaccharin and the sodium 1-phenyltetrazole salt used therein molar equivalent amounts of a respective appropriate 2-chloromethyl- R_4 -saccharin described in TABLE D above and an appropriate L_nR_1 moiety, the following compounds listed in TABLE E, where R_3 in each instance, unless noted otherwise, is hydrogen, are prepared.

TABLE E

n		n	
J	ι		
_	7		

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Example	n	L	R ₁	R ₄
40A	1	S	1 - C ₆ H ₅ - 5 - tet.	7 – CF ₃
40B	1	S	$1 - C_6 H_5 - 2 - (1,3,4 - triaz.)$	6 – cyclohexyl
40C	1	S	$1 - (4 - Mor.CH_2CH_2) - 5 - tet.$	6 – CH₃SO₂NH
40D	1	S	3 - (pyridyl) - 5 - tet.	6 − CF₃SO₂NH
40E	0	-	4 - (3 - pyridyl) - 5 - thioxo - 1 - tet.	6 – CN
40F	0	-	2 - Me - 5 - thioxo - 1 - tet.	6 - NH ₂ SO ₂
40G	1	S	2 - Me - 5 - tet.	6 - CH₃SO₂NHSO₂
40H	1	S	5 - Me - 2 - (1,3,4 - thiadiaz.)	6 - CH ₃ SO ₂
401	1	S	1 - Me - 2 - (1,3,4 - triaz.)	6 ~ CF ₃ SO ₂
40J	1	S	$4 - C_6 H_5 - 2 - (1,3,5 - thiadiaz.)$	6-HOOC
40K	1	S	5 - HS - 2 - (1,3,4 - thiadiaz.)	6 − HOCH ₂
40L	0	1	$4 - C_6 H_5 - 5 - \text{thioxo} - 1 - \text{tet}.$	6 - OHC
40M	1	S	$5 - CH_3 - 2 - (1,3,4 - thiadiaz.)$	6 - NH ₂ CH ₂
40N	1	S	1 − C ₆ H ₅ − 5 − tet	5 – CH ₃ O ^(a)

In compound 40N, (a) R_3 is isopropyl. This is prepared by reaction of N,N-diethylcarbamyl chloride with the lithium salt of 2-bromo-methoxyisopropylbenzene; reaction of the resulting (79%) N,N-diethyl-2-isopropyl-4-methoxybenzamide with s-butyl lithium followed by sulfur dioxide and sodium hydrox-ylamine sulfonate; heating the resulting (56%) N,N-diethyl-2-aminosulfonyl-4-methoxy-6-isopropylbenzamide in glacial acetic acid; reaction of the resulting (100%) diethylammonium salt of 4-isopropyl-6-methoxysaccharin with chloromethyl phenyl sulfide; reaction of the resulting (88%) 2-phenylthiomethyl-4-isopropyl-6-methoxysaccharin with sulfuryl chloride, and reaction of the resulting (88%) 2-chloromethyl-4-isopropyl-6-methoxysaccharin with the sodium salt of 1-phenyl-5-mer-captotetrazole.

BIOLOGICAL TEST RESULTS

Measurement of the inhibition constant, K_i, of a HLE-inhibitor complex has been described for truly reversible inhibition constants usually concerning competitive inhibitors. [Cha, Biochem. Pharmacol., 24, 2177-2185 (1975)]. The compounds of the present invention, however, do not form truly reversible inhibitor complexes but are consumed by the enzyme to some extent. Thus, instead of measuring a K_i, a K_i* is calculated which is defined as the ratio of the k_{off}/k_{on}, the rate of reactivation of the enzyme to the rate of inactivation of the enzyme. The values of k_{off} and k_{on} are measured and K_i* is then calculated.

The rate of inactivation, k_{on} , of enzymatic activity was determined for the compounds tested by measuring the enzyme activity of an aliquot of the respective enzyme as a function of time after addition of the test compound. By plotting the log of the enzyme activity against time, an observed rate of inactivation, k_{obs} , is obtained which can be represented as $k_{obs} = 1n2 / t_{1/2}$ where $t_{1/2}$ is the time required for the enzyme activity to drop by 50%. The rate of inactivation is then equal to

15 $k_{on} = k_{off}/[1]$

where [I] is the concentration of the inhibiting compound.

The reactivation constant, k_{off} , is similarly determined and the inhibition constant, K_i^* , is then calculated as

 $K_i^* = k_{off} / k_{on}$

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The values obtained for k_{on} and K_i^* for specific substituted saccharin derivatives are shown in TABLE F, the compounds being identified by the example numbers above where their preparations are described.

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TABLE F

5			Elastase	a-Chymotr	vpsin
		$10^{-3} \text{ x k}_{\text{on}}$	K _i *	$10^{-3} \times k_{on}$	K _i *
	Example	$(M^{-1}\underline{sec}^{-1})$	_(nM)	$(M^{-1} \operatorname{sec}^{-1})$	(nM)
10	1	0.63	102	1.2	917
	2	4.9	45	2.9	51
	3	45 0	0.5	5.8	26
	4	20	12		
15	5	44	6	6.0	25
	6	5.5	15	3.7	300
	7	5.2	15		
20	8	1.0	81	. 2.1	523
	9	2.5	32		
	10	7.0	11	7.0	157
	11	4.6			
25	12	0.97	82	1.1	1000
	13	0.3	285	2.6	423
	14	0.6	138	2.8	392
30	15	3.2	69		
	16		2,600		
	17	2.9	270		
35	18	1.0	100	0.21	620
	19		8,500		
	21	950	0.5		
	22A	12.4	6.0		
40	22B	2.4	90		
	22C	105	0.4		
	22D	75.8	1.3		
45	22E	4.7	91		
	22F	20	6.5		
	22G	3.0	50		
50	22H	25	1.2		
-	23A	30	9		

TABLEF (cont'd)

_			Elastase	α -Chymotry	
5		10-3 x k _{on}	K _i *	$10^{-3} \times k_{on}$	K _i *
	Example	$(M^{-1}_{sec}^{-1})$	_(nM)	$(M^{-1} \operatorname{\underline{sec}}^{-1})$	(nM)
	23B	50	4.4		
10	27	7.5	13		
	28	50.7	2.5		
	29	1.1	120		
15	3 0A		10,500		
13	30C	2.5	17		
	30D	0.8	85		
	30E	7.3	12.7		
20	30F		>>85,000		
	3 0G	9.3	7.7		
	30H	3.2			
25	3 0I	0.45	650		
25	3 0J	14.0	10		
	30L	6.5	7.6		
	30M	1.0	10		
30	30N	0.25	220		
	3 0-O	0.03	2,000		
	3 0Q	1.38	70.4		
05	30R	2.3	36		
35	30S	1.81	40.4		
	3 0 T	1.1	14.5		
	3 0U	5.6	13.6		
40	30V	24	15		
	30W	1.5	80		
	30X	0.55	143		
	30Z	7 0	0.5		
45	30AB	2.8	23		•
	30AC	0.19	246		
	30AD	7.0	11.6		

TABLE F (cont'd)

-			Elastase	<u>α-Chymotrypsin</u>		
5		$\frac{10^{-3} \times k_{on}}{}$	K _i *	$10^{-3} \times k_{on}$	K _i *	
	Example	(M ⁻¹ _sec ⁻¹)	_(nM)	$(M^{-1}_{sec}^{-1})$	(nM)	
	30AE	63.2	2.0			
10	30AF	4.2	12.5			
	30AG		>1,000			
	30AH	1.1	60			
15	30AI	53.8	2			
	30AJ	1.8	15.5			
	30AK	94	0.3			
	30AL	100	0.7			
20	30AM	12.3	4.5			
	30AN	19	30			
	30AP	1.3	51			
25	30AQ	0.9	260			
	30AR	0.3	153			
	30AS	1.7	40			
	30AT	45.3	3			
30	30AU	8.9	17.5			
	30AV	7.6	15			
	30AW	30	1			
35	30AX	7.5	16			
55	30AY	49	0.6			
	30BA	42	0.8			
	30BB		>500			
40	30BC	5.8	80			
	30BD	11.7	8			
	30BE	7.4	250			
45	30BF	0.02	40,000			
45	30BG	1.7	5 0			
	30BH		>10,000			
	30BI	39	0.5			

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TABLE F (cont'd)

		1	Elastase		ypsin
5		$10^{-3} \times k_{on}$	K _i *	$10^{-3} \times k_{on}$	K _i *
	Example	$(M^{-1} \operatorname{sec}^{-1})$	_(nM)	$(M^{-1} \operatorname{sec}^{-1})$	(nM)
	31C	3.4	300		
10	32A	0.6	145		
	33A	3.6	12		
	33B	4.0	18		•
	33 C	0.2	35 0		
15	34	0.4	256		
	3 5	3.6	22		
	37		>90,000	•	
20	38	43	2		

Examples 41 A-L

Further examples of compounds of formula I wherein m is 0 and R_2 , R_3 and R_4 are each hydrogen were prepared from 2-(chloromethyl)saccharin (or, if noted, 2-(bromomethyl)-saccharin) and in each example the corresponding R_1 - L_n -H except as noted and are described in TABLE G.

TABLE G

5			M.p.(°C.) Yi	eld(%)
J	Example	R ₁ -L _n	_ From	Method
	41A	{5-[2-(1-piperidinyl)ethylthio]-	100-101	60
		1,3,4-thiadiazol-2-yl}thio	EtOAc	Α
10	41B	{5-[2-(diethylamino)ethyl]-1,3,4-	93-94.5	44
		thiadiazol-2-yl}thio	C ₆ H ₁₂ /EtOAc	В
	41C	{5-[2-(4-morpholinyl)ethylamino]-	159-161	57
15		1,3,4-thiadiazol-2-yl}thio	EtOAc	В
	41D	4,5-dicyanoimidazol-1-yl	143-146	5 0
			CHCl ₃	D
20	41E	trifluoromethylsulfonyl	134-137	25
20				E
	41F	2-formyl-4-nitrophenoxy	168-171	
			EtOH/MeCN	I or J
25	41G	2-hydroxymethyl-4-nitrophenoxy	150-152	92
				G
	41H	2-chloromethyl-4-nitrophenoxy	148-(slow)	30.6
30		•		Н
	411	(5,7-dichloroquinolin-8-yl)oxy	220.5-222.5	46
			EtOH/MeCN	I
35	41J	4-chloromethyl-2-nitrophenoxy	131-133	32
33	,	•	EtOH	J
	41K	4-(4-nitrophenylazo)phenoxy	217-219	63
	,			J
40	41L	2,4-dichloro-3-(4-methyl-1-	>190*	32
	112	piperazinylcarbonyl)phenoxy	Et ₂ O	L
		L-L		

Notes to TABLE G

	A:	By alkylation with 1-(2-chloroethyl)piperidine in dimethyl-		
	formamic	le at room temperature of 2-[(5-mercapto-1,3,4-thiadiazol-2-yl)thio-		
5	methyl]sa	accharin, prepared in turn by condensation of 2-(bromomethyl)saccharin		
	and 1,3,4	-thiadiazol-2,5-dithiol with sodium methoxide in ethanol at reflux in		
	38% yiel	d (m.p. 204-206.5°C. from 1,2-dichloroethane, 38% yield).		
10	B:	From 2-(bromomethyl)saccharin with ethanol as solvent and		
	sodium methoxide as base.			
	D:	From 2-(bromomethyl)saccharin and tetrahydrofuran as solvent.		
15	E:	By oxidation of the corresponding sulfide.		
	G:	By reduction of the compound of Example 41F.		
	H:	By displacement of hydroxy in the compound of Example 41G by		
<u></u>	chloro.			
20	I:	With acetonitrile as solvent and methyltriazabicyclodecene as base.		
	J : -	With dimethylformamide as solvent and thallium ethoxide as base.		
	L:	With methyl ethyl ketone as solvent and potassium carbonate as		
25	base.			
	*Hydroc	hloride salt.		

30 Examples 42A - BU

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Further examples of compounds of formula I wherein m is 0 except Example 42B wherein m is 1, R_2 and R_4 are each hydrogen and R_3 is isopropyl were prepared from 2-chloromethyl-4-isopropylsaccharin (Example 22E above) (or, if noted, 2-bromomethyl-4-isopropylsaccharin) and in each example the corresponding R_1 – L_n – H except as noted and are described in TABLE H.

TABLE H

			M.p.(°C.)	Yield (%)
	Example	R_1 - L_{n-}	From	Method_
5	42A	(5-cyclohexylamino-1,3,4-	202-205	51
		thiadiazol-2-yl)thio	EtOH	Α
	42B	(2-phenyltetrazol-5-yl)thio	95-97	39.5
10		$(\mathbf{m} = 1)$	EtOH	В
	42C	methylsulfonyl	148.5-150.5	40.3
			EIOH	С
15	42D	1,1,3-trioxotetrahydro-1,2,5-	Foam	53
15		thiadiazol-2-yl		D
	42E	(2-methyl-4-pyron-3-yl)oxy	136.5-138	48
•			EtOAc/C ₆ H ₆	Е
20	42F	4-[(4-methylpiperazin-1-yl)-	160-165*	.60
		carbonyl]phenoxy	Et ₂ O	F .
	42G	(6-hydroxymethyl-4-pyron-3-yl)oxy	144-145	38
25			EtOAc/C ₆ H ₁₄	Α
	42H	{5-[2-(4-morpholinyl)ethylthio]-	53-93*	87
		1,3,4-thiadiazol-2-yl}thio	iPrOH/Et ₂ O	Н
	42I	2,4-dichloro-6-(4-morpholinyl-	153-155	53
30		sulfonyl)phenoxy	EtOH	I
	42J	2-chloro-4-(4-morpholinyl-	194-196	54
		sulfonyl)phenoxy	EiOH	J
35	42K	{5-[2-(1-piperidinyl)ethylthio}-	oil	50%
		1,3,4-thiadiazol-2-yl}thio		K
	42L	{5-[2-(diethylamino)ethyl]-1,3,4-	oil	38
40		thiadiazol-2-yl}thio		L
,,,	42M	{5-[2-(dimethylamino)ethylthio]-	124-125.5*	62
		1,3,4-thiadiazol-2-yl}thio	i PrOH	M
	42N	{5-[2-(4-morpholinyl)ethyl]-	142-143*	71%
45		1,3,4-thiadiazol-2-yl}thio	i Pr OH	L
	420	{5-[2-(1-piperidinyl)ethyl]-	149-150*	61
		1,3,4-thiadiazol-2-yl}thio	EtOH	0

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_TABLE H _(continued)

		(continued)		
			M.p.(°C.)	Yield (%)
5	Example	R_{\perp} - L_n	From	Method
	42P	(4,5-dichloropyridazin-3-yl)oxy	183-186	40
10			Et ₂ O	E
	42Q	4,5-di(methoxycarbonyl)-1,2,3-	153-154	30
	•	triazin-1-yl	EtOAc/C ₆ H ₁₄	Q
	42R	2-methoxycarbonyl-5-methoxy-	104-105	30
15		phenoxy		J
	42S	2-fluoro-4-(4-morpholinyl-	165-167	68.7
		sulfonyl)phenoxy	EtOH	J
20	42T	2-chloro-4-(1,1-dioxo-4-thia-	Not sharp	29.6
		morpholinylsulfonyl)phenoxy	EtOH	T
	42U	5-phenylsulfonyl-1,2,3-triazol-	163.5-168	32
25		1-yl	EtOAc	Q
20	42V	4-trimethylsilyl-5-phenylsulfonyl-	200.5-201.5	22
		1,2,3-triazol-1-yl	EtOAc	Q
	42W	4-methoxycarbonyl-1,2,3-triazol-	192-193	66
30		1-yl	EtOAc	Q
	42X	(2-ethyl-4-pyron-3-yl)oxy	93-98	85
				X
35	42Y	5-methoxycarbonyl-1,2,3-triazol-	150.5-151.5	23
		1-yl	EtOAc	Q
	42Z	4-ethoxycarbonyl-5-phenyl-1,2,3-	177.5-179	37
	•	triazol-1-yl	EtOAc	Q
40	42AA	4-phenyl-5-ethoxycarbonyl-1,2,3-	117-119	56
		triazol-1-yl	EtOAc/C ₆ H ₁	
	42AB	2,6-difluoro-4-(4-morpholinyl-	185-187	40
45		sulfonyl)phenoxy	EtOH	J
	42AC	4,6-difluoro-2-(4-morpholinyl-	176-178	38.6
		sulfonyl)phenoxy	EtOH	J
	42AD	4,5-difluoro-2-(4-morpholinyl-	140.5-142.5	58.6
50		sulfonyl)phenoxy	EtOH	J

TABLE H (continued)

			M.p.(°C.)	Yield (%)
5	Example	R ₁ -L _n	From	Method
	42AE	(3-phenylcoumarin-7-yl)oxy	145-147	63
		·	EtOAc/C ₆ H ₁₂	E
10	42AF	(4-phenylcoumarin-7-yl)oxy	219-221	53
			MeCN	E
	42AG	4-fluoro-2-(4-morpholinyl-	169-171	27.4
15		sulfonyl)phenoxy	EtOH	J
	42AH	2,5-difluoro-4-(4-morpholinyl-	158-160	44
		sulfonyl)phenoxy	EtOH	J
	42AI	3-[2-(4-morpholinyl)ethylamino-	Foam	10
20		carbonyl]phenoxy		F
	42AJ	pentafluorophenoxy	90-92	10
			C ₆ H ₁₄	F
25	42AK	2,4-dichloro-3-[2-(4-morpholinyl)-	Foam*	30
		ethoxycarbonyl]phenoxy		F
	42AL	(5-phenyl-1,3,4-oxadiazol-2-yl)-	132-134.5	60
		thio	EtOH	AL
30	42AM	4-carboxy-1,2,3-triazol-1-yl	182.5-183.5	21
			MeCN	Q
	42AN	4-phenyl-5-(4-methylphenyl-	173-174.5	32
35		sulfonyl)-1,2,3-triazol-1-yl	EtOAc	Q
	42AO	4-(4-methylphenylsulfonyl)-5-	179-181.5	40
		phenyl-1,2,3-triazol-1-yl	EtOAc	Q
40	42AP	(6-chloro-4-trifluoromethyl-	169-170	29
		coumarin-7-yl)oxy	EtOH	X
	42AQ	(4-methylcoumarin-7-yl)oxy	178-179.5	18
			iPrOH	E
45	42AR	3-[(4-methylpiperazin-1-yl)-	>150*	34
		sulfonyl]phenoxy	Et ₂ O	F
	42AS	3-[2-(4-morpholinyl)ethoxy]-	102-105*	35
50		phenoxy	Et ₂ O	F

TABLE H (continued)

		\		
5			$M.p.(^{o}C.)$	(ield (%)
	Example	R_1 - L_n -	From 1	<u>dethod</u>
	42AT	3-{2-[(dimethylamino)ethyl]methyl-	115-118*	60
		aminosulfonyl}phenoxy	Et ₂ O	F
10	42AU	[3-(benzothiazol-2-yl)coumarin-7-	239-240	30
		yl]oxy	MeCN	E
	42AV	(saccharin-6-yl)oxy	90-110	11
15				ΑV
	42AW	4-phenylsulfonyl-1,2,3-triazin-	219.5-220.5	87
		1-yl	EtOAc	Q
	42AX	ethoxythiocarbonylthio	oil	57
20				AX
	42AY	2-fluoro-4-(4-morpholinyl-	149-151	83.7
		sulfonyl)phenylthio	EtOH	J
25	42AZ	4-ethylsulfonyl-5-isopropyl-	137-139	20
		1,2,3-triazin-1-yl	EtOAc/C ₆ H ₁₂	Q
	42BA	4-isopropyl-5-ethylsulfonyl-	150-152	6-11
		1,2,3-triazin-1-yl	EtOAc/C ₆ H ₁₂	Q
30	42BB	2-fluoro-4-(4-morpholinyl-	198-201	42.5
		sulfonyl)phenylsulfinyl		ВВ
	42BC	2-fluoro-4-(4-morpholinyl-	219-221	87.5
35		sulfonyl)phenylsulfonyl	EtOH	BC
	42BD	4,5-di(aminocarbonyl)-1,2,3-	261-262	68
		triazin-1-yl	MeCN	Q
	42BE	4,5-dicarboxy-1,2,3-triazin-1-yl	211-215	15
40		(monosodium salt)	THF	Q
	42BF	4,5-dicarboxy-1,2,3-triazin-1-yl		Q,BF
	42BG	[4-(4-morpholinyl)-1,2,5-	132-134	5 0
45		thiadiazol-3-yl]oxy	EtOH	J
	42BH	4-methylsulfonylphenoxy	147-150	79
				J
50	42BI	[4-(ethoxycarbonylmethyl)thiazol-	96-97	20
50		2-yl]thio	CCl ₄ /C ₆ H ₁₂	BI

TABLE H (continued)

5			M.p.(°C.)	Yield (%)
,	Example	$R_1=L_n$	From	Method
	42BJ	4-trimethylsilyl-5-dimethylamino-	196-198	91
10		sulfonyl-1,2,3-triazol-1-yl	MeCN	Q
	42BK	4-(1,1-dimethyl)ethyl-5-dimethyl-	207-209	7
		aminosulfonyl-1,2,3-triazol-1-yl	EtOAc/C ₆ H ₁₂	Q
15	42BL	4-dimethylaminosulfonyl-5-(1,1-	201-202	3
		dimethyl)ethyl-1,2,3-triazol-1-yl	EtOAc/C ₆ H ₁₂	Q
	42BM	4-trimethylsilyl-1,2,3-triazol-	oil	81
		1-yl		Q
20	42BN	2,6-dichlorophenoxy	146-148	70
			Et_2O/C_6H_{14}	F
	42BO	2,4,6-trichlorophenoxy	140-142	28
25			t-BuOMe	ВО
	42BP	2,4-dichloro-3-(4-methyl-1-	>175*	75
		piperazinylcarbonyl)phenoxy	Et ₂ O	F
	42BQ	2,4-dichloro-3-carboxyphenoxy	194-196	57
30			C_6H_{14}/CH_2C_1	₂ BQ
	42BR	3-[2-(4-morpholinyl)ethoxy-	>115*	51
		carbonyl]phenoxy	Et ₂ O	F
35	42BS	2,4-dichloro-3-[2-(4-morpholinyl)-	149-152*	30
		ethylaminocarbonyl]phenoxy	Et ₂ O	F
	42BT	4-(4-morpholinylsulfonyl)phenoxy	glass	
40				BT
	42BU	4-trimethylsilyl-5-methoxy-	122.5-123.5	68
		carbonyl-1,2,3-triazol-1-yl	EtOAc/C6H1	2 Q

Notes to TABLE H

	A:	With dimethylformamide as solvent and triethylamine as base.
	B:	In three steps by, first, condensation of the thallium salt of
5	4-isoprop	ylsaccharin formed from 4-isopropylsaccharin and thallium ethoxide with
	1-phenyl-	5-(3-chloro-3-phenylthio-propyl)tetrazole in dimethylformamide in
	77% yield	l, second, oxidation of the resulting 2-[3-(1-phenyltetrazol-5-yl)-
10	1-(phenyl	thio)propyl]-4-isopropylsaccharin with m-chloroperbenzoic acid in
	87% yield	and, third, thermal elimination of the resulting 2-[3-(1-phenyltetrazol-
	5-yl)-1-(pl	henylthio)propyl]-4-isopropylsaccharin in 59% yield.
15	C:	In two steps, first, condensation of 2-(chloromethyl)-4-isopropyl-
15	saccharin	with sodium methylthiolate and, second, oxidation of the resulting
	2-(methylt	thio)-4-isopropylsaccharin.
	D:	With toluene as solvent, 1,1,3-trioxotetrahydro-1,2,5-thiadiazole
20	and tetrabi	utylammonium bromide.
	E:	With dimethylformamide as solvent and sodium hydride as base.
	F:	With methyl ethyl ketone as solvent and potassium carbonate as
25	base. *Hy	drochloride salt.
	H: In	two steps, first, condensation of 2-(chloromethyl)-4-isopropylsaccharin
	with 1,3,4-	thiadiazol-2,5-dithiol monopotassium salt in methanol in 83% yield
30	and, secon	d, condensation of the resulting 2-[(5-mercapto-1,3,4-thiadiazol-2-yl)-
	thiomethyl]-4-isopropylsaccharin with 4-(2-chloroethyl)morpholine hydrochloride
	in dimethy	lformamide as solvent and triethylamine as base in 87% yield.
	*Hydrochl	oride salt.
35	I:	With dichloromethane as solvent and diazabicycloundecene as base.
	J:	With acetonitrile as solvent and methyltriazabicyclodecene as base.
	K:	By alkylation of 2-[(5-mercapto-1,3,4-thiadiazol-2-yl)thiomethyl]-
40	4-isopropy	lsaccharin (Example 42H) with 1-(2-chloroethyl)piperidine
	hydrochlori	ide in dimethylformamide as solvent and triethylamine as base.
	L:	With ethanol as solvent and sodium methoxide as base.
45	M:	By alkylation of 2-[(5-mercapto-1,3,4-thiadiazol-2-yl)thiomethyl]-
	4-isopropyl	saccharin (Example 42H) with (2-chloroethyl)dimethylamine hydrochloride
	in dimethyl	formamide as solvent and triethylamine as base. *Maleate salt.

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	N:	With ethanol as solvent and sodium methoxide as base.*Maleate salt.			
		With ethanol as solvent and triethylamine as base. *Maleate salt.			
5		In two steps, first, condensation of 2-(chloromethyl)-4-isopropyl-			
3	•	sodium azide and a catalytic amount of 18-crown-6 ether in benzene			
		ahydrofuran or benzene-dimethylformamide at room temperature			
		y cloaddition of the resulting 2-(azidomethyl)-4-isopropylsaccharin			
10		ponding acetylene, in this example dimethyl acetylenedicarboxylate,			
		vent with or without heating.			
		In two steps, first, condensation of 2-(chloromethyl)-4-isopropyl-			
15		2-chloro-4-(4-thiamorpholinyl)phenol with acetonitrile as solvent			
		zabicyclodecene as base in 74.7% yield and, second, oxidation of the			
	-	-chloro-4-(4-thiamorpholinyl)phenoxymethyl]-4-isopropyl-saccharin			
20		perbenzoic acid in 38.5% yield.			
	X:	With dimethylformamide as solvent and cesium carbonate as base.			
	AV:	In two steps, first, condensation of 2-(chloromethyl)-4-isopropyl-			
25	saccharin and	2-benzyl-6-hydroxysaccharin with methyl ethyl ketone as solvent			
25	and potassium carbonate as base in 43% yield and, second, debenzylation by				
	catalytic hydrogenation over palladium on carbon in methanol containing				
	ammonium formate in 26% yield.				
30	AX:	With methyl ethyl ketone as solvent and potassium O-ethyldithio-			
	carbonate.				
	BB:	By oxidation of the compound of Example 42AY with one molar			
35	equivalent of m-chloroperbenzoic acid.				
	BC:	By oxidation of the compound of Example 42BB with one molar			
	equivalent of	m-chloroperbenzoic acid.			
40	BF:	Obtained impure.			
	BI:	With acetonitrile as solvent and sodium hydride as base.			
	BO:	With dimethylformamide as solvent and potassium carbonate as			
45	base.				
40	BQ:	In two steps, first, condensation of 2-(chloromethyl)-4-isopropyl-			
		benzyl 2,6-dichloro-3-hydroxybenzoate with dimethylformamide as			
		odium hydride as base and, second, debenzylation by catalytic			
50		over palladium on carbon in methanol containing acetic acid.			
	BT:	With tetrahydrofuran as solvent and triethylamine as base.			

55 Examples 43A - BY

Preparation of 2 - Chloromethyl - 4 - isopropyl - 6 - methoxysaccharin

To a solution of 300 mL N,N,N',N' - tetramethylethylenediamine (TMEDA, 1.99 moles) in 4 L anhydrous ether was added 1550 mL sec-BuLi (1.3 M) and the mixture was cooled to -70°C under a nitrogen atmosphere. A solution of 454.2 g 2-isopropyl-4-methoxy-N,N-diethyl-benzamide (1.82 moles) in 300 mL anhydrous ether was added dropwise over 30 minutes. The temperature was maintained at or below -60°C during the addition. After the addition the mixture was stirred at -70°C for one hour, allowed to warm to -50°C, held at -50°C for 30 minutes, then cooled back to -70°C. By cannulation tube a solution of 200 g SO₂ in 200 mL dry ether precooled to -40°C was added under positive nitrogen pressure over a 20-minute period. The temperature of the reaction mixture during the addition was maintained below -40°C. A white powdery precipitate of aryl lithium sulphinate separated out almost immediately. After the addition the cooling bath was removed and the mixture was stirred at ambient temperature for two hours, then cooled to -5°C. With continued stirring 190 mL sulfuryl chloride (2.36 moles) was added dropwise over a 15 - minute period while maintaining the temperature below 10°C. After further stirring for 30 minutes at 0-5°C, a white insoluble precipitate was filtered off and washed with 2 L ahydrous ether. Removal of the solvent at atmospheric pressure afforded the resulting sulfonyl chloride (a crude dark oil) which was dissolved in 1.4 L THF. The solution was cooled to -10°C, ad 540 mL concentrated aqueous ammonia (28%) was added in portions over 15 minutes. The temperature was kept at 15°C or below throughout the addition. After stirring for 15 minutes at ambient temperature the THF and excess ammonia were removed under vacuum to give a dark oil, which was diluted with 6.0 L water and acidified with 3N HCl to pH 1. The resulting light yellow solid was collected by filtration, washed with 800 mL water, dried at 60°C under vacuum for 18 hours and recrystallized from a mixture of 800 mL of ethyl acetate and 3 L of hexane to give 429 g (72%) of 2-aminosulfinyI-6-isopropyI-4-methoxy-N,Ndiethyl - benzamide,mp 122 - 125° C.

A solution of 429.6 g of the diethylbenzamide (1.31 mole) in 1.5 L acetic acid was refluxed for 20 hours, then cooled to room temperature. The solvent was removed under vacuum. The oily residue was dissolved in 6 L water and the pH was adjusted to 1 with 6N HCl. The crude product was collected by filtration, washed with 2 L water, dried at 60°C under vacuum for 18 hours and recrystallized from ethyl acetate/hexane to give 303 g (91%) of 4-isopropyl-6-methoxysaccharin, mp 188°C.

To a suspension of 24 g paraformaldehyde (0.8 mole) and 86.4 g chlorotrimethylsilane (1.6 moles) in 200 mL 1,2-dichloroethane was added 0.8 ml anhydrous tin (IV) chloride and the resulting solution stirred on a steam bath for one hour. 4-Isopropyl-6-methoxysaccharin (51.4 g, 0.2 mole) was added to the clear solution and the mixture was refluxed for 18 hours, cooled to room temperature and poured into water. The organic layer was separated, washed with 50 mL 2N sodium hydroxide solution, dried over anhydrous magnesium sulfate and concentrated under vacuum. The residue was purified by crystallization from ethyl acetate/hexane to give 57 g (87%) of 2-chloromethyl-4-isopropyl-6-methoxysaccharin, mp 151°C.

Examples 43A-CA of compounds, wherein m is 0 except Example 43F wherein m is 1, R_2 is hydrogen, R_3 is isopropyl and R_4 is 6-methoxy, were prepared from 2-chloromethyl-4-isopropyl-6-methoxysaccharin and in each example the corresponding R_1 - L_n -H except as noted and are described in TABLE I.

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TABLE I

(°C.) Yield (%)
Method_
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Α
145 71
В
5-189.5 69
С
177 32
₁₂ /C ₆ H ₆ D
127 70
H E
161.5
H
F .
4

0

TABLE_I (continued)

5			M.p.(°C.)	Yield (%)
Ů	Example	R_1 - L_n -	From	Method_
	43G	(5-phenyl-1,3,4-oxadiazol-2-	136-137.5	53
		yl)thio	EtOH	E
10	43H	[1-(3-succinoylaminophenyl)-	122-125	78.6
		tetrazol-5-yl]thio	EtOH	В
	43I	2-benzoyl-4,5-dibromoimidazol-	177-180	32
15		1-yl	MeCN/EtOH	I
	43J	(5-benzyl-1,3,4-oxadiazol-2-yl)-	121-123	76
		thio	EtOH	Е
	43K	(5-hydroxy-6-methyl-6,7-dihydro-	*	51
20		1H-1,2,4-triazolo[3,4-b][1,3]-		K
		thiazin-3-yl)thio		
	43L	[2-(3-pyridyl)-1,3,4-oxadiazol-5-	139-141	<i>7</i> 7.6
25		yl]thio	EtOH	K
	43M	(3-ethoxy-4-methyl-1,2,4-triazol-	167-168	70
		5-yl)thio		K
20	43N	[5-(4-trifluoromethylphenyl)-	201.5-203	72.7
30		1,3,4-oxadiazol-2-yl]thio	EtOH	Κ .
	430	[5-(4-methoxyphenyl)-1,3,4-	138-139	85.1
		oxadiazol-2-yl]thio	EtOH	K
35	43P	[5-(4-pyridyl)-1,3,4-oxadiazol-	184-185	45.4
		2-yl]thio	EtOH	K
	43Q	[5-(4-biphenylyl)-1,3,4-	153-155	7 8.9
40		oxadiazol-2-yl]thio	EtOH	K
	43R	[5-(2-pyrazinyl)-1,3,4-	153-154	54.5
		oxadiazol-2-yl]thio	EtOH/MeCN	K
	43T	[5-(2-pyridyl)-1,3,4-	148-149	71.7
45		thiadiazol-2-yl]thio	EtOH	K
	43U	2-thioxo-2,3-dihydro-1,3,4-	250-251	36
		oxadiazol-3-yl	MeCN	K

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TABLE_I (continued)

			M.p.(°C.)	Yield (%)
5	Example	$R_1=L_n$	From	Method_
	43 V	[5-(3-furyl)-1,3,4-oxadiazol-	109-111	
10	45 (2-yl]thio	EtOH	K
	43W	(4-methyl-5-ethoxycarbonyl-	131-132	
		thiazol-2-yl)thio	Еюн	K
	43X	2-thioxo-2,3-dihydro-5-(2-	275-276	
15		pyridyl)-1,3,4-thiadiazol-3-yl		K
	43Y	(4-phenylthiazol-2-yl)thio	121-122	63.0
			EtOH	K
20	43Z	(4,5-dimethylthiazol-2-yl)thio	Noncrys-	53.6
		•	talline	K
	43AA	2,3-dihydro-2-oxo-5-phenyl-1,3,4-	204-205	73
		thiadiazol-3-yl	MeCN	K
25	43AB	[4-(4-morpholinyl)-1,2,5-	124.5-125.5	48
		thiadiazol-3-yl]thio	EtOH	K
	43AC	[5-(2-pyridyl)-1,3,4-oxadiazol-	161-163	64.4
30		2-yl]thio	EtOH/MeCN	K
	43AD	(3-phenyl-2-thioxo-2,3-dihydro-	112-114	7 3.6
		1,3,4-thiadiazol-5-yl)thio	EtOH	AD
35	43AE	2,3-dihydro-2-oxo-5-phenyl-1,3,4-	190-191	61
55		oxadiazol-3-yl		K
	43AF	4-(4-morpholinylsulfonyl)-3-	201-203	58
		trifluoromethylphenoxy	EtOH	K
40	43AG	2,5-difluoro-4-(4-morpholinyl-	171-172	4 6
		sulfonyl)phenoxy		K
	43AH	2,6-dichloro-4-(4,5-dihydro-	176-178	5 0
45		oxazol-2-yl)phenoxy	Et ₂ O	AH
	43A1	4,5-dicyano-1,2,3-triazol-1-yl	163.5-165	
			CCI ₄	I

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TABLE 1 (continued)

5			M.p.(°C.)	Yield (%)
	Example	R_1 - L_{n-}	From	Method_
	40		100 100	76
10	43AJ	2,6-dichloro-4-(2-methyltetrazol-	190-192	
	40 4 77	5-yl)phenoxy	EtOH	АН
	43AK	4,5-dicyano-1,2,3-triazol-2-yl	185-187	•
	40.15		C ₆ H ₆	I ex
15	43AL	4,5-di(t-butylsulfonyl)-1,2,3-	207.5-209	87
		triazol-1-yl	EtOAc	D 7 0
	43AM	3,5-difluoro-4-(4-morpholinyl-	146-149	7 0
20		carbonyl)phenoxy	445 440	K
	43AN	3,5-difluorophenoxy	127-129	79
				K
	43AO	4,5-di(1-piperidinylcarbonyl)-	240-242	8
25		1,2,3-triazol-1-yl		I
	43AP	4,5-di(trifluoromethyl)-1,2,3-	200.5-202.5	71
		triazol-1-yl	C_6H_6	D
30	43AQ	4,5-di(1-piperidinylcarbonyl)-	225-226	64
		1,2,3-triazol-2-yl	EtOAc	I
	43AR	3,5-difluoro-4-(4-morpholinyl-	Glass	5 0
		sulfonyl)phenoxy		K
35	43AS	2-methylthio-5-methyl-6-oxo-1,2-	200-202	41
		dihydro-1,2,4-triazin-1-yl		K
	43AT	[5-(3,5-dimethoxyphenyl)-1,3,4-	144-145	75
40		oxadiazol-2-yl]thio	EtOH	K
	43AU	[5-(4,5-dimethoxyphenyl)-1,3,4-	65-80	35.7
		oxadiazol-2-yl]thio	EtOH	K
	43AV	cyanothio	102-104	44.4
45			EtOH	
	43AW	(4-methyl-5-oxo-6-hydroxy-4,5-		39
		dihydro-1,2,4-triazin-3-yl)thio	EtOAc	AW

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TABLE I (continued)

5	Example	R ₁ -L _n	M.p.(°C.) From	Yield (%) Method
	43AX	2,6-dichloro-4-ethoxycarbonyl-	oil	
10	15.22	phenoxy		K
	43AY	(cycloheptatrienon-2-yl)oxy	148-149	47
		(cyclonopianical = ye,exy		АН
15	43AZ	[5-(4-pentyloxyphenyl)-1,3,4-	113-114	65.5
,,,		oxadiazol-2-yl]thio	EtOH	K
	43BA	(4-ethoxycarbonylisoxazol-5-yl)-	158-160	14
		oxy	EtOH	K
20	43BB	(2,5-dioxopyrrolidin-1-yl)oxy	227-228	54
		•	C ₇ H ₁₆	ВВ
	43BC	2-methyl-4,5-di(hydroxymethyl)-	167-168	10
25		3-pyridyloxy	Et ₂ O	K
	43BD	2,4-dichloro-3-[2-(4-morpholinyl)-	>130*	43
		ethoxycarbonyl]phenoxy	Et ₂ O	E
	43BE	5-phenylsulfonyl-1,2,3-triazol-	164-165	25
30		1-yl	EtOAc	D
	43BF	2-diethylphosphonylphenoxy	104-106	27
				BF
35	43BG	2,6-difluoro-4-(4-morpholinyl-	172-174	27
		sulfonyl)phenoxy		K
	43BH	2,5-difluoro-4-(4-methyl-1-	201-203	37
40		piperazinylsulfonyl)phenoxy		BH
	43BI	2,6-difluoro-4-(4-methyl-1-	201-203	8
		piperazinylsulfonyl)phenoxy		вн
	43BJ	3-benzyloxy-4,5-dihydro-5-oxo-	155.5-157	7 0
45		1,2,4-oxadiazol-4-yl		AH
	43BK	1,2,5-thiadiazol-3-yl	107-109	39.5
			EtOH	K

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TABLE I (continued)

5			M.p.(°C.)	Yield (%)
J	Example	R_{1} - L_{n} -	From	Method
	43BL	{5-{4-[2-(2-methoxyethoxy)-	106-108	79.7
10		ethoxy]phenyl}-1,3,4-oxadiazol-		K
		2-yl}thio		
	43BM	[5-(3,4-methylenedioxyphenyl)-	162-163	
15		1,3,4-oxadiazol-2-yl]thio	EtOH/MeCN	K
	43BN	[5-(2,5-dimethoxyphenyl)-1,3,4-	124-126	71
		oxadiazol-2-yl]thio	EtOH	K
	43BO	(5-methoxycarbonylisoxazol-	131-132	
20		3-yl)oxy	EtOH	K
	43BP	(1-methyl-2-ethoxycarbonylindol-	150-152	38.9
		3-yl)oxy		AH
25	43BQ	[5-(2-methoxyphenyl)-1,3,4-	oil	
		oxadiazol-2-yl]thio		K
	43BR	(5-phenyloxazol-2-yl)thio	128-129	
			EtOH	K
30	43BS	2-(4-methoxybenzoyl)indol-1-yl	177-178	58
				I
	43BT	2-methyl-3-(2,6-dichlorobenzoyl)-	268-270	86
35		indol-1-yl	EtOAc	I
	43BU	(2-phenyl-5-methylthiazol-4-yl)oxy	157-158	51
			EtOH	K
	43BV	(2-methyl-5-phenylthiazol-4-yl)oxy	102-103.5	11
40			EtOH	K
	43BW	[1-(3-pyridyl)tetrazol-5-yl]thio	134-136	63.6
			EtOH	K
45	43BX	2,3,5-trifluoro-4-(4-morpholinyl-	170-171.5	49.3
		sulfonyl)phenoxy	EtOH	K
	43BY	[1-(2,5-dimethoxyphenyl)tetrazol	142-143	15.8
		-5-yl]thio	EtOH	K
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Notes to TABLE I

	A:	With dimethylformamide as solvent and 1-phenyltetrazole-5-thiol
5	sodium salt.	
	B:	With dichloromethane as solvent and triethylamine as base.
	C:	With dimethylformamide as solvent and sodium methoxide as base.
	D:	In two steps, first, condensation of 2-(chloromethyl)-4-isopropyl-
10	6-methoxysac	charin with sodium azide and a catalytic amount of 18-crown-6 ether
	in benzene or	toluene at room temperature and, second, cycloaddition of the
	resulting 2-(a	zidomethyl)-4-isopropylsaccharin with the corresponding acetylene,
15	in this examp	le phenylsulfonyltrimethylsilylacetylene, in the same solvent with or
	without heati	ng.
	E:	With methyl ethyl ketone as solvent and potassium carbonate as
20	base. *Hydro	ochloride salt.
20	F:	In three steps by, first, condensation of the thallium salt of
	4-isopropyl-	6-methoxysaccharin formed from 4-isopropyl-6-methoxysaccharin
	and thallium	ethoxide with 1-phenyl-5-(3-chloro-3-phenylthiopropyl)tetrazole in
25	dimethylform	amide in 57% yield, second, oxidation of the resulting
	2-[3-(1-pheny	yltetrazol-5-yl)-1-(phenylthio)propyl]-4-isopropyl-6-methoxy-
	saccharin wit	h m-chloroperbenzoic acid in 60.5% yield and, third, thermal
30	elimination o	f the resulting 2-[3-(1-phenyltetrazol-5-yl)-1-(phenylthio)propyl}-
	4-isopropyl-6	methoxysaccharin in 75% yield.
	I:	With dimethylformamide as solvent and sodium hydride as base.
	K:	With acetonitrile as solvent and methyltriazabicyclodecene as base.
35	*Mixture (1:1	l) of geometric isomers.
	AD:	With methyl ethyl ketone as solvent and 3-phenyl-2-thioxo-
	2,3-dihydro-1	1,3,4-thiadiazol-5-thiol potassium salt.
40	AH:	With dimethylformamide as solvent and potassium carbonate as
	base.	
	AV:	With acetone as solvent and potassium thiocyanate.
45	AW:	With dimethylformamide as solvent and diisopropylethylamine as
45	base.	
	BB:	With acetonitrile as solvent and diisopropylethylamine as base.
	BF:	With tetrahydrofuran as solvent and potassium t-butoxide as base.
50	BH:	With acetonitrile-dimethylformamide as solvent and
	methyltriazab	icyclodecene as base.

55 Example 44

A solution of 2-chloromethyl-4-ethyl-5,7-dimethoxysaccharin (Example 22L, 0.4 g) and 1-phenyltetrazol-5-thiol sodium salt (0.28 g) in dimethylformamide (3 mL) was heated at 110 °C. for two

hours, then poured into water. Recrystallization of the resulting solid from ethanol – water afforded 2-(1-phenyltetrazol-5-yl)thiomethyl – 4 – ethyl – 5,7 – dimethoxysaccharin, (0.44 g, 74% yield), mp 162-164 °C.

5 Example 45

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A. A solution of 5 - chloro - 2 - benzyl - 2H - isothiazol - 3 - one - 1 - oxide (Example 26A, 1.50 g) and 1 - methoxy - 3 - trimethylsilyloxybutadiene (d = 0.885, 1.30 mL) was heated in a pressure tube at 65° C. overnight. More 1 - methoxy - 3 - trimethylsilyloxybutadiene (0.74 mL) was added and the solution was again heated in a pressure tube at 65° C. overnight then stripped of volatiles. Dichloromethane was added, the mixture was stripped of volatiles, and the addition and stripping were repeated, finally under high vacuum. Dichloromethane was added to the golden amber oil, which solidified after several hours affording 2 - benzyl - 6 - hydroxy - 1, 2 - benzisothiazol - (1H) - 3 - one - 1 - oxide (1.16 g, 68% yield).

- B. A solution of 2-benzyl-6-ydroxy-1,2-benzisothiazol-(1H)-3-one-1-oxide (3.75 g) in methanol (75-100 mL) was added dropwise to a solution of o-monoperphthalic acid magnesium salt (8.14 g) in water (70-100 mL) at room temperature. Methanol (200-250 mL) was added to dissolve the resulting precipitate, and the solution was stirred overnight at room temperature. An equal volume of water was added, and the mixture was extracted with dichloromethane. The dichloromethane extract was washed with water and saturated aqueous sodium chloride solution, dried over sodium sulfate and stripped of dichloromethane. A solution of the resulting solid (4.17 g) in dichloromethane was washed with water and saturated aqueous sodium chloride solution, dried over sodium sulfate and stripped of dichloromethane, affording 2-benzyl-6-hydroxysaccharin (3.87 g, 98% yield).
- C. A mixture of 2 benzyl 6 hydroxysaccharin (0.86 g), 2 bromoethyl 2 methoxyethyl ether (d = 1.347, 0.45 mL), potassium carbonate (1.24 g), methyl ethyl ketone (50 mL) and dimethylformamide (2 mL) was heated under reflux for five hours, then poured into ice water (500 mL). The resulting solid was collected, washed with water and dried affording 2 benzyl 6 [2 (2 methoxyethoxy) ethoxy]saccharin (0.92 g, 78% yield, mp 86 88° C.).
- D. A mixture of 2-benzyl-6[2-(2-methoxyethoxy)ethoxy]saccharin (2.05 g), methanol (75-100 mL), ammonium formate (1.10 g) and palladium on carbon (10%, 1.0 g) was heated under reflux for 40 minutes, allowed to cool and filtered. The filtrate was stripped of volatiles affording 6-[2-(2-methoxyethoxy)ethoxy]saccharin ammonium salt, a mixture of which with chloromethyl phenyl sulfide (0.79 g) and dimethylformamide was heated for eight hours at 100°C ., stirred overnight at room temperature, and poured into ice-water (600 mL). The resulting mixture was extracted with dich-loromethane. The dichloromethane extract was washed with water and saturated aqueous sodium chloride solution, dried over sodium sulfate and stripped of dichloromethane. Column chromatography of the residue (1.70 g) on silica gel using dichloromethane-acetone (98:2) as eluant and again using dichloromethane-acetone (99:1) as eluant afforded 2-phenylthiomethyl-6-[2-(2-methoxyethoxy)-ethoxy]saccharin (0.96 g, 45% yield).
- E. Sulfuryl chloride (0.34 g) was added dropwise with stirring to a solution of 2 phenylthiomethyl 6 [2 (2 methoxyethoxy)ethoxy]saccharin (0.96 g) in dichloromethane. Stirring was continued at room temperature for three hours and the solution was stripped of volatiles. Dichloromethane was added and the solution was stripped again. Hexane was added to the residue and the mixture was stirred at room temperature overnight. The resulting white solid was collected and dried affording 2 chloromethyl 6 [2 (2 methoxyethoxy)ethoxy]saccharin (0.69 g, 89% yield,) mp 113 115° C.
- F. By a method similar to that of Example 44 condensation of 2-chloromethyl-6-[2(2-methox-yethoxy)ethoxy]saccharin (34 g.) and 1-phenyltetrazol-5-thiol sodium salt (0.19 g) in methylethyl-ketone at 60°C. and purification of the product (470 mg) by column chromatography on silica gel first with dichloromethane and then with dichloromethane-acetone (up to 97:3) as eluant afforded 2-(1-phenyltetraol-5-yl)thiomethyl-6-[2-(2-methoxyethoxy)ethoxy]saccharin as an oil (390 g, 82% yield).

Example 46

A solution of 2 – (5 – phenyl – 1 – tetrazolyl)thiomethyl – 4 – ethylsaccharin (Example 30AE, 0.020 g) and 1 – phenyltetrazol – 5 – thiol sodium salt (0.0026 g) in dimethylformamide (1 mL) was heated at 100 °C. for three days, then poured into water. Recrystallization of the resulting solid from ethanol – water afforded 2 – (4 – phenyl – 5 – thioxo – 1 – tetrazolyl)methyl – 4 – ethylsaccharin, (0.012 g, 60% yield), mp 127 – 129 °C.

Example 47

A. By the method of Example 22A 3,4-dimethoxy-2-propyl-N,N-dimethylbezamide(9.2 g) was aminosulfonylated with sulfur dioxide and hydroxylamine-O-sulfonic acid (5.6 g) to give 2-aminosulfonyl-4,5-dimethoxy-6-propyl-N N-dimethylbenzamide (7.4 g, 63% yield), which was cyclized in quantitative yield to 4-propyl-5,6-dimethoxysaccharin, phenylthiomethylation of which with chloromethyl phenyl sulfide (1.42 mL) gave 2-phenylthiomethyl-4-propyl-5,6-dimethoxysaccharin (4.07 g), reaction of part (3.59 g) of which with sulfuryl chloride (2.12 mL) gave 2-chloromethyl-4-propyl-5,6-dimethoxysaccharin, (2.84 g, 97% yield).

B. By the method of Example 44 condensation of 2 - chloromethyl - 4 - propyl - 5,6 - dimethoxysaccharin (0.6 g) and 1 - phenyltetrazol - 5 - thiol sodium salt (0.36 g) in dimethyl - formamide (5 mL) and purification of the product by column chromatography on silica gel using ethyl acetate - hexane (3:7) as eluant followed by trituration with hexane afforded 2 - (1 - phenyltetrazol - 5 - yl)thiomethyl - 4 - propyl - 5,6 - dimethoxy - saccharin, (0.65 g, 76% yield), mp 145 - 146° C.

Example 48

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A. By the method of Example 22A 2-aminosulfonyl-4,5-dimethoxy-6-isopropyl-N,N-dimethyl-benzamide (10.75 g) was prepared and cyclized to 4-isopropyl-5,6-dimethoxysaccharin (mp 18-188°C. from ether-hexane), phenylthiomethylation of part (5 g) of which with chloromethyl phenyl sulfide (2.48 mL) gave 2-phenylthiomethyl-4-isopropyl-5,6-dimethoxysaccharin (4.07 g), reaction of which with sulfuryl chloride (three molar equivalents) gave 2-chloromethyl-4-isopropyl-5,6-dimethoxysaccharin,(85% yield), mp 117-119°C. from ethyl acetate-hexane.

B. By the method of Example 44 condensation of 2-chloromethyl-4-isopropyl-5,6-dimethoxysac-charin (1.46 g) and 1-phenyltetrazol-5-thiol sodium salt (0.92 g) in dimethylformamide (5 mL) and recrystallization of the product from ethanol-water afforded 2-(1-phenyltetrazol-5-yl)thiomethyl-4-isopropyl-5,6-dimethoxy-saccharin, (1.05 g, 51% yield), mp 69-71°C.

Example 49

A. Ethanethiol (43.9 g) was added with stirring to a suspension of aluminum chloride (62.74 g) in chloroform (500 mL) at 0°C. To the resulting solution was added a solution of 4-isopropyl-6-methoxysaccharin (20.0 g) in chloroform (550 mL) during 30 minutes. The resulting solution was warmed to and maintained at 60°C. for 3-4 hours, then poured into ice-water acidified with hydrochloric acid. The resulting solid was collected by filtration, washed with water and dried affording 4-isopropyl-6-hydroxysaccharin, (18.4 g., 97% yield.)

B. By the method of Example 21 phenylthiomethylation of 4-isopropyl-6-hydroxysaccharin (0.004 mole) with chloromethyl phenyl sulfide (0.61 mL) gave 2-phenylthiomethyl-4-isopropyl-6-hydroxy-saccharin 0.32 g, 21% yield), reaction of which with sulfuryl chloride (0.73 g) gave 2-chloromethy-4-isopropyl-6-hydroxysaccharin, (84% yield), mp 149-150°C.

C. By the method of Example 44 condensation of 2-chloromethyl-4-isopropyl-6-hydroxysaccharin (0.3 g) and 1-phenyltetrazol-5-thiol sodium salt (0.23 g) in dimethylformamide (10 mL) and flash chromatography of the resulting product on silica gel using hexane-ethyl acetate (7:3) as eluant afforded 2-(1-phenyltetrazol-1-yl)thiomethyl-4-isopropyl-6-hydroxy-saccharin, (0.3 g, 67% yield), mp 188.5-189.5 °C.

Example 50

A. A solution of 5-chlor-2-benzyl-2H-isothiazol-3-one-1-oxide (Example 26A, 5.8 g) and 2-ethoxyfuran (3.76 g) in benzene (40 mL) was heated to 50 °C. More benzene (30 mL) was added to the solidified reaction mixture, which was then heated under reflux for 15 minutes, then at 70 °C. for 45 minutes, then chilled overnight in the refrigerator. The resulting pale yellow solid was collected, washed with cold benzene and dried affording 2-benzyl-4-ethoxy-7-hydroxy-1,2-benzisothiazol-(1H)-3-one-1-oxide,(3.05 g, 40% yield).

B. m - Chloroperbenzoic acid (8.6 g) was added with stirring to solution of 2 - benzyl - 4 - ethoxy - 7 - hydroxy - 1,2 - benzisothiazol - (1H) - 3 - one - 1 - oxide (7.9 g) in dichloromethane (about 500 mL) and methanol (31 mL). Stirring was continued for several hours, more m - chloroperbenzoic acid (8.6 g) was added and stirring was continued for several days. Dichloromethane was added. The solution was

washed with water. The solid which separated when the solution was washed again with water was collected (2.33 g), mp 196-199 °C. More solid separated when the dichloromethane layer was washed with saturated aqueous sodium chloride solution and was collected (4.10 g), mp 196-199 °C. Both solids were determined to be 2-benzyl-4-ethoxy-7-hydroxysaccharin.

- C. A mixture of 2-benzyl-4-ethoxy-7-hydroxysaccharin (1.0 g), allyl bromide (0.36 g), potassium carbonate (0.62 g) and methyl ethyl ketone (about 20 mL) was heated at 80 °C. for one hour. More allyl bromide (0.36 g) was added and heating at 80 °C. was continued for one and one-half hours. The mixture was cooled and poured into ice-water (500 mL). The solid was collected affording 2-benzyl-4-ethoxy-7-allyloxysaccharin, (1.08 g, 96% yield), mp 148-149 °C.
- D. A mixture of 2 benzyl 4 ethoxy 7 allyloxysaccharin (0.25 g) and triglyme was heated at 200 °C. for 20 minutes, examined for extent of reaction by thin layer chromatography, heated at 200 °C. for 20 minutes more, stirred overnight and poured into ice water. The resulting tacky, brown solid was dissolved in dichloromethane The dichloromethane solution was washed with saturated aqueous sodium chloride solution, dried over sodium sulfate and stripped of dichloromethane. Column chromatography of the residue (0.29 g) on silica using dichloromethane and dichloromethane acetone (up to 99:1) as eluants afforded 2 benzyl 4 ethoxy 6 allyl 7 hydroxysaccharin (90 mg, 36% yield) whose mass spectrogram showed a molecular ion at mass 373.
- E. A mixture of 2-benzyl-4-ethoxy-6-allyl-7-hydroxysaccharin (2.45 g), potassium carbonate (2.76 g) and methyl iodide (1.84 g) in acetone was heated at 50 °C.with stirring for two hours, then poured with stirring into ice-water (500 mL). The resulting gooey precipitate crystallized overnight affording <math>2-benzyl-4-ethoxy-6-allyl-7-methoxysaccharin, (2.35 g, 92% yield), mp 92-94 °C.
- F. By the method of part D of Example 45 2 benzyl 4 ethoxy 6 allyl 7 methoxysaccharin (2.35 g) was simultaneously debenzylated and hydrogenated with ammonium formate (1.51 g) and palladium on carbon (10%, 1.25 g) in methanol (70 100 mL) affording 4 ethoxy 6 propyl 7 methoxysac charin ammonium salt (1.94 g), phenylthiomethylation of which with chloromethyl phenyl sulfide (0.95 g) in dimethylformamide and purification of the product (a yellow oil, 2.67 g) by column chromatography on silica gel using dichloromethane hexane (80:20) as eluant afforded 2 phenylthiomethyl 4 ethoxy 6 propyl 7 methoxysaccharin, (0.85 g, 34% yield).
- G. By the method of part E of Example 45 2 phenylthiomethyl 4 ethoxy 6 propyl 7 methox ysaccharin (0.85 g) was reacted with sulfuryl chloride (0.30 g) in dichloromethane and purified with hexane affording 2 chloromethyl 4 ethoxy 6 propyl 7 methoxysaccharin, (0.62 g, 89% yield), mp 131 133 °C.
- H. By the method of part F of Example 45 condensation of 2 chloromethyl 4 ethoxy 6 propyl 7 methoxysaccharin (0.62 g) and 1 phenyltetrazol 5 thiol sodium salt (0.36 g initially and a small amount more after 3.5 hours reaction time, total reaction time 8 hours) and isolation of the oily product by extraction with dichloromethane and purification thereof first by column chromatography on silica gel using dichloromethane as eluant and then by crystallization of the resulting oil (0.62 g., 71% yield) from ethanol gave 2 (1 phenyltetrazol 5 yl)thiomethyl 4 ethoxy 6 propyl 7 methoxysaccharin, mp 110 111 °C.

Example 51

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- A. s Butyllithium (0.87M in cyclohexane, 20.4 mL) was added dropwise during one hour with stirring at -78°C. to a solution of 4-ethyl-5,7-dimethoxysaccharin (Example 22L, 2.2 g) in tetrahydrofuran (100 mL). After continued stirring at -78°C. for one hour methyl iodide (1.5 mL) was added. Stirring was continued at -78°C for 15 minutes, the temperature was allowed to rise to room temperature, and the mixture was quenched in water. Aqueous sodium hydroxide (0.5%, 200 mL) was added. The mixture was washed with ethyl acetate (200 mL), acidified with concentrated hydrochloric acid and extracted with ethyl acetate (200 mL). The ethyl acetate extract was washed with aqueous sodium thiosulfate (10%, 50 mL) and saturated aqueous sodium chloride (50 mL), dried over sodium sulfate and stripped of ethyl acetate affording 4-ethyl-5,7-dimethoxy-6-methylsaccharin (0.73 g, 32% yield).
- B. By the method of Example 21 phenylthiomethylation of 4 ethyl 5,7 dimethoxy 6 methylsac charin (0.65 g) with chloromethyl phenyl sulfide (0.24 mL) gave 2 phenylthiomethyl 4 ethyl 5,7 dimethoxy 6 methylsaccharin, reaction of which with sulfuryl chloride gave 2 chloromethyl 4 ethyl 5,7 dimethoxy 6 methylsaccharin, (0.16 g, 22% yield).
- C. By the method of Example 44 condensation of 2-chloromethyl-4-ethyl-5,7-dimethoxy-6-methylsaccharin (0.17 g) and 1-phenyltetrazol-5-thiol sodium salt (0.084 g) in dimethylformmide (4 mL) and purification of the product by flash column chromatography on silica gel using hexane-ethyl

acetate (75:25) as eluant afforded 2 - (1 - phenyltetrazol - 5 - yl)thiomethyl - 4 - ethyl - 5,7 - dimethoxy - 6 - methylsaccharin,(0.15 g, 76% yield), mp 44 - 46° C.

Example 52

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A. A mixture of 2-benzyl-4-ethoxy-7-hydroxysaccharin (part B of Example 50,0.5 g), t-butyl bromoacetate (0.3 g, 0.25 mL), potassium carbonate (0.30 g) and methyl ethyl ketone (about 10 mL) was heated at 70-80°C. for about one hour, let cool and poured into ice-water (400 mL). The mixture was stirred overnight and the solid was collected affording 2-benzyl-4-ethoxy-7-(t-butoxycarbonyl-methoxy)saccharin.(0.58 g, 86% yield).

B. By the method of part D of Example 45 in two successive preparations 2-benzyl-4-ethoxy-7-(t-butoxycarbonylmethoxy)saccharin (0.5 g, 6.5 g) was debenzylated with ammonium formate (0.25 g, 3.66 g) and palladium on carbon (10%, 0.25 g, 2 g) in methanol affording 4-ethoxy-7-(t-butoxycar-bonylmethoxy)saccharin ammonium salt (0.44 g, 5.63 g), phenylthiomethylation of which with chloromethyl phenyl sulfide (0.2 g, 2.38 g) in dimethylformamide and purification of the combined products (0.46 g, 6.42 g) by column chromatography on silica gel using dichloromethane and dichloromethane - acetone (98:2) as eluants afforded 2-phenylthiomethyl-4-ethoxy-7-(t-butox-ycarbonylmethoxy)saccharin, (1.92 g, 40% yield).

C. By the method of part E of Example 45 2 – phenylthiomethyl – 4 – ethoxy – 7 – (t – butoxycarbony – lmethoxy)saccharin (1.92 g) was reacted with sulfuryl chloride (0.54 g) in dichloromethane at about 5 °C. and purified with hexane affording 2 – chloromethyl – 4 – ethoxy – 7 – (t – butoxycarbonylmethoxy) – saccharin,(1.54 g, 95% yield), mp 117 – $1\overline{19}$ °C.

D. By the method of part F of Example 45 condensation of 2-chloromethyl-4-ethoxy-7-(t-butoxycarbonylmethoxy)saccharin (1.42 g) and 1-phenyltetrazol-5-thiol sodium salt (0.70 g initially and a small amount more after 6 hours reaction time, total reaction time 8 hours) and isolation of the product (1.20 g) by extraction with dichloromethane and purification of part (0.42 g) thereof by column chromatography on silica gel using dichloromethane – acetone (up to 98:2) as eluant afforded 2-(1-phenyltetrazol-5-yl)thioniethyl-4-ethoxy-7-(t-butoxycarbonylmethoxy)saccharin, (300 mg, 71% yield), mp 110-112°C.

Example 53

A solution of 2-(1-phenyltetrazol-5-yl)thiomethyl-4-ethoxy-7-(t-butoxycarbonylmethoxy)-saccharin (Example 52, 0.47 g) in trifluoroacetic acid (5-10 mL) and dichloromethane (5-10 mL) was stirred at room temperature for two hours, then stripped of volatiles, then stripped three times more from dichloromethane, then stripped once more from acetone. A solution of the residual oil in acetone (1 mL) was added to ice-water (100 mL) containing concentrated hydrochloric acid (1 mL). The resulting solid was collected, washed with water and dried (0.34 g). A solution thereof in dichloromethane was eluted through silica gel in a 15-mL sintered glass funnel with chloroform-methanol (95:5) affording 2-(1-phenyltetrazol-5-yl)thiomethyl-4-ethoxy-7-carboxymethoxysaccharin, (70 mg, 17% yield), mp 187-189°C.

Example 54

A. Bromination of 1,2 – dimethoxy – 4 – isopropylbenzene (31 g) with N – bromosuccinimide on Kieselgel in tetrachloridemethane by the method of Hisatoshi et al. (Bulletin of the Chemical Society of Japan, vol. 32, pp. 591 – 593, 1989) gave 5 – bromo – 1,2 – dimethoxy – 4 – isopropylbenzene, lithiation of which with n – butyllithium in ether gave 5 – lithio – 1,2 – dimethoxy – 4 – isopropylbenzene, diethylamino – car – bonylation of which in the same solvent gave 2 – isopropyl – 4,5 – dimethoxy – N,N N – diethylbenzamide (15.2 g) as a viscous oil.

B. By the method of Example 43 aminosulfonylation of this 2-isopropyl-4,5-dimethoxy-N,N-diethylbenzamide with s-butyllithium, sulfur dioxide, sulfuryl chloride and concentrated ammonia gave 2-isopropyl-4,5-dimethoxy-6-aminosulfonyl-N,N-diethylbenzamide (4.5 g), mp 181-182°C., cyclization of which in acetic acid gave 4-isopropyl-6,7-dimethoxysaccharin, (2.86 g), mp 210-212°C.

C. Diisopropylethylamine (0.5 mL) was added to a solution of 4-isopropyl-6,7-dimethoxysaccharin (0.5 g) in dimethylformamide (3 mL). After 15 minutes chloromethyl phenyl sulfide (0.35 g) was added The mixture was heated at 80°C. for 16 hours, diluted with ethyl acetate, washed with aqueous sodium

carbonate solution, hydrochloric acid (3N) and saturated aqueous sodium chloride solution, dried over sodium sulfate, and stripped of solvents. Purification of the residue by flash chromatography on silica gel using dichloromethane as eluant gave 2-phenylthiomethyi-4-isopropyl-6,7-dimethoxysaccharin - (0.35 g), chlorination of which with sulfuryl chloride (0.1 mL) in dichloromethane (3 mL) and purification of the product by trituration with hexane gave 2-chloromethyl-4-isopropyl-6,7-dimethoxysaccharin, (0.3 g).

D. By a method similar to that of Example 44 condensation of 2- chloromethyl -4- isopropyl -6,7- dimethoxysaccharin (0.095 g) and 1- phenyltetrzol -5- thiol sodium salt (0.120 g) in acetonitrile (2 mL) and purification of the product by crystallization from ethanol afforded 2-(1- phenyltetrazol -5- yl) - thiomethyl -4- isopropyl -6,7- dimethoxy -6- methyl - saccharin, (0.099 g, 70% yield), mp 169-171 °C.

Example 55

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A. By the methods of parts D and E of Example 45 debenzylation of 2-benzyl-4-ethoxy-7-hydroxysaccharin (part B of Example 50) with ammonium formate and palladium on carbon in methanol gave 4-ethoxy-7-hydroxysaccharin ammonium salt, phenylthiomethylation of which with chloromethyl phenyl sulfide in dimethylformamide gave 2-phenylthiomethyl-4-ethoxy-7-hydroxysaccharin, reaction of which with sulfuryl chloride in dichloromethane gave 2-chloromethyl-4-ethoxy-7-hydrox-ysaccharin.

B. By the method of part F of Example 45 condensation of 2-chloromethyl-4-ethoxy-7-hydrox-ysaccharin (1.95 g) and 1-phenyltetrazol-5-thiol sodium salt (1.5 g, reaction time about 8 hours) and purification of part of the product (2.5 g., 88% yield) from ethanol gave 2-(1-phenyltetrazol-5-yl)-thiomethyl-4-ethoxy-7-hydroxysaccharin, mp 178-180°C.

Example 56

A mixture of 2-(1-phenyltetrazol-5-yl)thiomethyl-4-ethoxy-7-hydroxysaccharin (Example 55,0.5 g), dimethylcarbamyl chloride (0.14 mL, 0.16 g), dizabicyclooctane (0.27 g) and N,N-dimethylacetamide (20 mL) was heated at 80° C. for 4-5 hours, cooled and poured with stirring into ice-water (400 mL). Column chromatography of the resulting solid on silica gel by elution with dichloromethane and dichloromethane – acetone (up to 98:2) gave first 2-(4-phenyl-5-thioxo-1-tetrazolyl)methyl-4-ethoxy-7-dimethylaminocarbonylsaccharin (mp $172-173^{\circ}$ C. after recrystallization from ethanol) and then 2-(1-phenyltetrazol-5-yl)thiomethyl-4-ethoxy-7-dimethylaminocarbonyloxysaccharin (mp $145-146^{\circ}$ C. after sonication in ether).

Example 57

2-(1-phenyltetrazol-5-yl)thiomethyl-4-isopropyl-5,6-dimethoxysaccharin (0.48 g) was added with stirring at 0°C. to a solution of aluminum chloride (0.4 g) and ethanethiol (0.15 mL) in chloroform (3 mL). The temperature was allowed to rise to and remain at room temperature for 18 hours. The reaction mixture was passed through silica gel with ethyl acetate – hexane (2:3) as eluant affording as a colorless oil 2-(1-phenyltetrazol-5-yl)thiomethyl-4-isopropyl-5-hydroxy-6-methoxy-saccharin,(0.3 g, 77% yield), whose structure was proven by proton and carbon-13 magnetic resonance, mass spectrometric and nuclear Overhauser enhancement analyses. The latter analysis showed 19% enhancement of the methoxy protons signal and 10% enhancement of the C-7 proton signal.

BIOLOGICAL TEST RESULTS

In the test of elastase inhibition using human leukocyte elastase described above at pages 89 – 90 the compounds of formula I of Examples 41 – 57 showed K_i* values in the range from 0.024 nM (Example 43AL) to 1000 nM (Example 42D, for example).

In the claims "corresponding" means that a variable of the structural formula of claim 1 (formula I above) which is repeated in another formula has the same meaning in the other formula as it has in the formula of claim 1 unless otherwise stated and that "substituted phenyl" and "substituted heterocyclyl" mean phenyl and heterocyclyl substituted as defined in claim 1.

Claims

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1. A compound having the formula

 R_4 $N \longrightarrow (CH = CH)_{in}CH \cdot L_n \cdot R_4$ Formula 1

wherein:

L is -0-, -S-, -SO- or $-SO_2-$;

m and n are each independently 0 or 1;

R₁ is phenyl substituted by 1 – (4 – lower – alkylpiperazin – 1 – yl)carbonyl, 4 – morpholinylsulfonyl, formyl, lower – alkoxycarbonyl, 4 – thiamorpholinylsulfonyl or the S – dioxide thereof, hydroxy – lower – alkyl, halo – lower – alkyl, 4 – morpholinyl – lower – alkylaminocarbonyl, 4 – morpholinyl – lower – alkox – ycarbonyl, 1 – (4 – lower – alkyl – piperzin – 1 – yl)sulfonyl, 4 – morpholinyl – lower – alkoxy, di – lower – alkylamino – lower – alkylaminosulfonyl or an N – lower – alkyl derivative thereof, lower – alkylsulfonyl, 4,5 – dihydrooxazol – 2 – yl, lower – alkyltetrazol – 5 – yl, 4 – morpholinyl – carbonyl, nitrophenylazo, car – boxyl or di – lower – alkylphosphonyl,

or any of the above in combination with substitution of phenyl by halogen, lower-alkyl, lower-alkoxy, nitro, amino, lower-alkylamino or diallkylamino as defined in published PCT application WO 90/13549,

or heterocyclyl selected from pyridazin -3-yl, 4-pyron-3-yl, quinolin -8-yl, 1,3,4-coxadiazol-2-yl, coumarin -7-yl, saccharin -6-yl, imidazol -1-yl, 1,3,4-triazol-2-yl, thiazol -2-yl, 2-thioxo-2,3-dihydro-1,3,4-coxadiazol-3-yl, 2-thioxo-2,3-dihydro-1,3,4-thiadiazol-3-yl, 2-thioxo-2,3-dihydro-1,3,4-thiadiazol-5-yl, 1,2,3-triazol-2-yl, 1,2,4-triazin-5-yl, 1,2,4-triazin-5-yl, 1,2,4-triazin-3-yl, 1,2,4-triazin-3-yl, 1,2,4-triazin-3-yl, 1,2,4-triazin-3-yl, 1,2,4-triazin-3-yl, 1,2,3-triazol-3-yl, 1,3,4-triazol-3-yl, 1,3,4-tria

or said heterocyclyl substituted on any available nitrogen atom by lower – alkyl, hydroxy – lower – alkyl, cycloalkyl, 2 –, 3 – or 4 – pyridinyl, lower – alkoxycarbonyl, lower – alkoxycarbonyl – lower – alkyl, aminocarbonyl – lower – alkyl, lower – alkylaminocarbonyl – lower – alkyl, di – lower – alkylaminocarbonyl – lower – alkyl, lower – alkylamino – lower – alkyl, di – lower – alkylamino – lower – alkyl, 1 – piperidinyl – lower – alkyl, 1 – pyrrolidinyl – lower – alkyl or phenyl (or phenyl substituted by amino, lower – alkylamino, di – lower – alkylamino, lower – alkylamino, lower – alkylamino, carboxy, lower – alkoxycarbonyl, lower – alkoxy or halo),

or said heterocyclyl substituted on any available carbon atom by nitro, lower – alkyl, amino, lower – alkylamino, di – lower – alkylamino, cycloalkylamino, mercapto, lower – alkylthio, amino – lower – alkyl – thio, lower – alkylaminolower – alkylthio, di – lower – alkylamino – lower – alkylthio, 4 – morpholinyl – lower – alkylthio, 1 – piperidinyl – lower – alkylthio, 1 – pyrrolidinyl – lower – alkylthio, lower – alkylamino – lower – alkyl, 4 – morpholinyl – lower – alkylamino, cyano, 1 – piperidinyl – lower – alkyl, hydroxy – lower – alkyl, phenylsulfonyl, toluenesulfonyl, halo, tri – lower – alkylsilyl, carboxy or alkali metal salt thereof, furyl, trifluoromethyl, 2 – benzothiazolyl, lower – alkylsulfonyl, aminocarbonyl, benzyl, 4 – morpholinyl, pyridinyl, lower – alkoxy, pyrazinyl, lower – alkoxycarbonyl – lower – alkyl, di – lower – alkylaminosulfonyl, 1 – piperidinylcarbonyl, benzyloxy, hydroxy, 4 – morpholinyl – lower – alkyl, benzoyl or benzoyl substituted by lower – alkoxy or halo, or phenyl (or phenyl substituted by amino, lower – alkylamino, di – lower – alkylamino, lower – alkoxy – poly – lower – alkoxy, methylenedioxy lower – alkyl, lower – alkoxy, halo, trifluoromethyl, lower – alkoxy – poly – lower – alkoxy, methylenedioxy

or lower alkoxycarbonyl),

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or heterocyclyl selected from 1H-(5-tetrazolyl), 5-oxo-1-tetrazolinyl, 5-thioxo-1-tetrazolinyl (when R_2 as defined hereinbelow is other than phenylthio), pyrimidinyl, 2-benzoxazolyl, 2-benzothiazolyl, 2-phthalimidyl, 2-(1,3,4-thiadiazoyl), 5-(1,2,4-thiadiazolyl), 5-thioxo-3-(1,2,4-thiadiazolinyl), 4-(5-oxo-1,3,4-thiadiazolinyl), 4-(5-thioxo-1,3,4-thiadiazolinyl), 3-(1,2,4-triazolyl), 4-(1,2,4-triazolyl), 1,2,3-triazol-1-yl, 2-imidazolyl and 3-(1,2,4-triazolo[4,3-a]pyridinyl) when substituted on any available carbon atom by di-lower-alkylamino-lower-alkyl, 4-morpholinyl-lower-alkylamino, cyano, 1-piperidinyl-lower-alkyl, hydroxy-lower-alkyl, phenyl-sulfonyl, toluenesulfonyl, halo, tri-lower-alkylsilyl, carboxy or alkali metal salt thereof, furyl, trifluoromethyl, 2-benzothiazolyl, lower-alkoxycarbonyl, aminocarbonyl, benzyl, 4-morpholinyl, pyridinyl, lower-alkoxy, pyrazinyl, lower-alkoxycarbonyl-lower-alkyl, di-lower-alkylaminosulfonyl, 1-piperidinylcarboxyl, benzyloxy, hydroxy, 1-piperidinylcarboxyl, benzoyl or benzoyl substituted by lower-alkoxy or halo, or phenyl substituted by trifluoromethyl, lower-alkoxy-poly-lower-alkoxy, methylenedioxy or lower alkoxycarbonyl,

or, when R₄ is carboxy-lower-alkoxy, lower-alkoxycarbonyl-lower-alkoxy or di-lower-al-kylaminocarbonyloxy, heterocyclyl is 1 - phenyltetrazol-5-yl,

or, when L is -O- and n is 1, R^1 is cycloheptatrienon -2-yl or, when L is -S- and n is 1, R^1 is cyano or lower-alkoxythiocarbonyl or, when L is $-SO_2$ -and n is 1, R^1 is lower-alkyl or trifluoromethyl;

R₂ is hydrogen, lower - alkoxycarbonyl, phenyl or phenylthio;

R₃ is hydrogen, halo, primary or secondary lower – alkyl, lower – alkoxy, lower – alkoxycarbonyl, phenyl, fluoro – lower – alkyl, lower – alkenyl, cyano or di – lower – alkylamino; and

R₄ is hydrogen or from one to three substituents selected from halo, cyano, nitro, amino, lower – alkanamido, phenyl – lower – alkanamido, diphenyl – lower – alkanamido, lower – alkylsulfonylamino, polyfluoro – lower – alkylsulfonyl – amino, aminosulfonyl, lower – alkyl, polyhalo – lower – alkyl, cycloalkyl, polyhalo – lower – alkoxy, hydroxy, lower – alkoxy, hydroxymethyl, formyl, aminomethyl, lower – alkylsulfonyl, polyhalo – lower – alkylsulfonyl, lower – alkylsulfonyl, lower – alkoxy – lower – alkoxy or di – lower – alklaminocarbonyloxy; provided that

- (1) when n is 0, R_1 can only be heterocyclyl and CHR_2 can only be bonded to a ring nitrogen atom of R_1 and
- (2) when m is 0, n is 1, L is $-O_-$, $-S_-$ or $-SO_-$, and R_2 , R_3 and R_4 are all hydrogen or, when m is 0, n is 1, L is $-S_-$, R_2 and R_4 are hydrogen and R_3 is halo, or when m is 0, n is 1, L is $-SO_-$ or $-SO_2_-$, R_2_- is lower-alkoxycarbonyl and R_3_- and R_4_- are both hydrogen, R_1_- cannot be substituted phenyl.
- 2. A compound as claimed in claim 1, wherein L is -0- or -S-, m is 0, n is 0 or 1 and R_2 is hydrogen.
- 40 3. A compound as claimed in claim 2, wherein

 R_3 is hydrogen, chloro, bromo, methyl, ethyl, propyl, isopropyl, sec. – butyl, methoxy, ethoxy, isopropoxy or phenyl and

 R_4 is hydrogen, 7-chloro, 5-nitro, 6-nitro, 5-amino, 5-acetylamino, 5-(3,3-diphenyl-pro-pionamido), 5-(1,1,3,3-tetramethylbutyl), 6-hydroxy, 7-hydroxy, 5-methoxy, 6-methoxy, 7-methoxy, 5,6-dimethoxy, 5,7-dimethoxy, 6,7-dimethoxy, 6-[2-(2-methoxyethoxy)ethoxy], 7-[2-(2-methoxyethoxy)ethoxy], 7-carboxymethoxy, 7-[2-(2-methoxyethoxy)ethoxy], 7-methoxy, 5,7-dimethoxy, 6-methyl or 5-hydroxy-6-methoxy.

- 50 4. A compound as claimed in claim 3, wherein L is -O-, n is 1 and
 - R₁ is substituted phenyl selected from
 - 2 formyl 4 nitrophenyl,
 - 2 hydroxymethyl 4 nitrophenyl,
 - 2 chloromethyl 4 nitrophenyl,
 - 4 chloromethyl 2 nitrophenyl,
 - 4 (4 nitrophenylazo)phenyl,
 - 2,4 dichloro 3 (4 methyl 1 piperazinylcarbonyl)phenyl,
 - 4 [(4 methylpiperazin 1 yl)carbonyl]phenyl,

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2,4 - dichloro - 6 - (4 - morpholinylsulfonyl)phenyl,
         2 - chloro - 4(4 - morpholinylsulfonyl)phenyl,
         2 - methoxycarbonyl - 5 - methoxyphenyl,
         2 - fluoro - 4 - (4 - morpholinylsulfonyl)phenyl,
         2 - chloro - 4 - (1.1 - dioxo - 4 - thiamorpholinylsulfonyl)phenyl,
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         2.6 - difluoro - 4 - (4 - morpholinylsulfonyl)phenyl,
         4,6 - difluoro - 4(4 - morpholinylsulfonyl)phenyl,
         4,5 - difluoro - 4 - (4 - morpholinylsulfonyl)phenyl,
         4 - fluoro - 2 - (4 - morpholinylsulfonyl)phenyl,
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         2.5 - difluoro - 2 - (4 - morpholinylsulfonyl) - phenyl,
         3 - [2 - (4 - morpholinyl)ethylaminocarbonyl]phenyl,
         2.4 - dichloro - 3 - [2 - (4 - morpholinyl) - ethoxycarbonyl]phenyl,
         3 - [(4 - methylpiperazin - 1 - yl)sulfonyl]phenyl,
         3 - [2 - (4 - morpholinyl)ethoxy]phenyl,
         3 - \{2 - [(dimethylamino)ethyl]methylaminosulfonyl\}phenyl,
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         2.4 - dichloro - 3 - carboxyphenyl,
         3 - [2 - (4 - morpholinyl)ethoxycarbonyl]phenyl,
         2,4 - dichloro - 3 - [2(4 - morpholinyl)ethylaminocarbonyl]phenyl,
         4 - (4 - morpholinylsulfonyl) - 3 - trifluoromethylphenyl,
         2.5 - difluoro - 4 - (4 - morpholinylsulfonyl)phenyl,
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         2,6 - dichloro - 4 - (4,5 - dihydrnoxazol - 2 - yl)phenyl,
         2.6 - dichloro - 4 - (2 - methyltetrazol - 5 - yl)phenyl,
         3,5 - difluoro - 4 - (4 - morpholinylcarbonyl)phenyl,
         3,5 - difluoro - 4 - (4 - morpholinylsulfonyl)phenyl,
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         2,6 - dichloro - 4 - ethoxycarbonylphenyl,
         2,4 - dichloro - 3 - [2 - (4 - morpholinyl)ethoxycarbonyl]phenyl,
         2 - diethylphosphonylphenyl,
         2.5 - difluoro - 4 - (4 - methy - 1 - piperazinylsulfonyl)phenyl,
         2,6 - difluoro - 4 - (4 - methyl - 1 - piperazinylsulfonyl)phenyl or
         2,3,5 - trifluoro - 4 - (4 - morpholinylsulfonyl)phenyl.
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         A compound as claimed in claim 4, wherein R<sub>1</sub> is phenyl substituted by fluoro and 4 - morpholinylsul -
         fonyl.
         The compoundas claimed in, claim 5 wherein R<sub>1</sub> is 2,3,5 - trifluoro - 4 - (4 - morpholinylsulfonyl)phenyl,
         R<sub>3</sub> is isopropyl and R<sub>4</sub> is 6 - methoxy.
     7. A compound as claimed in claim 3, wherein L is -O-, n is 1 and
         R<sub>1</sub> is heterocyclyl or substituted heterocyclyl selected from
4Ω
         5,7 - dichloroquinolin - 8 - vl.
         2 - methyl - 4 - pyron - 3 - yl
         6 - hydroxymethyl - 4 - pyron - 3 - yl
         4,5 - dichloropyridazin - 3 - yl,
         2 - \text{ethyl} - 4 - \text{pyron} - 3 - \text{yl}
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         3 - phenylcoumarin - 7 - yl,
         4 - phenylcoumarin - 7 - yl,
         6 - chloro - 4 - trifluoromethylcoumarin - 7 - yl,
         4 - methylcoumarin - 7 - yl,
         3 - (benzothiazol - 2 - yl)coumarin - 7 - yl,
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         saccharin - 6 - yl,
         4 - (4 - morpholinyl) - 1,2,5 - thiadizol - 3 - yl,
         3 - \text{methylthio} - 6 - \text{methyl} - 1,2,4 - \text{triazin} - 5 - \text{yl}
         4 - ethoxycarbonylisoxazol - 5 - yl,
         2,5 - dioxopyrrolidin - 1 - yl,
         2 - methyl - 4,5 - di(hydroxymethyl) - 3 - pyridyl,
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         5 - methoxycarbonylisoxazol - 3 - yl,
         1 - methyl - 2 - ethoxycarbonylindol - 3 - yl.
```

2 - phenyl - 5 - methylthiazol - 4 - yl or

2 - methyl - 5 - phenylthiazol - 4 - yl.

```
8. A compound as claimed in claim 3, wherein L is -S-, n is 1 and
          R<sub>1</sub> is heterocyclyl or substituted heterocyclyl selected from
          5 - (2 - diethylaminoethyl) - 1,3,4 - thiadiazol - 2 - yl,
5
          5 - [2 - (4 - morpholinyl)ethylamino] - 1,3,4 - thiadiazol - 2 - yl,
          5 - [2 - (4 - morpholinyl)ethyl] - 1,3,4 - thiadiazol - 2 - yl,
          5 - [2 - (1 - piperidinyl)ethyl] - 1,3,4 - thiadiazol - 2 - yl,
          5 - phenyl - 1,3,4 - oxadiazol - 2 - yl,
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          4 - (ethoxycarbonylmethyl)thiazol - 2 - yl,
          5 - (2 - \text{furyl}) - 1.3.4 - \text{oxadiazol} - 2 - \text{vl}
          5 - benzyl - 1.3.4 - oxadiazol - 2 - yl,
          5 - hydroxy - 6 - methyl - 6,7 - dihydro - 1H - 1,2,4 - triazolo[3,4 - b][1,3]thiazin - 3 - yl,
          5 - (3 - pyridyl) - 1,3,4 - oxadiazol - 2 - yl,
15
          3 - \text{ethoxy} - 4 - \text{methyl} - 1,2,4 - \text{triazol} - 5 - \text{yl},
          5 - (4 - trifluoromethylphenyl) - 1,3,4 - oxadiazol - 2 - yl,
          5-(4-methoxyphenyl)-1.3.4-oxadiazol-2-yl
          5 - (4 - pyridyl) - 1.3.4 - oxadiazol - 2 - yl,
          5 - (4 - biphenylyl) - 1,3,4 - oxadiazol - 2 - yl,
          5 - (pyrazinyl) - 1,3,4 - oxadiazol - 2 - yl,
20
          5 - (2 - pyridyl) - 1,3,4 - thiadiazol - 2 - yl,
          5 - (3 - furyl) - 1,3,4 - oxadiazol - 2 - yl,
          4 - methyl - 5 - ethoxycarbonylthiazol - 2 - yl,
          4 - phenylthiazol - 2 - yl,
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          4,5 - dimethylthiazol - 2 - yl,
          4-(4-morpholinyl)-1,2,5-thiadiazol-3-yl,
          5 - (2 - pyridyl) - 1,3,4 - oxadiazol - 2 - yl,
          3 - phenyl - 2 - thioxo - 2,3 - dihydro - 1,3,4 - thiadiazol - 5 - yl,
          5 - (3.5 - dimethoxyphenyl) - 1.3.4 - oxadiazol - 2 - yl,
          5-(4,5-dimethoxyphenyl)-1,3,4-oxadiazol-2-yl,
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          4 - methyl - 5 - oxo - 6 - hydroxy - 4,5 - dihydro - 1,2,4 - triazin - 3 - yl,
          5 - (4 - pentyloxyphenyl) - 1,3,4 - oxadiazol - 2 - yl,
          5 - \{4 - [2 - (2 - methoxyethoxy)ethoxy]phenyl\} - 1,3,4 - oxadiazol - 2 - yl,
          5 - (3.4 - \text{methylenedioxyphenyl}) - 1.3.4 - \text{oxadiazol} - 2 - \text{yl}
          5-(2.5-dimethoxyphenyl)-1.3.4-oxadiazol-2-yl,
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          5 - (2 - methoxyphenyl) - 1,3,4 - oxadiazol - 2 - yl or
          5 - phenyloxazol - 2 - yl.
     9. A compound as claimed in claim 3, wherein n is 0 and
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          R<sub>1</sub> is heterocyclyl or substituted heterocyclyl selected from
          4.5 - dicyanoimidazol - 1 - yl,
          1,1,3 - trioxotetrahydro - 1,2,5 - thiadiazol - 2 - yl,
          4.5 - di(methoxycarbonyl) - 1.2.3 - triazin - 1 - yl
          5 - phenylsulfonyl - 1,2,3 - triazol - 1 - yl,
45
          4 - trimethylsilyl - 5 - phenylsulfonyl - 1,2,3 - triazol - 1 - yl,
          4 - methoxycarbonyl - 1,2,3 - triazol - 1 - yl,
          5 - methoxycarbonyl - 1,2,3 - triazol - 1 - yl,
          4 - \text{ethoxycarbonyl} - 5 - \text{phenyl} - 1,2,3 - \text{triazol} - 1 - \text{yl},
          4 - phenyl - 5 - ethoxycarbonyl - 1,2,3 - triazol - 1 - yl,
          4 - carboxy - 1,2,3 - triazol - 1 - yl,
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          4 - phenyl - 5 - (4 - methylphenylsulfonyl) - 1,2,3 - triazol - 1 - yl,
          4-(4-methylphenylsulfonyl)-5-phenyl-1,2,3-triazol-1-yl,
          4 - phenylsulfonyl - 1,2,3 - triazin - 1 - yl,
          4 - \text{ethylsulfonyl} - 5 - \text{isopropyl} - 1,2,3 - \text{triazin} - 1 - \text{yl},
          4 - isopropyl - 5 - ethylsulfonyl - 1,2,3 - triazin - 1 - yl,
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          4,5 - di(aminocarbonyl) - 1,2,3 - triazin - 1 - yl,
          4,5 - dicarboxy - 1,2,3 - triazin - 1 - yl,
          4 - trimethylsilyl - 5 - dimethylaminosulfonyl - 1,2,3 - triazol - 1 - yl,
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4 - (1.1 - dimethyl)ethyl - 5 - dimethylaminosulfonyl - 1,2,3 - triazol - 1 - yl,
           4 - dimethylaminosulfonyl - 5 - (1,1 - dimethyl)ethyl - 1,2,3 - triazol - 1 - yl,
           4 - trimethylsilyl - 1,2,3 - triazol - 1 - yl,
           4 - trimethylsilyl - 5 - methoxycarbonyl - 1,2,3 - triazol - 1 - yl,
           4 - phenylsulfonyl - 5 - trimethylsilyl - 1,2,3 - triazol - 1 - yl,
5
           2 - benzoyl - 4,5 - dibromoimidazol - 1 - yl,
           2 - thioxo - 2.3 - dihydro - 1.3.4 - oxadiazol - 3 - yl
           2 - \text{thioxo} - 2.3 - \text{dihydro} - 5 - (2 - \text{pyridyl}) - 1.3.4 - \text{thiadiazol} - 3 - \text{yl}
           2.3 - dihydro - 2 - oxo - 5 - phenyl - 1.3.4 - thiadiazol - 3 - yl,
           2,3 - dihydro - 2 - oxo - 5 - phenyl - 1,3,4 - oxadiazol - 3 - yl,
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           4,5 - dicyano - 1,2,3 - triazol - 1 - yl,
           4,5 - dicyano - 1,2,3 - triazol - 2 - yl,
           4.5 - di(t - buty|sulfony|) - 1.2.3 - triazol - 1 - yl,
           4,5 - di(1 - piperidinylcarbonyl) - 1,2,3 - triazol - 1 - yl,
           4.5 - di(trifluoromethyl) - 1.2.3 - triazol - 1 - yl,
15
           4,5 - di(1 - piperidinylcarbonyl) - 1,2,3 - triazol - 2 - yl,
           2 - methylthio - 5 - methyl - 6 - oxo - 1, 2 - dihydro - 1, 3, 4 - triazin - 1 - yl,
           3 - \text{benzyloxy} - 4.5 - \text{dihydro} - 5 - \text{oxo} - 1.2.4 - \text{oxadiazol} - 4 - \text{yl}
           1.2.5 - thiadiazol - 3 - vl.
           2 - (4 - methoxybenzoyl)indol - 1 - yl or
20
           2 - \text{methyl} - 3 - (2.6 - \text{dichlorobenzoy}) \text{ indol} - 1 - \text{yl}.
```

- 10. A pharmaceutical composition comprising a proteolytic enzyme inhibiting concentration of a compound of Formula 1 as claimed in claim 1, and a pharmaceutical carrier.
- 11. A pharmaceutical composition comprising a proteolytic enzyme inhibiting concentration of a compound as claimed in any one of claims 2 to 9 and a pharmaceutical carrier.
- 12. The use of acompound of Formula I as claimed in claim 1, for preparation of a medicament for treatment of a degenerative disease in a patient in need of such treatment.
 - 13. The use of a compound as claimed in any of claims 2 to 9 for preparation of a medicament for treatment of a degenerative disease in a patient in need of such treatment.

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PARTIAL EUROPEAN SEARCH REPORT

Application Number

which under Rule 45 of the European Patent Convention shall be considered, for the purposes of subsequent proceedings, as the European search report

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Category	Citation of document with in of relevant pas		Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 5)
D,X	WO-A-9 013 549 (STER * claims *	RLING DRUG INC.)	1-13	C07D275/06 A61K31/425
\	EP-A-O 236 930 (MERF PHARMACEUTICALS INC. * claims *	EELL DOW	1-13	C07D417/06 C07D417/12 C07D417/14 A61K31/435
A	EP-A-O 220 651 (AMER CORPORATION) * claims *	ICAN HOME PRODUCTS	1-13	
				TECHNICAL FIELDS SEARCHED (Int. Cl. 5)
				C07D
INCO	MPLETE SEARCH			
The Searc the provis out a mea Claims se Claims se Claims no	the Division considers that the present in Division considers that the present in so of the European Patent Conventioningful search into the state of the art arched completely: arched incompletely: t searched: t the limitation of the search:			
	see sheet C			
	Place of search	Date of completion of the search	1	Prominer
1	THE HAGUE	03 FEBRUARY 1993		HENRY J.C.
X : pari	CATEGORY OF CITED DOCUMEN	e invention lished on, or		
doc: A : tech	ticularly relevant if combined with and ument of the same category anological background pwritten disclosure	ly, corresponding		

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INCOMPLETE SEARCH
Claims searched completely: 2-13
Claim searched incompletely: 1

Claim 1 which rely to an other patent publication is not clear. (See Article 84 and Rule 29(6) EPC).